

# PHYS 150: Physics

INSTRUCTOR

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La Tourette Hall

## Goal of the course:

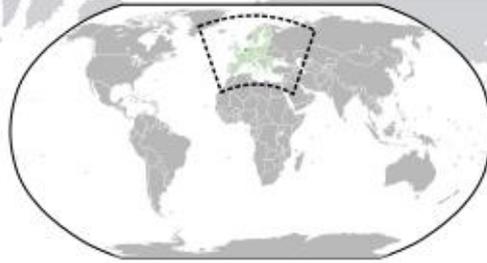
- Teach physics
- Try to make you think a little like a physicist
  - Look at the world in terms of (mathematical) models
  - Critical (scientific) thinking

I am from the Netherlands





Oldest tricolor flag



16.8 million inhabitants

Cf.

France: 65 million

Great-Britain: 62 million

Germany: 82 million

Illinois: 12.9 million

Florida: 19.3 million





First:

English

Nederlands

Country

The Netherlands  
(not Holland)

Nederland

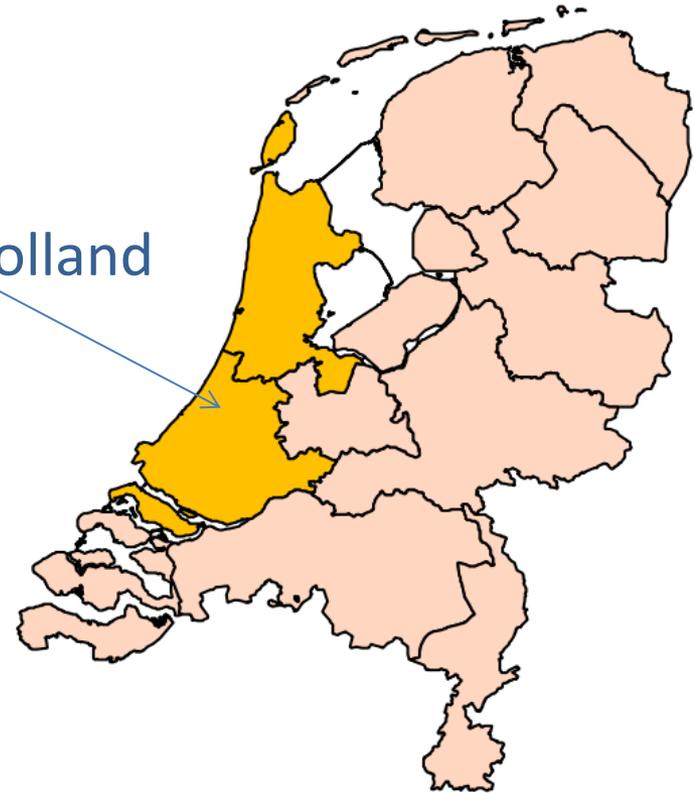
People

Dutch

Nederlanders

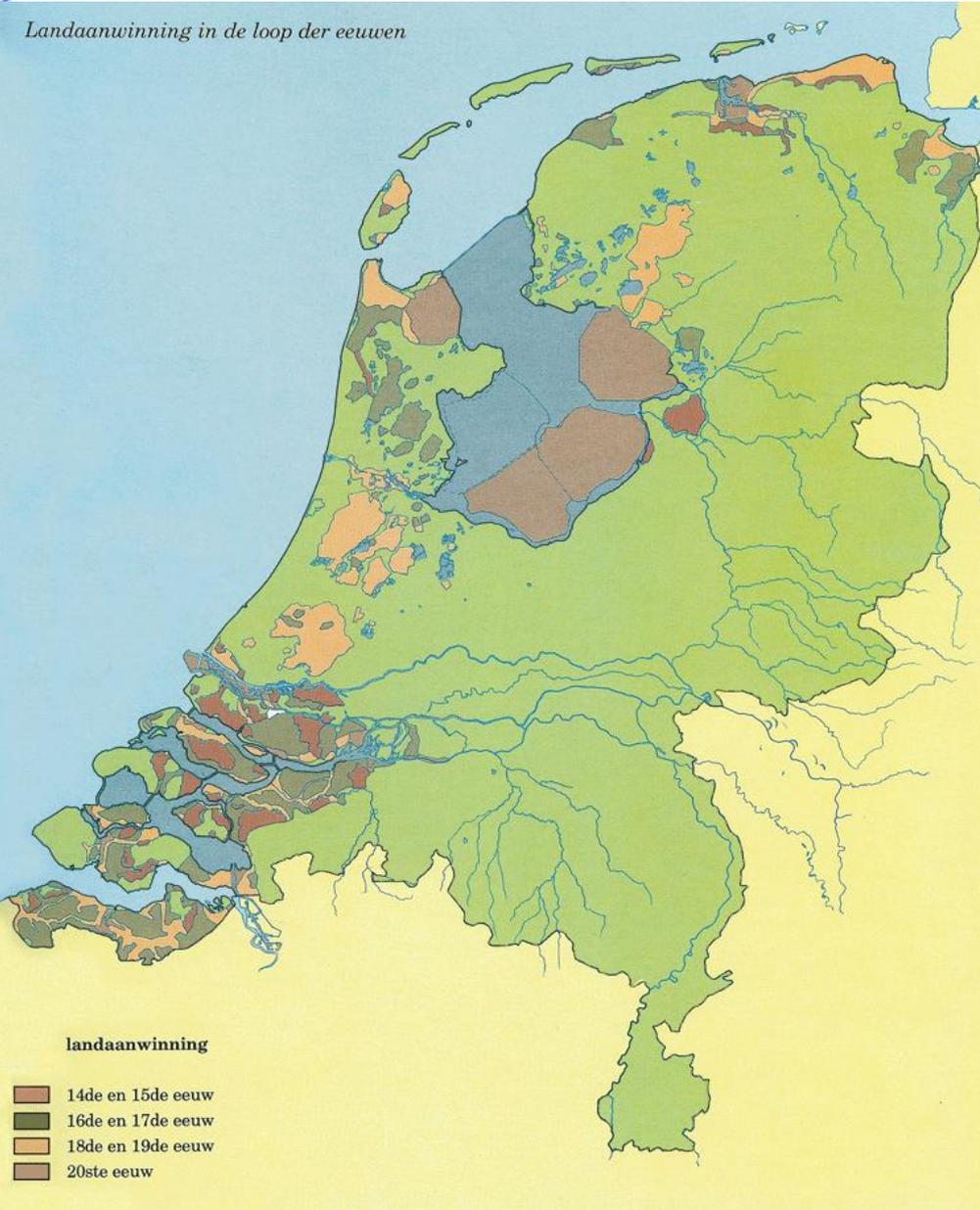


Provinces of Holland



We had nothing to do with those: Dutch=Deutsch=German

# 40% below sea level



Some were a bit more challenging



Haarlemmermeerpolder

1849-1852

185 km<sup>2</sup>

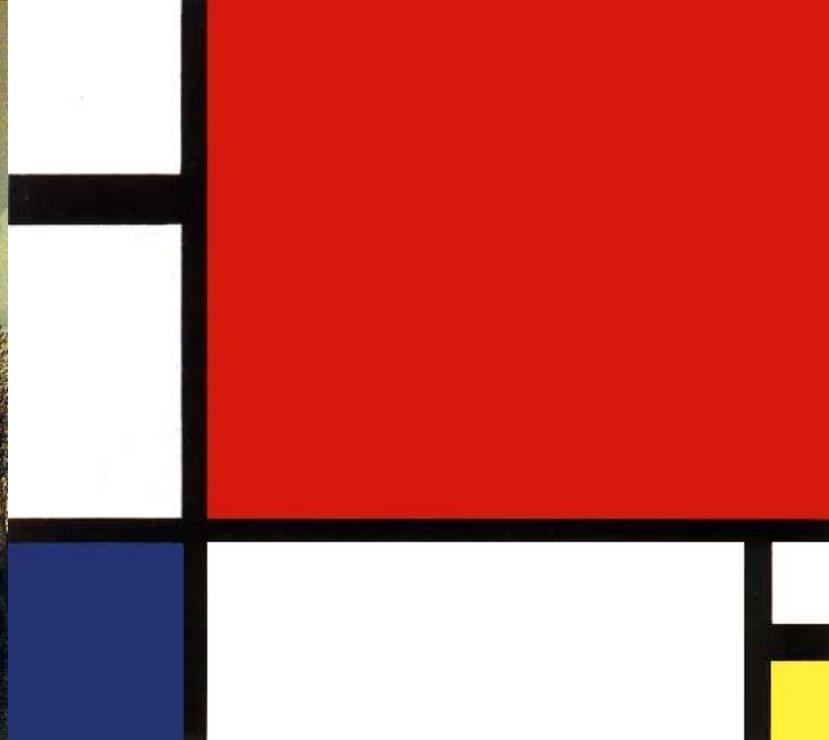
Steam power!

Amsterdam Airport

Also known as Schiphol, which probably stands for Ship's hell



Holland is famous for its  
painters



The country where biking is a mode of transportation



**WHERE'S WALDO?**



# Why Orange?



Royal Family: House of Oranje-Nassau



Principality of Orange



The Pilgrim Fathers came to America in 1620 and founded the city of Plymouth



Which turned into.....



In 1614 (6 years earlier), the Dutch established New Amsterdam

The city of Amsterdam in New Netherlands



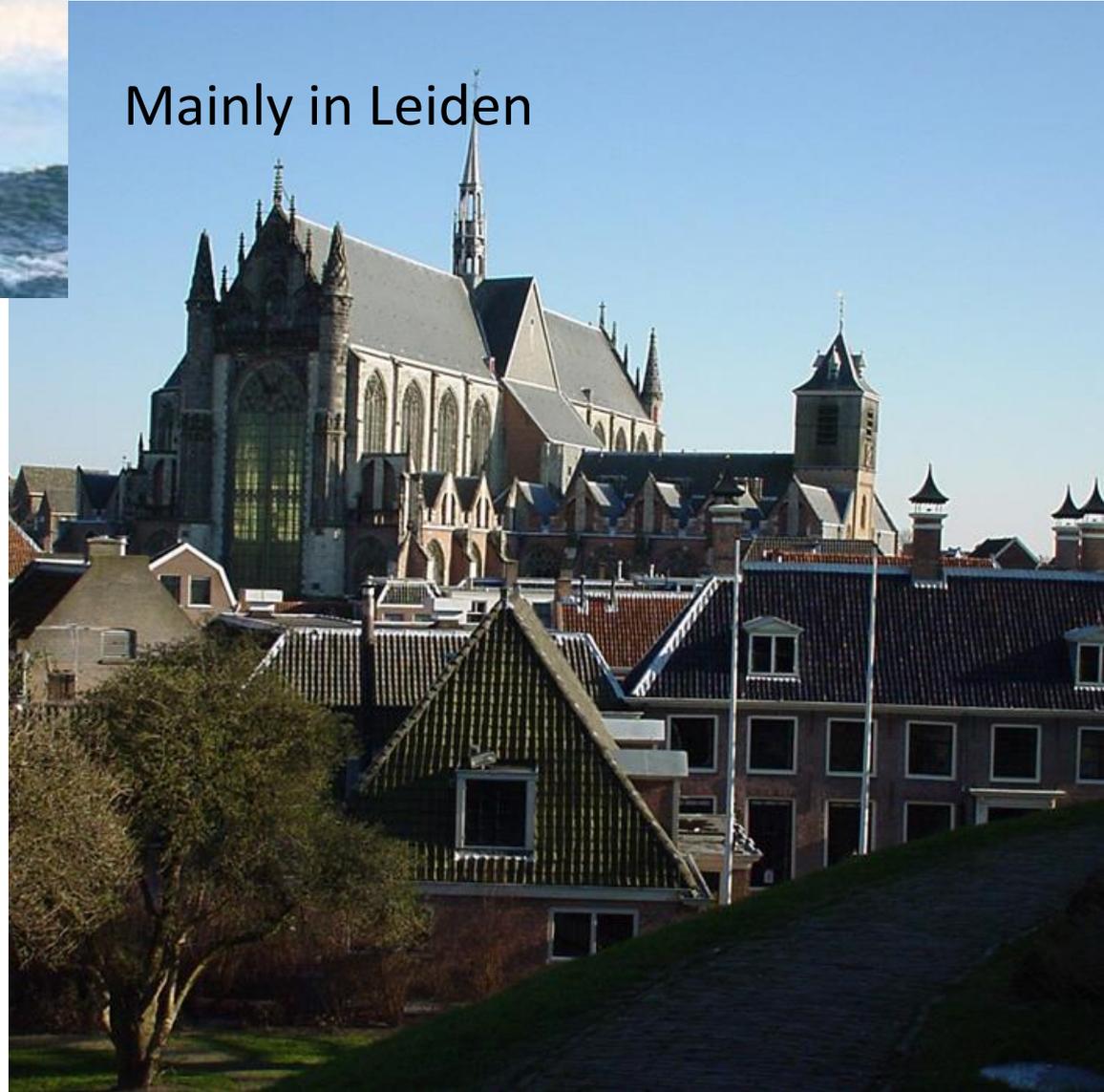
Which turned into:





Before moving to America, they lived in the Netherlands for about 14 years

Mainly in Leiden



They found religious freedom in the Netherlands, but they felt that the Dutch “morals were much too libertine”

“Although they prospered in Holland they were concerned because their children were influenced by the Dutch culture.”

So much for freedom....

Staten Island named after the Dutch Staten Generaal

The Bronx named after Johan Bronck

Harlem=Haarlem

Brooklyn=Breukelen

Flushing=Vlissingen



You know that wall that the Dutch built to keep the native Americans away?



De Waalstraat became **Wall Street**



The end of all this misery

England kept Manhattan



Dutch got the island of Run

Because, hey, we can't live without



nutmeg....



# Maybe you think of Dutch physicists



Lorentz 1903    Zeeman 1903    van der Waals 1910    Kamerlingh-Onnes 1913    Zernike 1953    van der Meer 1984    't Hooft 1999    Veltman 1999    Geim 2010

## 9 Nobel prizes

USA 85 Nobel prizes in physics

Rescale population

UK	$23/63.2 \times 16.75=6.1$
Germany	$25/81.8 \times 16.75=5.1$
USA	$85/315.1 \times 16.75=4.5$
France	$13/65.4 \times 16.75 =3.3$
Italy	$6/60.8 \times 16.75=1.7$
Canada	$3/35.0 \times 16.75=1.4$
China	$2/1353 \times 16.75=0.02$
Belgium	0

1986-1990

Delft University of Technology

1842 founded by King William II  
To train civil servants for the  
Dutch East Indies

Title: Ingenieur



Delft University of Technology

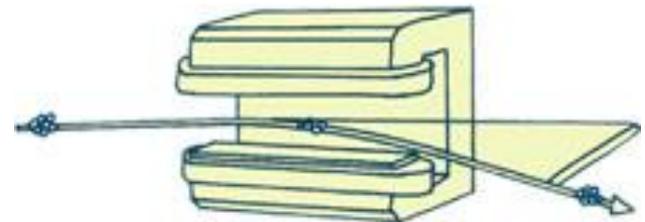
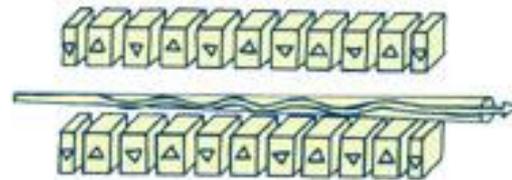
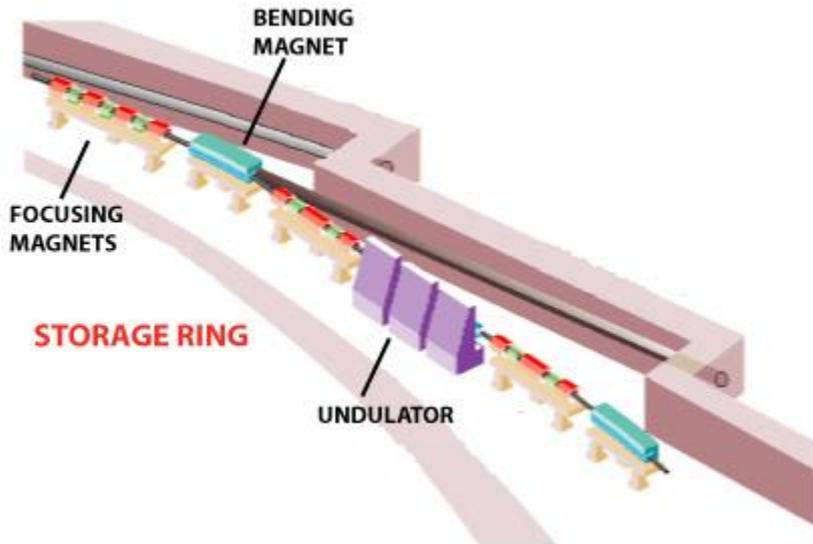
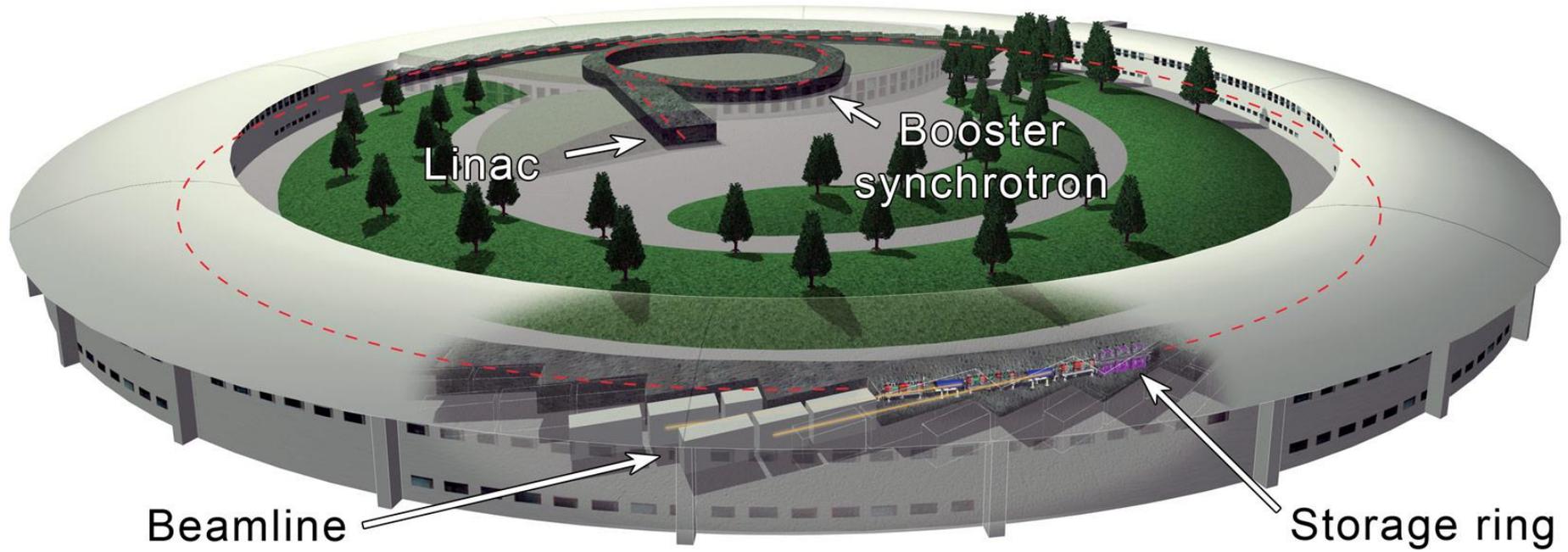




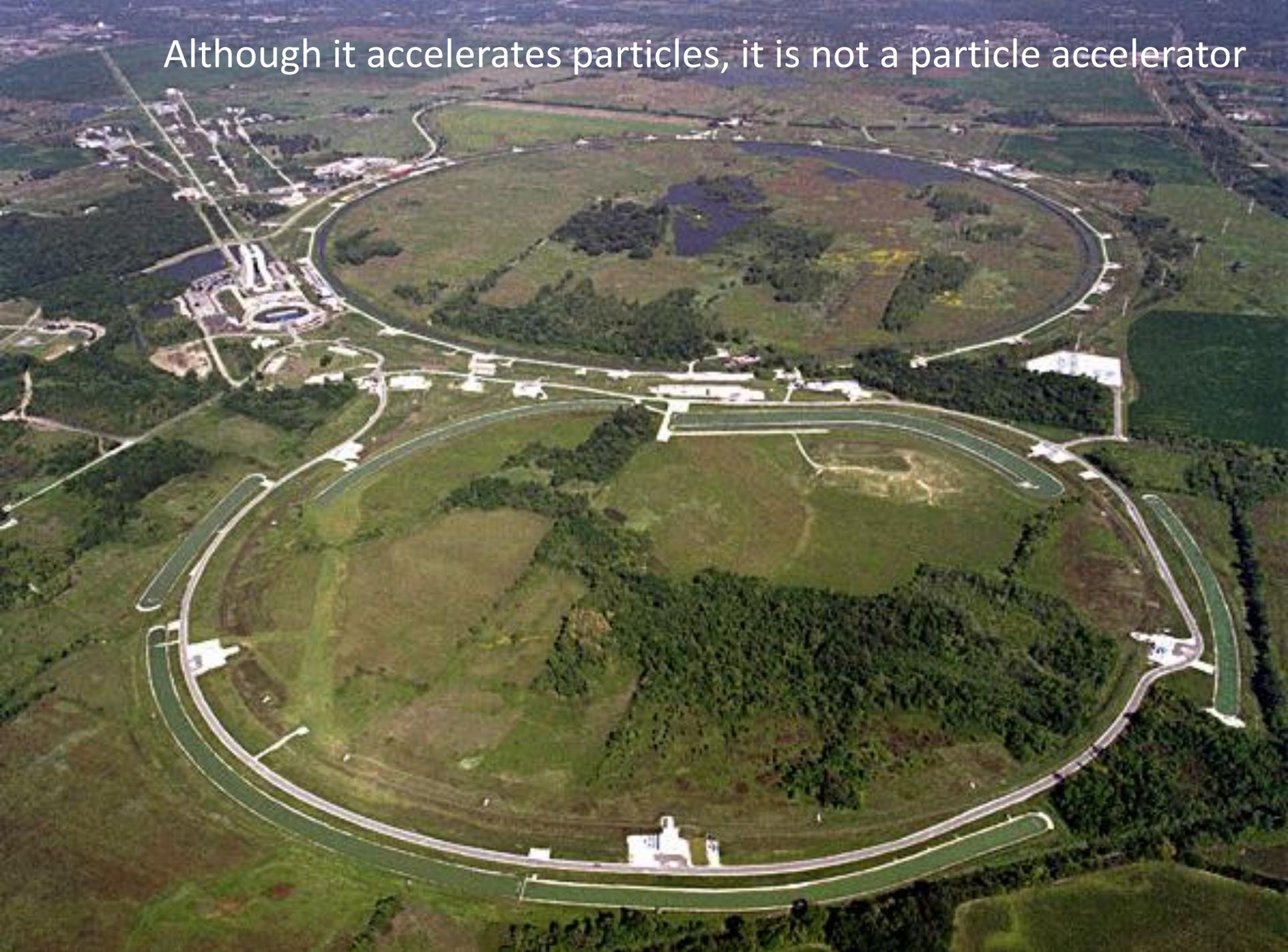
# 1994-1997 European Synchrotron Radiation Facility



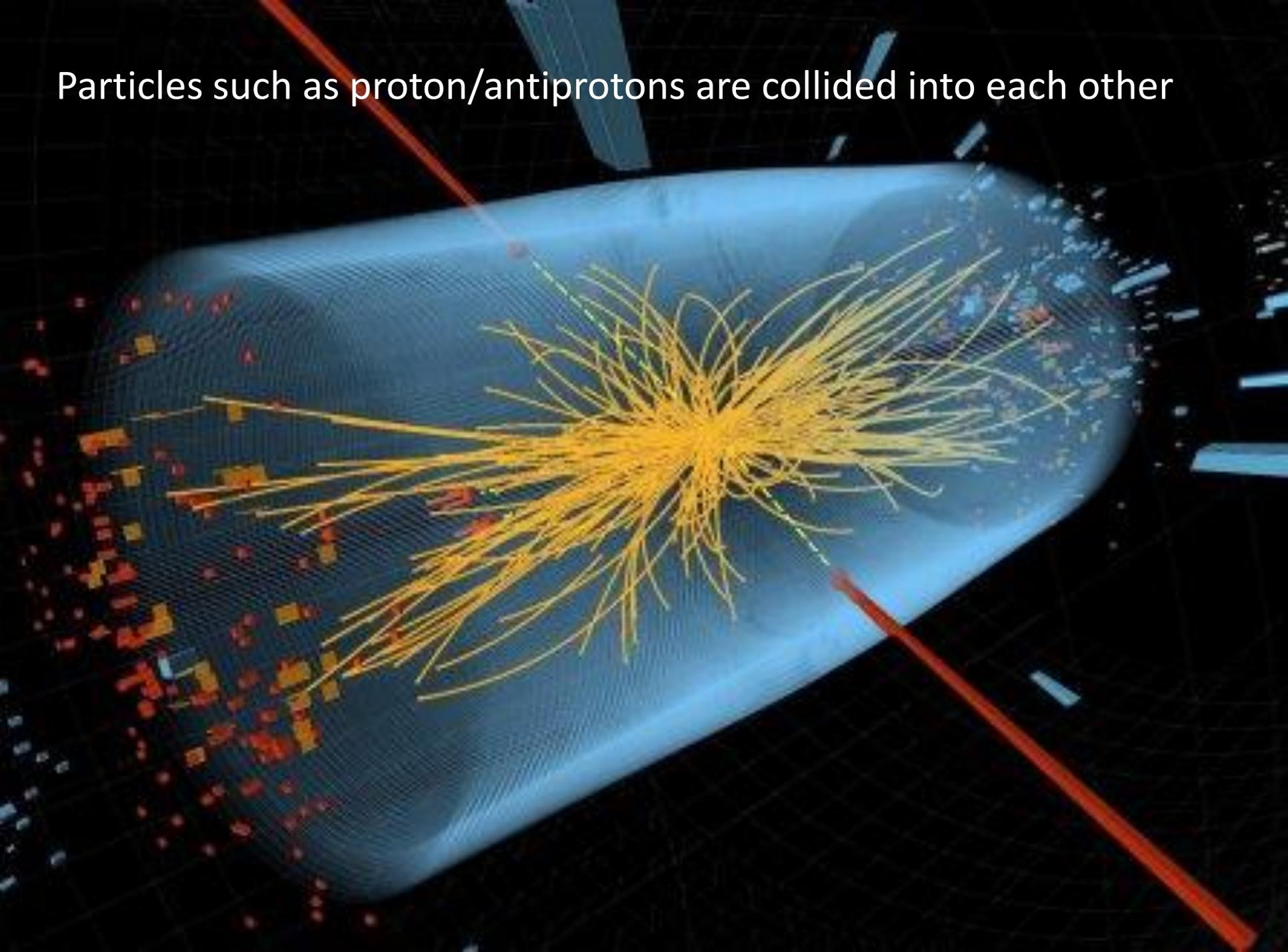
What is a synchrotron? A giant source of X-rays



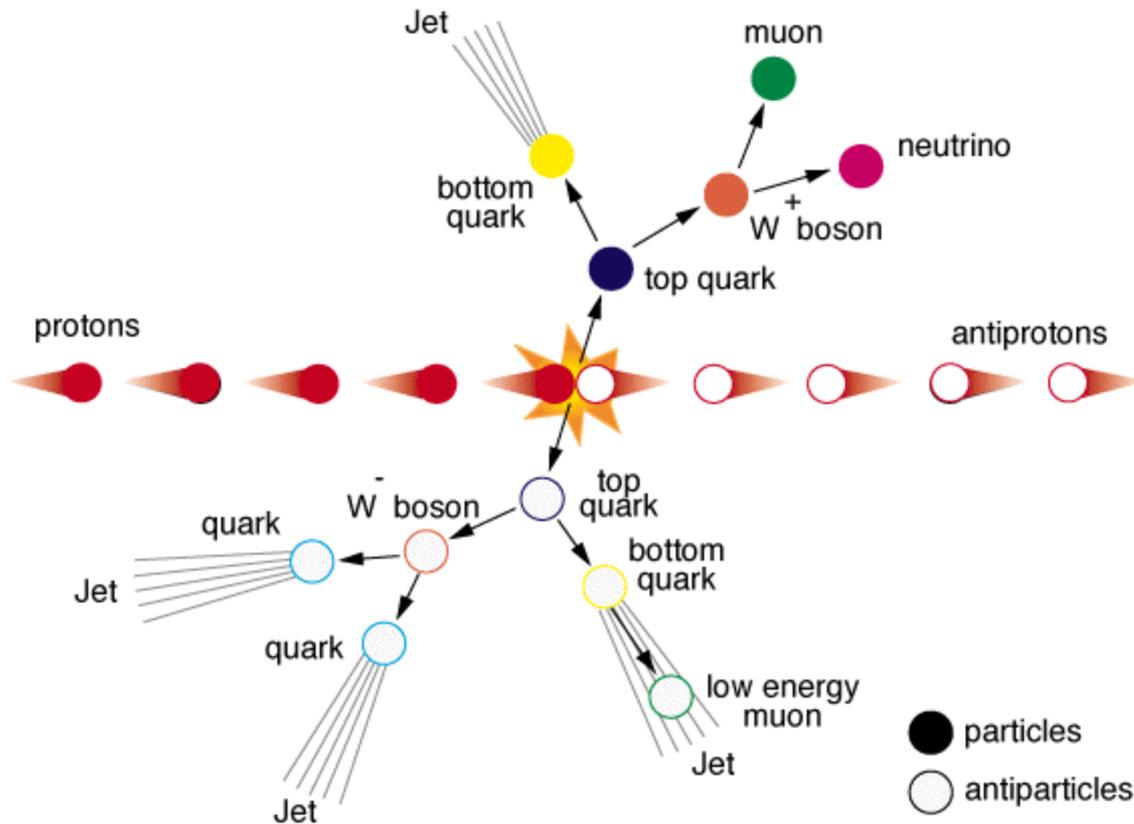
Although it accelerates particles, it is not a particle accelerator



Particles such as proton/antiprotons are collided into each other



In the collision, exotic particles, such as quark, can be produced. One is looking at the decay products of these particles.



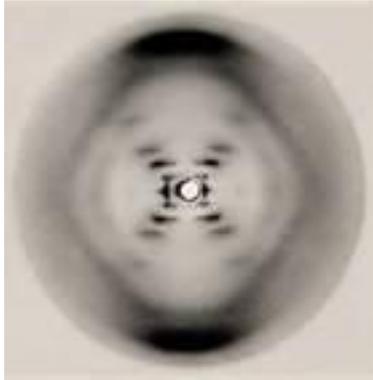
A Top Antitop Quark Event from the D-Zero Detector at Fermilab



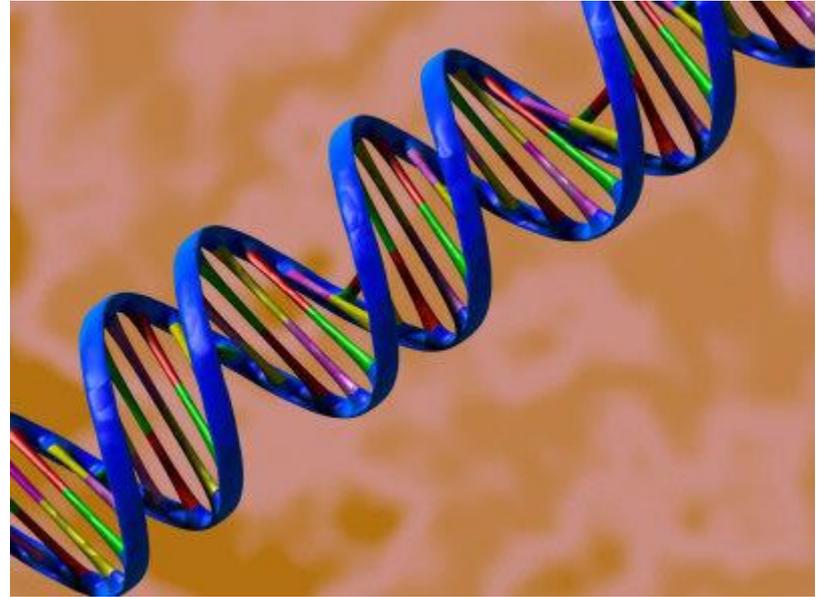
What can you do with X-rays, besides the usual?



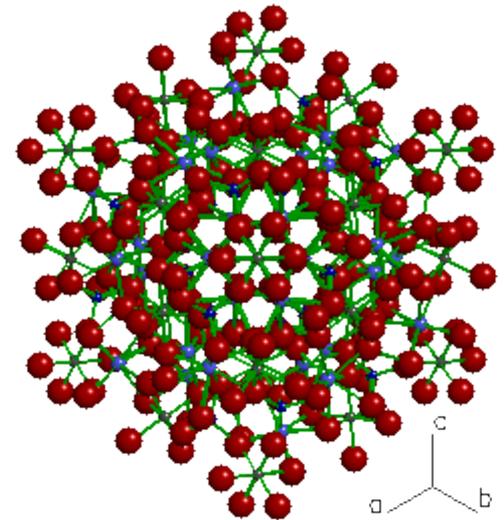
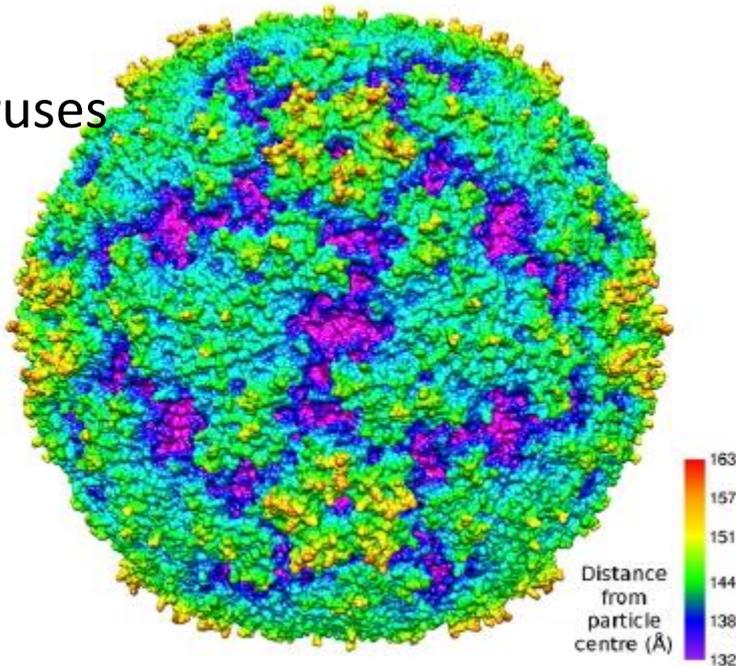
# Determine the structure of crystals and molecules



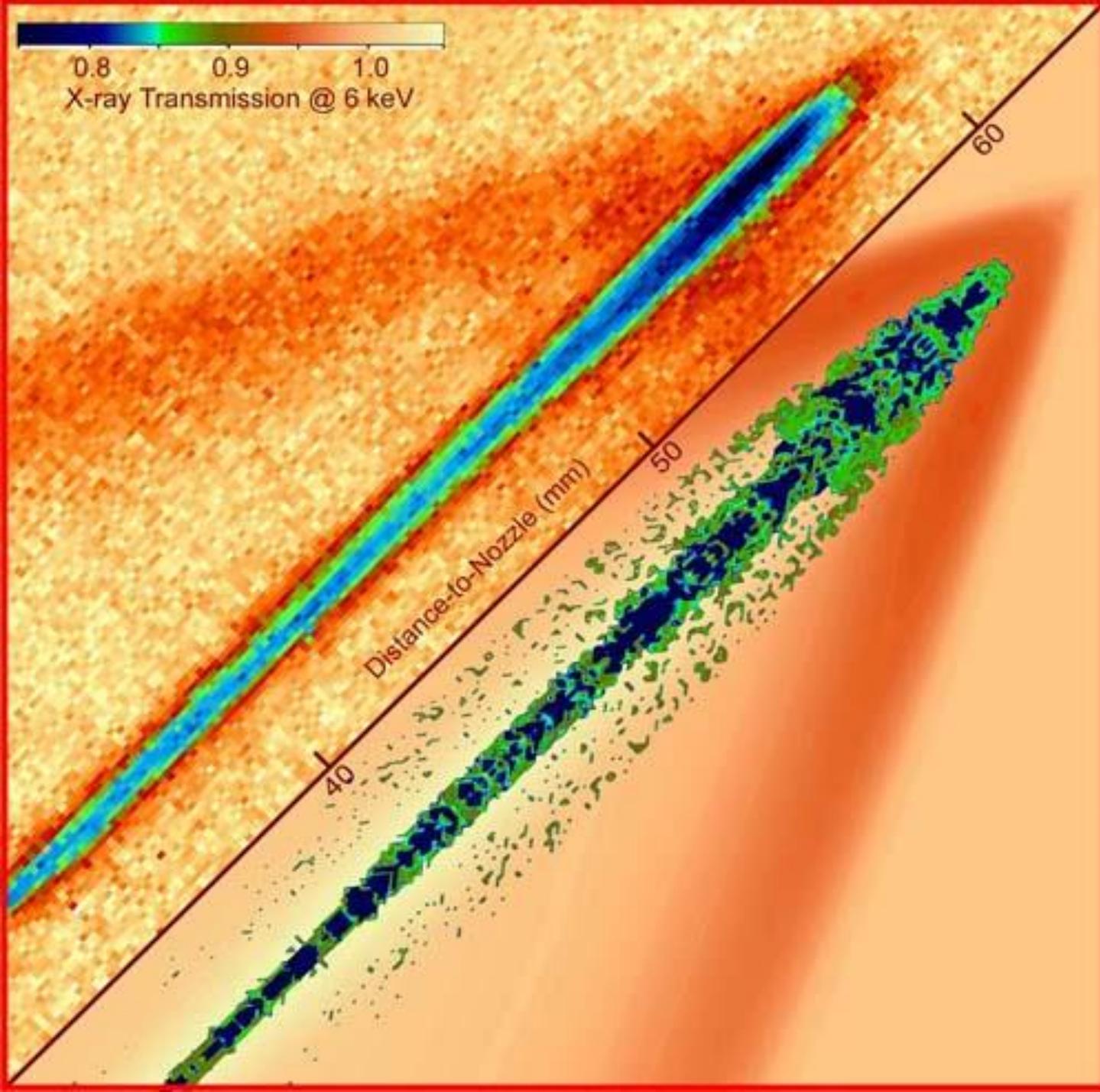
DNA



Or viruses

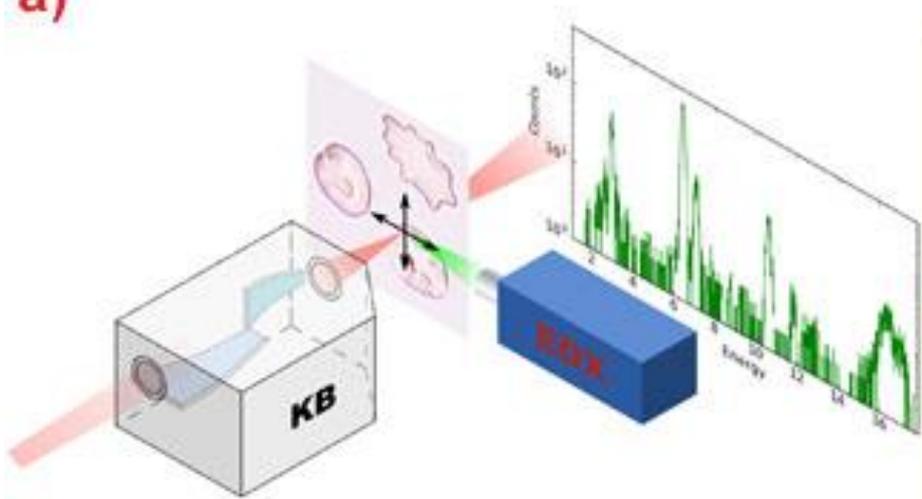


We can look  
inside fuel sprays

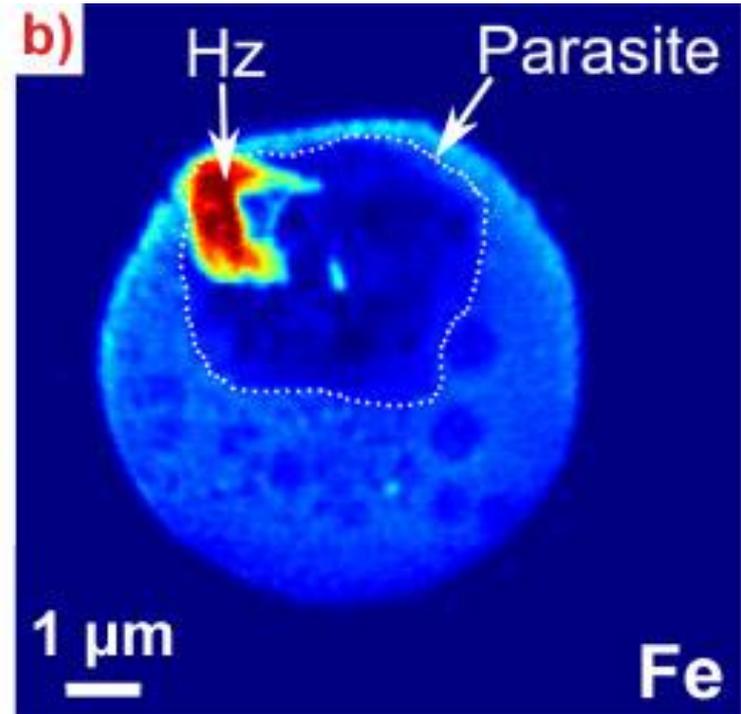


# X-ray microscopy

a)

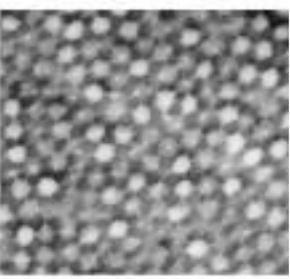
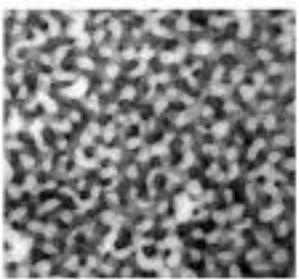
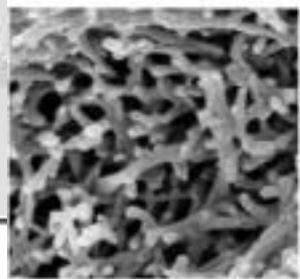
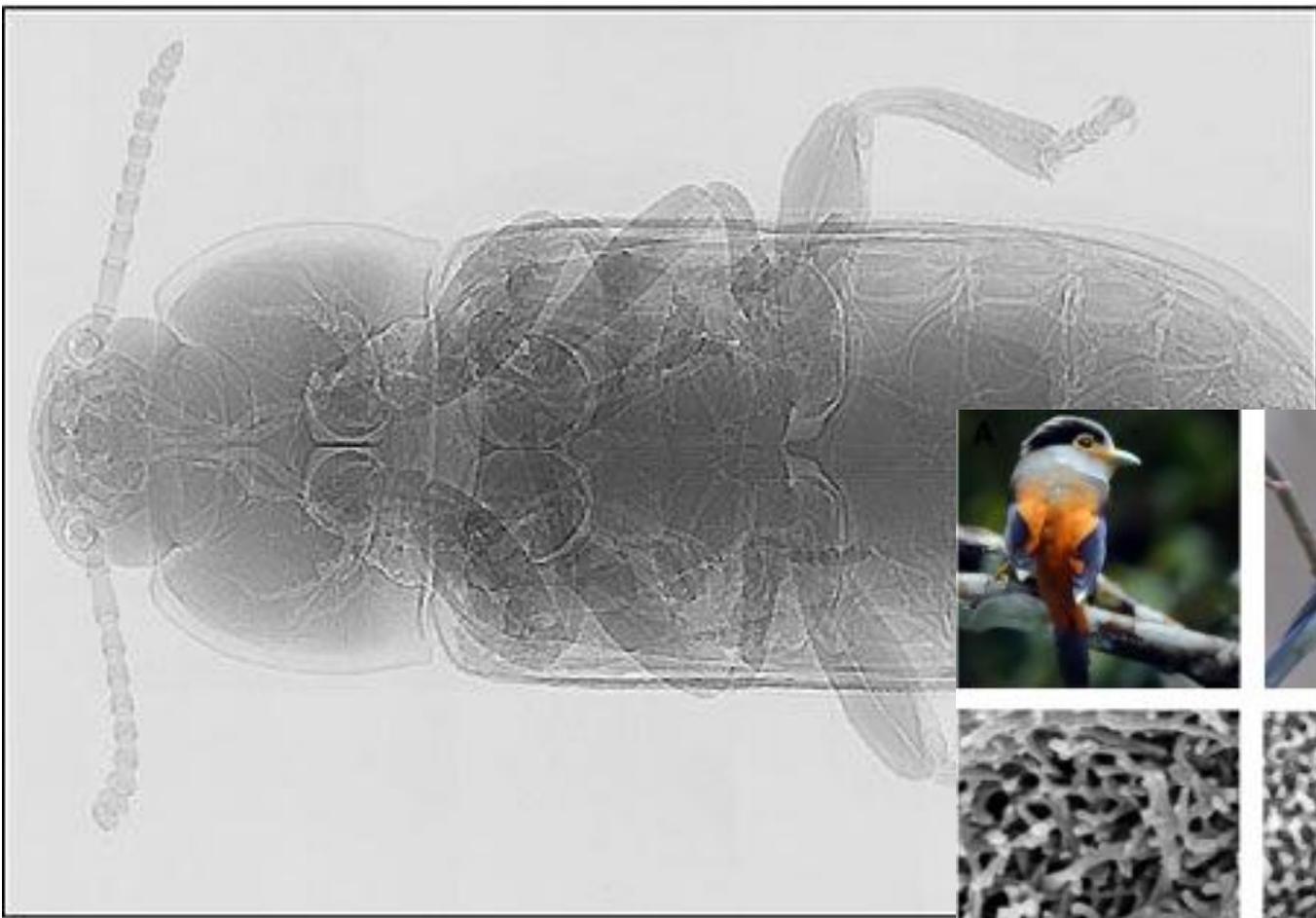


b)

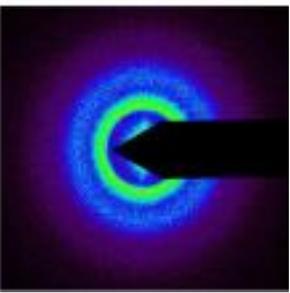
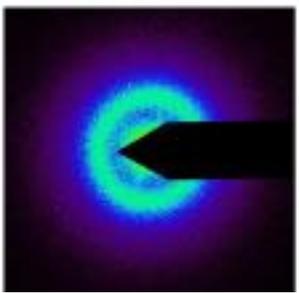
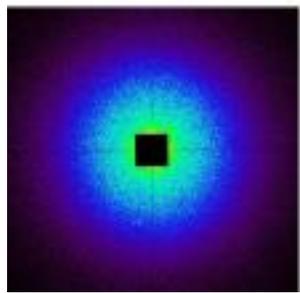


X-ray fluorescence microscopy reveals the location of iron in a red blood cell infected with a malaria parasite.

It can show that insects can breathe!



It provides insight into the different blues of birds



1997-98 postdoctoral position at NIU



**Northern Illinois University**



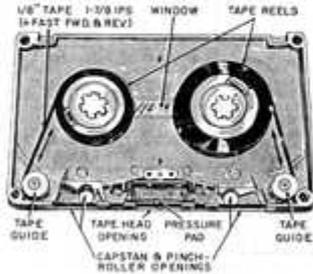
1998-2002 Philips Research  
Senior Research Scientist

**PHILIPS**



# Magnetic Recording History Pictures 4

## Philips Compact Cassette



Philips of 1965,  
from *Electronic World*, Nov. 1966

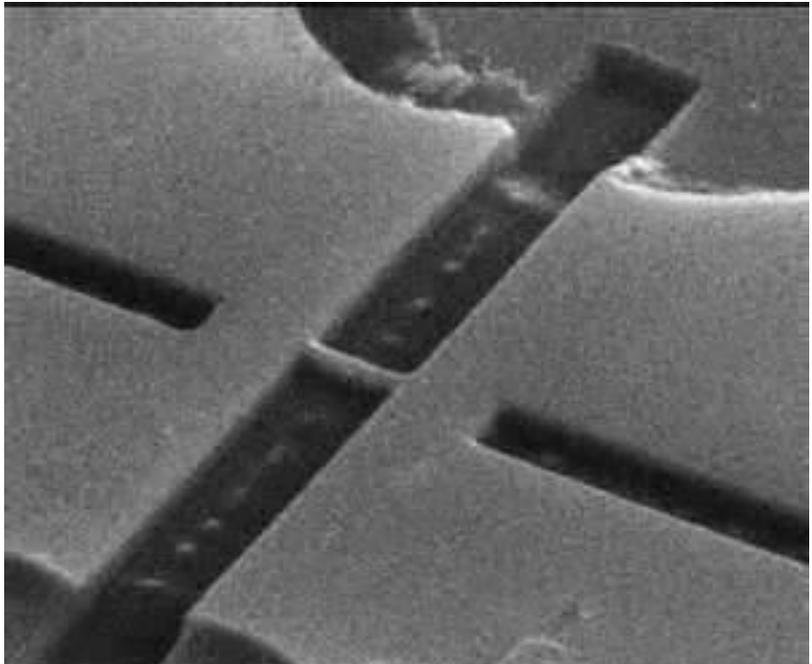


Philips of 1965,  
from *Electronic World*, Nov. 1966

Co-invented the CD player



# Electronic Microscopes





**Northern Illinois University**

2002 Associate Professor

Title now: Prof. Dr. Ir.

2008 Professor

2009 Presidential Research Professor

2013 Distinguished Research Professor  
(Essentially a retired PRP)



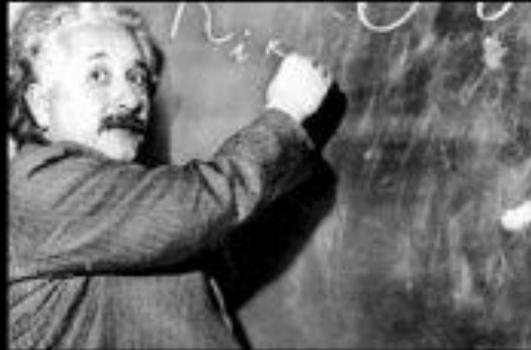
No not you, the NIU president

2005 joint position with Argonne National Laboratory

# SCIENCE TEACHER



What my friends think I do



What my mom thinks I do



What society thinks I do



What my students think I do

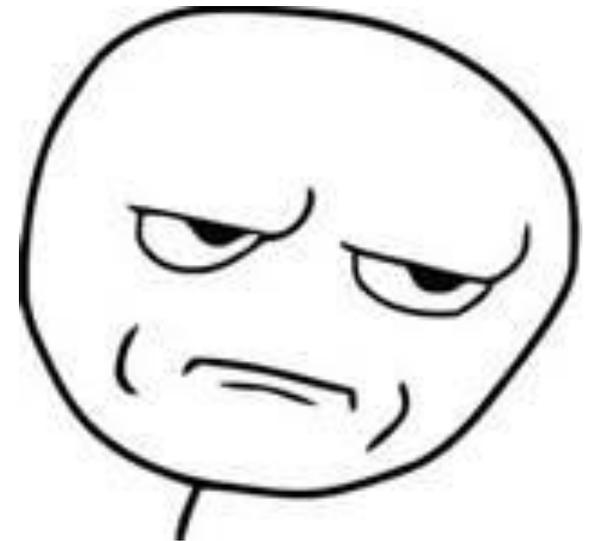


What I think I do



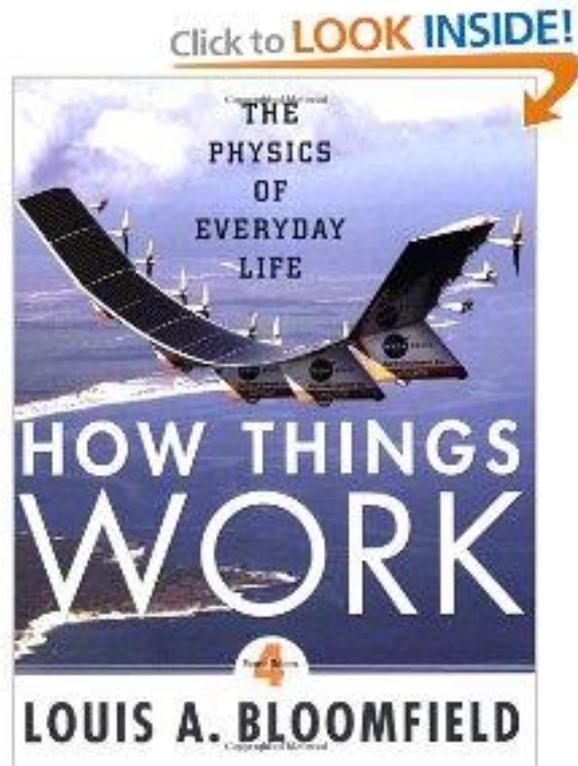
What I do

All right back to Physics 150...



This is what my wife says she feels about physics

# BOOK: MWF Class:



Lyle Marschand

Skating, falling balls, and bumper cars are included to explain the Laws of Motion.

Air conditioners and automobiles are used to explore thermodynamics

Sounds like fun

# In the end, it just reads like any other physics books

Skates almost completely eliminate friction, at least in one direction, so that you can glide effortlessly across the ice or roller rink and experience your own inertia. For simplicity, imagine that your skates are perfect and experience no friction at all as you glide. Also, for this and the next couple of sections, let's forget not only about friction but also about air resistance. Since the air is calm and you're not moving too fast, air resistance isn't all that important to skating anyway.

Now that you're ready to skate, we'll begin to examine five important physical quantities relating to motion and look at their relationships to one another. These quantities are position, velocity, mass, acceleration, and force.

Let's start by describing where you are. At any particular moment, you're located at a **position**—that is, at a specific point in space. Whenever we report your position, it's always as a **distance** and **direction** from some reference point: how many meters north of the refreshment stand or how many kilometers west of Cleveland.

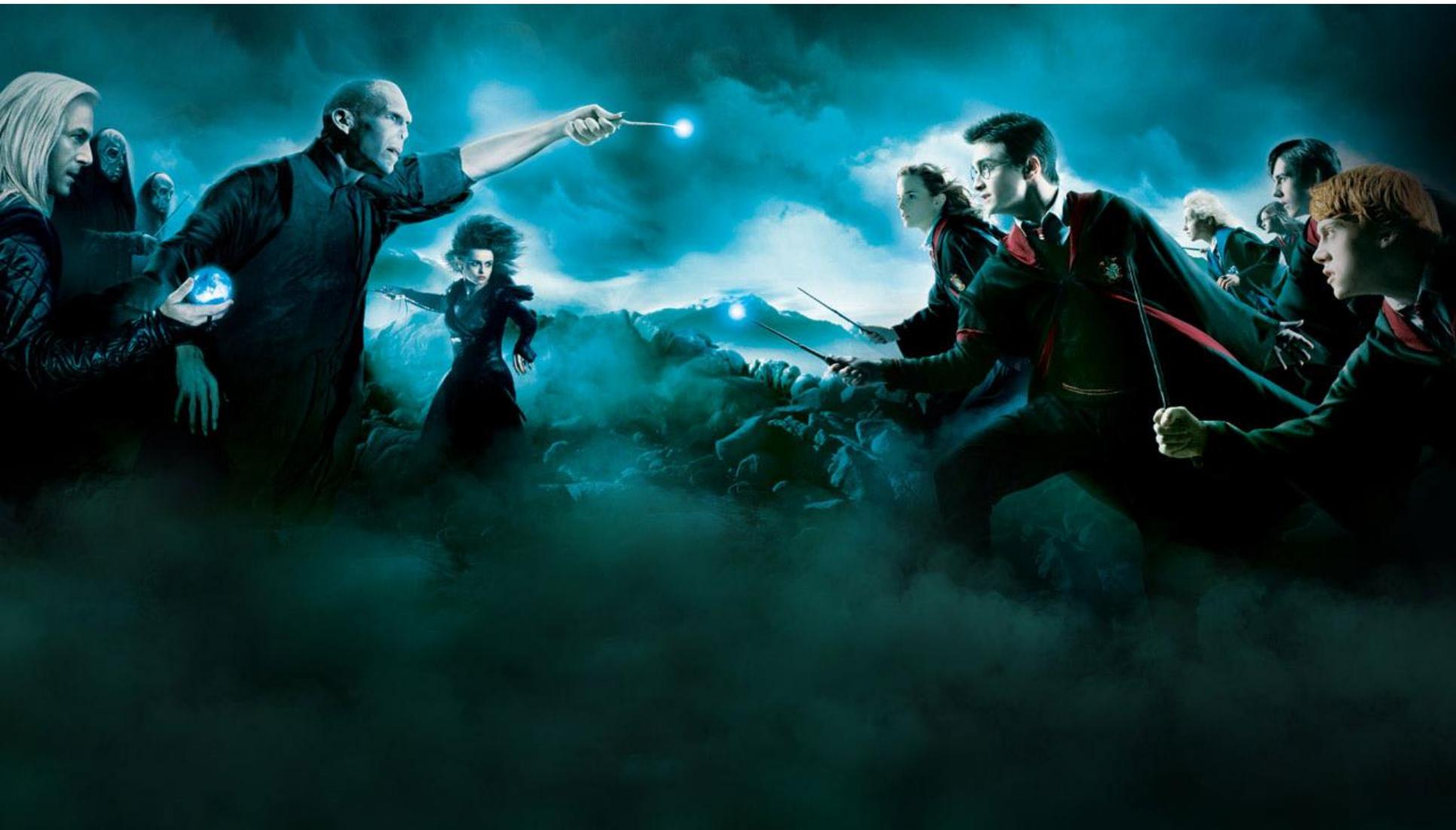
Position is an example of a vector quantity. A **vector quantity** consists of both a magnitude and a direction; the **magnitude** tells you how much of the quantity there is, while the direction tells you which way the quantity is pointing. Vector quantities are common in nature. When you encounter one, pay attention to the direction part; if you're looking for buried treasure 30 paces from the old tree but forget that it's due east of that tree, you'll have a lot of digging ahead of you.

You're on your feet and beginning to skate. If you're moving, then your position is changing. In other words, you have a velocity. **Velocity** measures how quickly your position changes; it's our second vector quantity and consists of the speed at which you're moving and the direction in which you're heading. Your **speed** is the distance you travel in a certain amount of time,

$$\text{speed} = \frac{\text{distance}}{\text{time}},$$

With just as many words as an average Harry Potter

But without:



Ok, let's try to read it:

Now that you're ready to skate, we'll begin to examine five important physical quantities relating to motion and look at their relationships to one another. These quantities are position, velocity, mass, acceleration, and force.

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This course:

No book. Lecture notes to be at [www.niu.edu/~veenendaal/150.htm](http://www.niu.edu/~veenendaal/150.htm)

Pros

- They're free (yeah!)
- They have some nice equations!

Cons

- They're written by me
- They have some nice equations!

Please let me know if you find error/mistakes

Exercise and homework files will be posted along the way



All course information is on blackboard

Let me know if you have problems finding something!

**CLASS MEETINGS:** Lecture Section 1: T/Th 3.30-4.45 pm, Tourette 200

**Instructor:** Michel van Veenendaal, Tourette 223; 815-753-0667  
**Office Hours:** Tuesday/Thursday 1.30-3.30PM, or by appointment.  
**Email:** [veenendaal@niu.edu](mailto:veenendaal@niu.edu)  
**Web page:** See blackboard

**Note: Honors students will be required to do a special project – see me to discuss.**

## **SCHEDULE OF TESTS:**

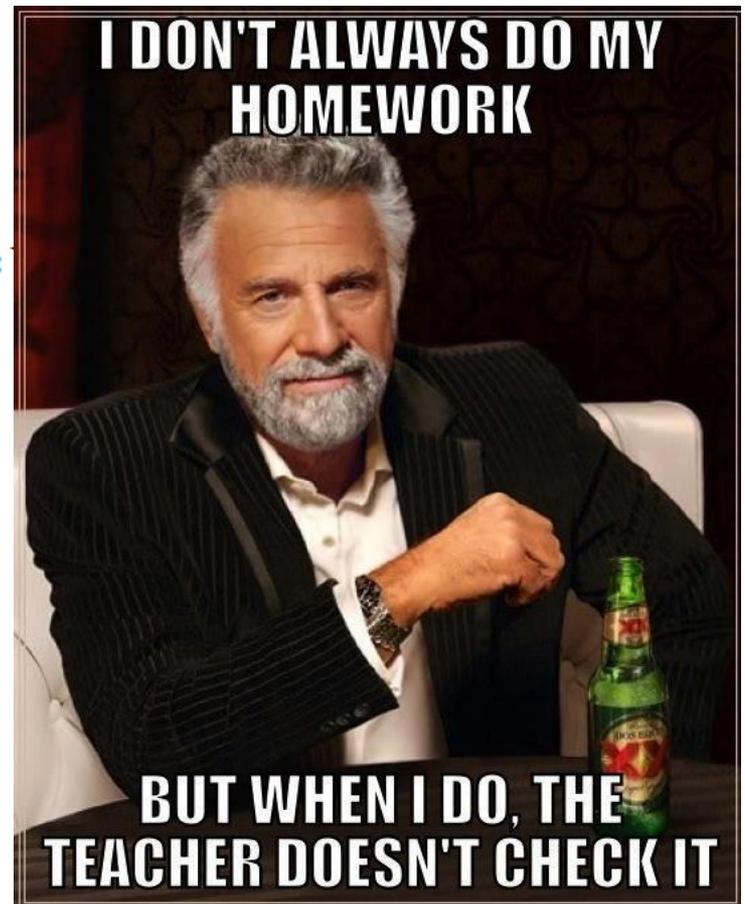
TEST 1	<b>Tuesday 9/24</b>
TEST 2	<b>Thursday 10/17</b>
TEST 3	<b>Tuesday 11/19</b>

**FINAL**      **Tuesday, 12/10, 4-5.50pm** (be aware of potential exam conflicts)

Homework is not graded.

### **MINIMUM REQUIREMENTS TO PASS THE COURSE:**

1. Take 2 of the exams and the Final exam
2. Pass the Lab (150A students only).



**WARNING:**

The final grades are curved.

Marschand's class: Grading system that you are probably used to

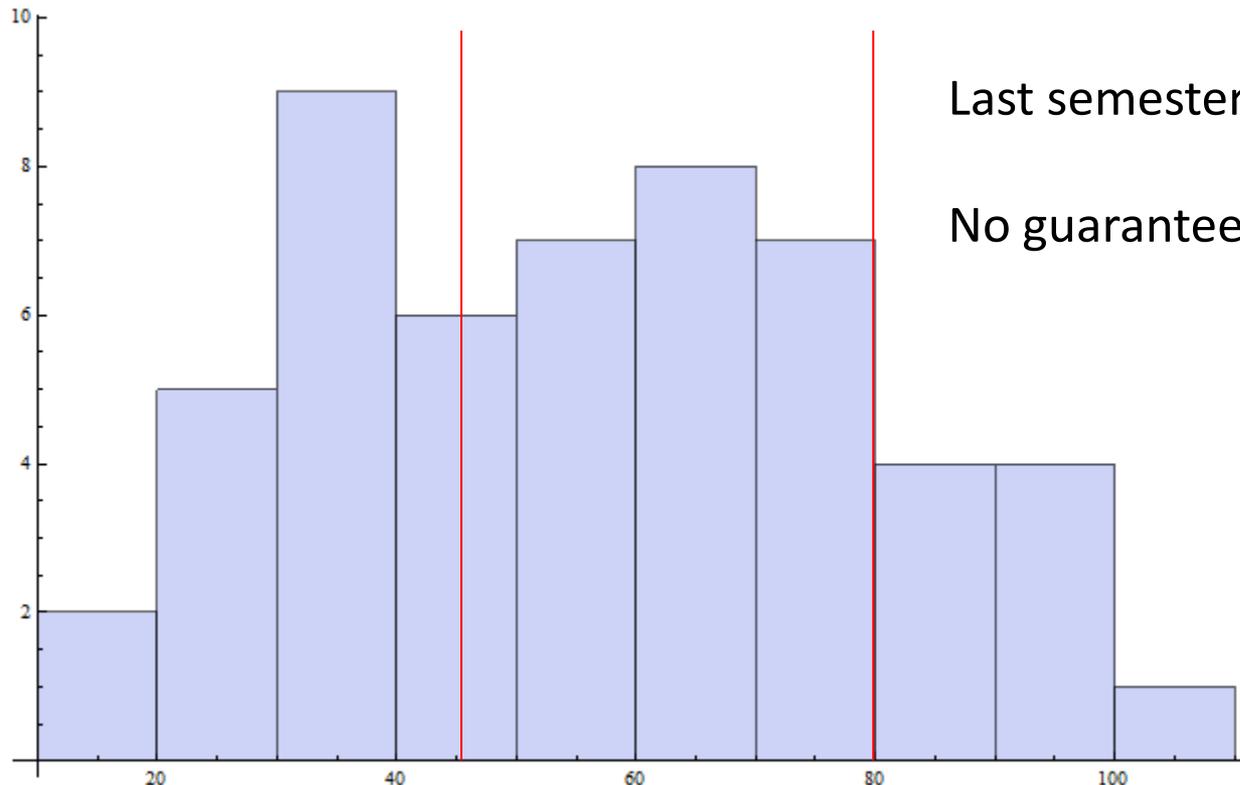
<u>Scale:</u> 100 - 88	A	A – 16%
87 - 75	B	B – 31%
74 - 62	C	C – 31%
61 - 50	D	D – 10%
		F – 13%

# More senior undergraduate/graduate-type grading

The final grades are curved. Therefore, the final grade depends on the distribution of the grades. Typical relationships between the weighted tests/final and the grades are

A	$> 80 \pm 5$
B	$65 \pm 5 - 80 \pm 5$
C	$45 \pm 5 - 65 \pm 5$
D	TBD

Note that these ranges are indicative and that the instructor remains the right to change the values depending on the performance of the students and the difficulties of the exams.



## Marschand

- Exams are “easier”
- Grading is very strict
- Percentage A, B, C, etc.

A – 16%

B – 31%

C – 31%

D – 10%

F – 13%

- Make you feel more comfortable, since you can answer more questions correctly

## MvV

- Exams are more difficult
- Grading is easier
- About the same  
(last semester I actually gave more A's and B's)  
(I'm rather stuck to these historical statistics. We cannot favor one 150 class over another)
- Leaves you with a more uncomfortable feeling

# What's the philosophy? Compare baseball

- I can throw “easy” balls at you and expect you to hit the ball 9 out of 10 times for an A
- You are well-trained for those balls



<u>Scale:</u> 100 - 88	A
87 - 75	B
74 - 62	C
61 - 50	D

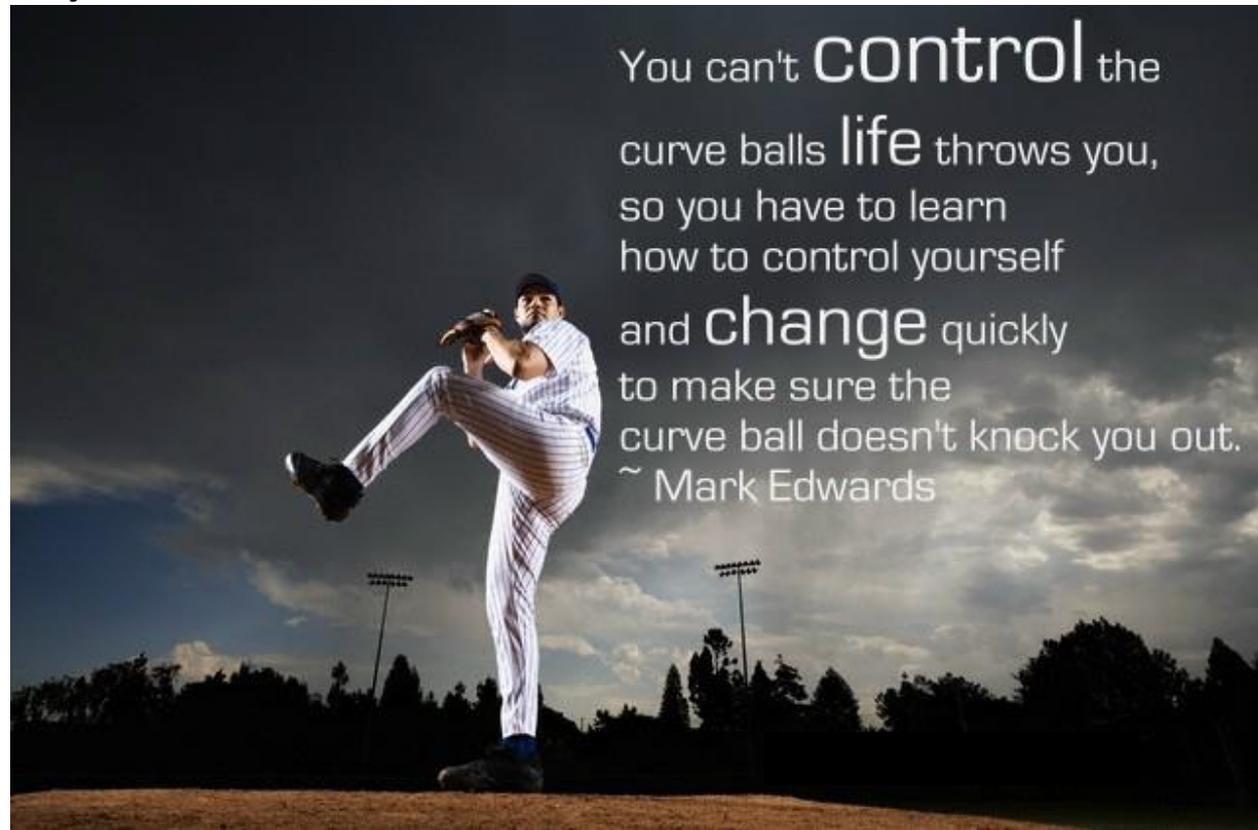
- I can throw a couple of curveballs and fast balls as well and see how you perform on those.
- You are not always well-trained for all questions, but it shows me how you think on your feet.
- Since I don't expect you to always do very well, the grading is more flexible.



Typical complaint: “The questions at the exam were different from those in class”

That’s exactly the point...

Life is not always “easy” balls



“If we had practiced harder on these questions, I would have had a better grade”

Not in this philosophy. The more practice you get on hard questions, the harder I have to make the exams.

Graduate physics students rarely score  $>90\%$ . This does not mean they do not get A's

Having people hit homeruns all the time is boring...

In essence:

Scoring that most of you are used to

A: Hits 90% of the “easy” balls

B: Hits 75% of the “easy” balls

C: Hits less

A: Hits the “easy” ones, and some/all of the curveballs and fastballs

B: Hits most of the “easy” balls, but has problems with the curveballs and fastballs

C: Has problems with the “easy” pitches

“If we had more testing opportunities, I would have had a better grade”

For ~80%, the final grade can be determined from a single test.

Since the final grades are curved, the points are less relevant. It's the performance relative to the other students that determine the grade.

- You can get a 10 point bonus and still have exactly the same grades.

# What is physics?

## Wikipedia

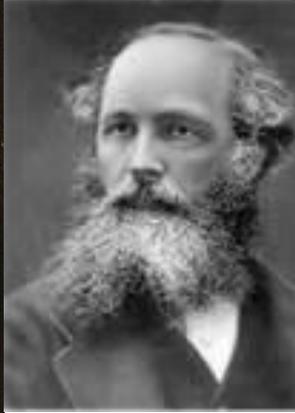
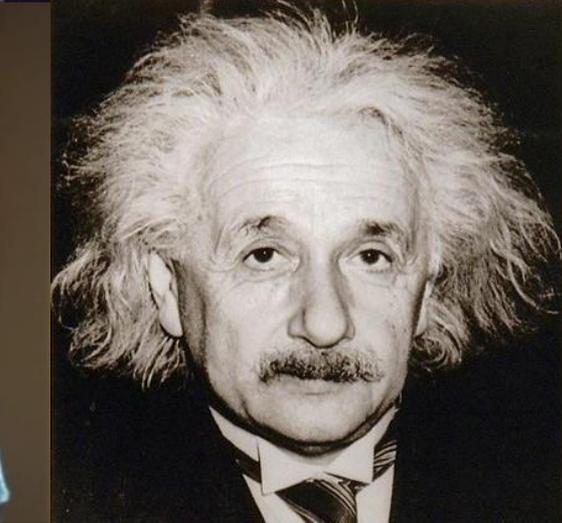
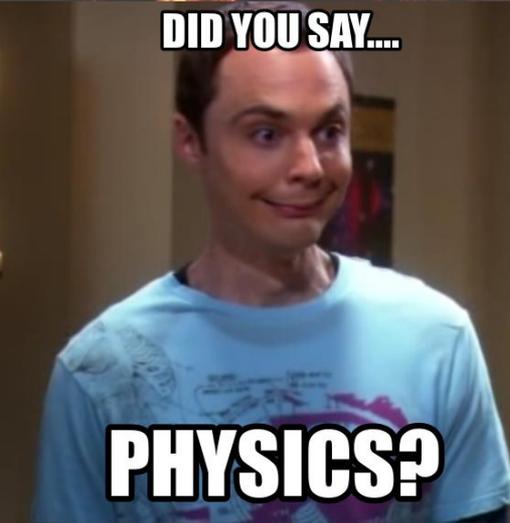
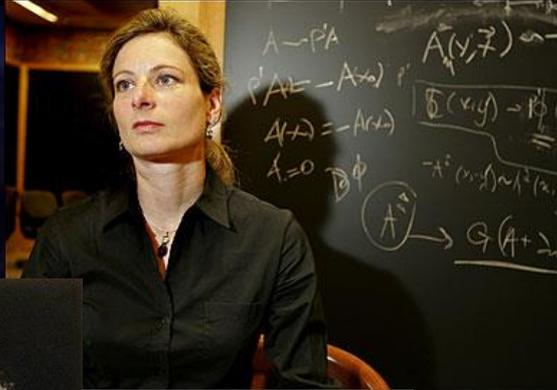
**Physics** (from [Ancient Greek](#): φύσις *physis* "nature") is a [natural science](#) that involves the study of [matter](#)<sup>[1]</sup> and its [motion](#) through space and time, along with related concepts such as [energy](#) and [force](#). More broadly, it is the general analysis of [nature](#), conducted in order to understand how the [universe](#) behaves.

**Chemistry**, a branch of [physical science](#), is the study of the composition, properties and behavior of [matter](#). Chemistry is concerned with atoms and their interactions with other atoms, and particularly with the properties of [chemical bonds](#). Chemistry is also concerned with the interactions between atoms (or groups of atoms) and various forms of energy (e.g. photochemical reactions, changes in phases of matter, separation of mixtures, properties of polymers, etc.).

Chemistry is sometimes called "[the central science](#)" because it connects physics with other [natural sciences](#) such as [geology](#) and [biology](#).

Chemistry is a branch of [physical science](#) but [distinct from physics](#).<sup>[5]</sup>

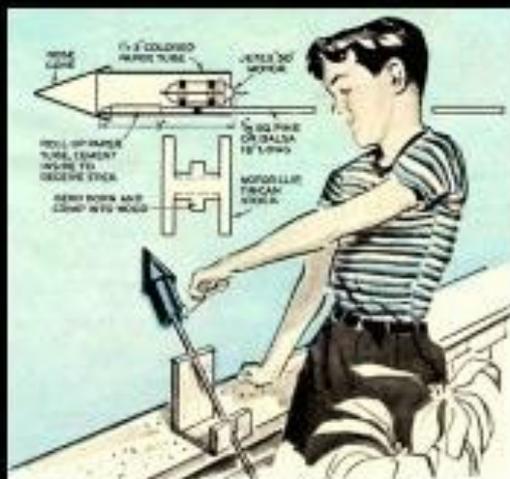
# What is a physicist and how does a physicist look at the world?



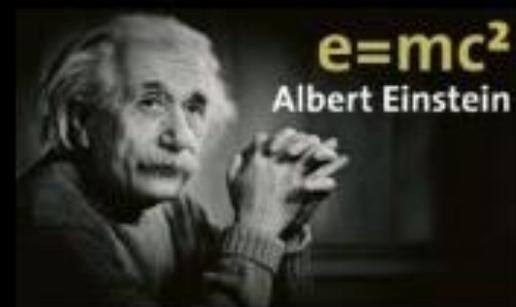
# Physicists



What my friends think I do.



What my mom thinks I do.



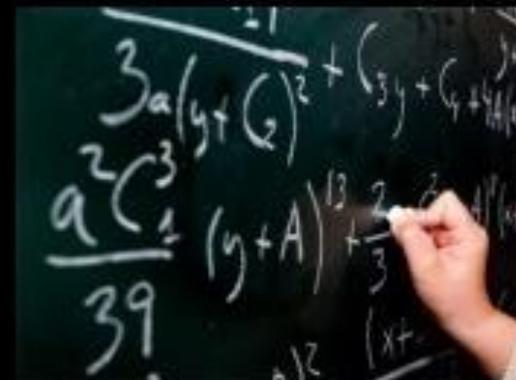
What society thinks I do.



What my boss thinks I do.



What I think I do.



What I actually do.

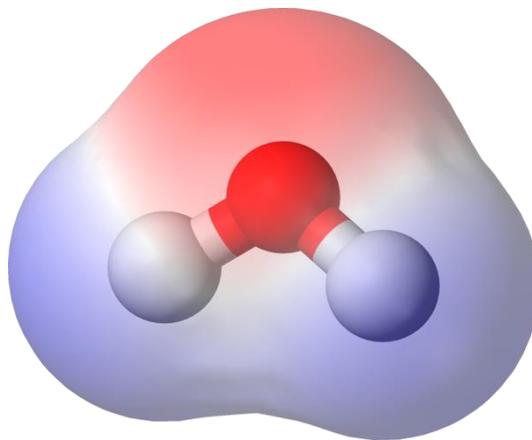
Let us look at some examples:

Most of you recognize this as a chemical reaction



However, why are molecules formed?

$\text{H}_2\text{O}$ :



Linus Pauling  
Quantum Chemist

This is due to chemical bonds

However, Pauling who laid the foundations for the nature of the chemical bond worked with Arnold Sommerfeld, Niels Bohr, Erwin Schrödinger



Who were all physicists and were at the foundation of Quantum Mechanics.

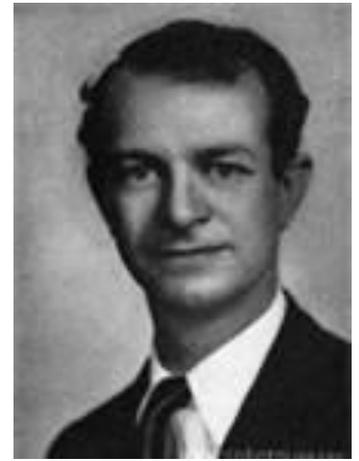
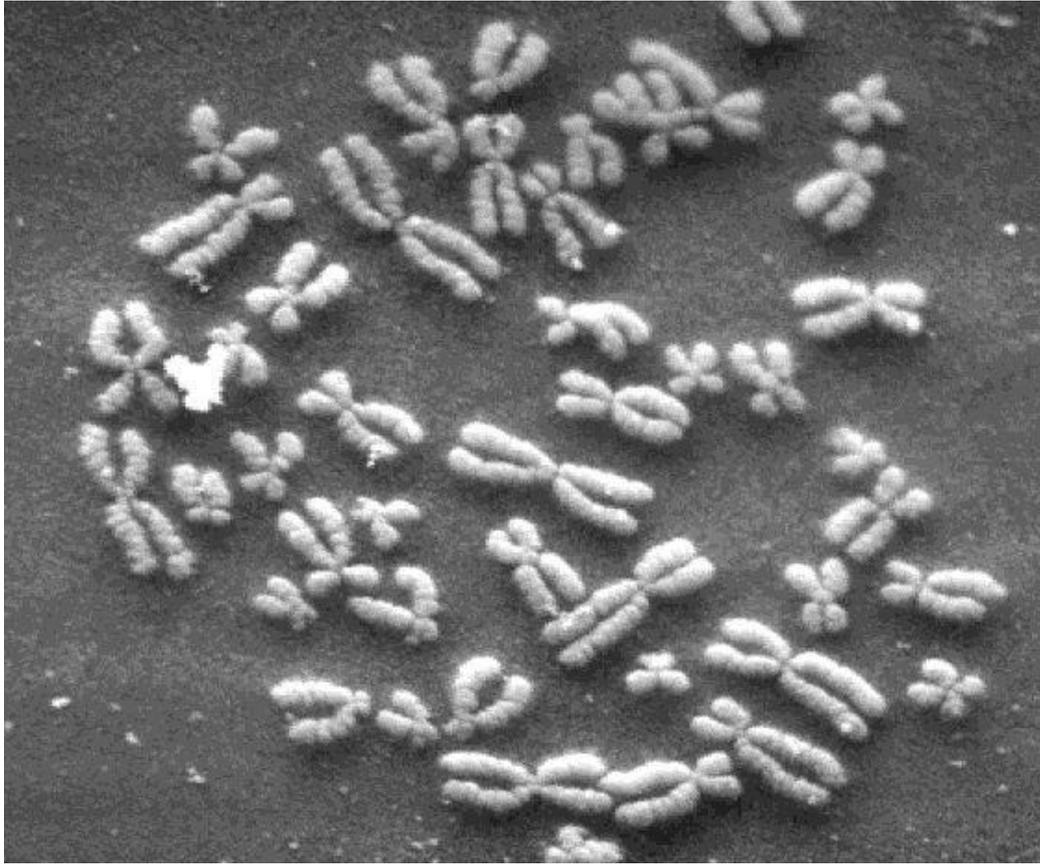


In Zurich, he heard of the valence-bond model of Walter Heitler and Fritz London.

So Quantum Chemistry is Quantum Mechanics applied to molecules

After WW-II, Pauling became interested in the structure of DNA.

Since DNA is at the heart of life, this could be called biology. It is generally known as molecular biology.

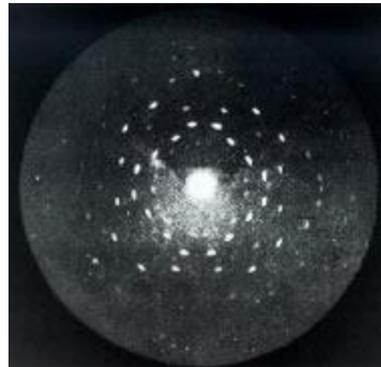


However, somebody else beat him to it.

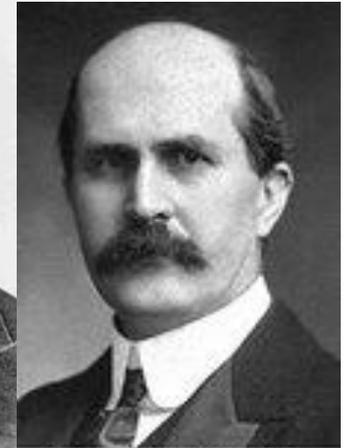
The best way to look at atomic structure is via X-ray diffraction, the basis for X-ray diffraction was made by physicists

1895: discovery of X-rays by Wilhelm Röntgen

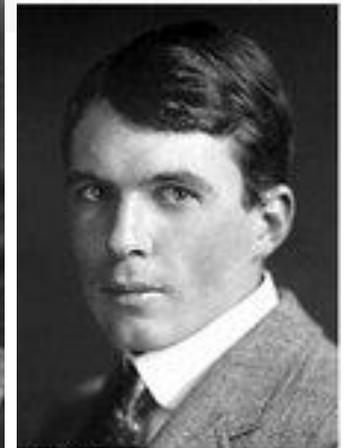
1912: discovery of X-ray diffraction by Max von Laue  
(other contributors Ewald, Sommerfeld)



1913: interpretation by William and Lawrence Bragg

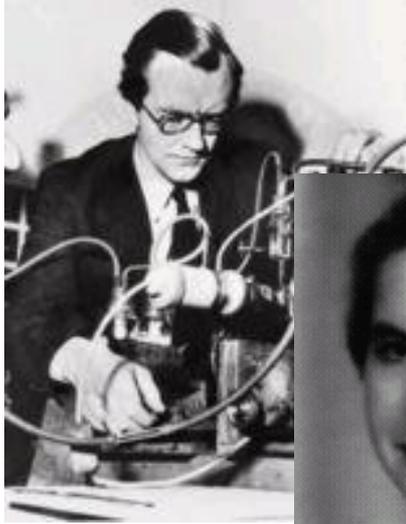


Sir William Henry Bragg  
(1862 - 1942)



William Lawrence Bragg  
(1890 - 1971)

Data analysis was not always that straightforward:

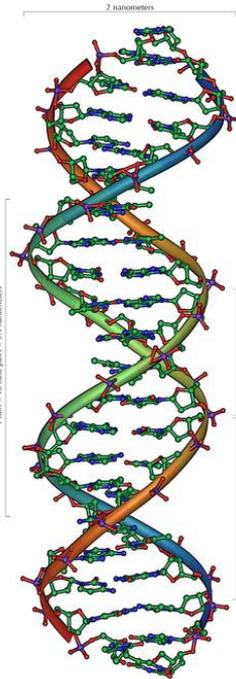
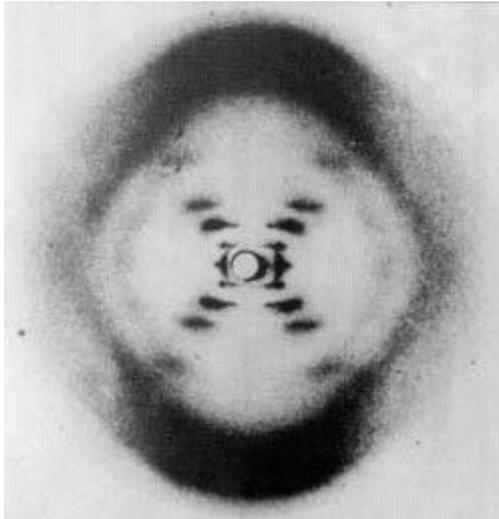


Maurice Wilkins  
Physicist

The Experimentalists (London):

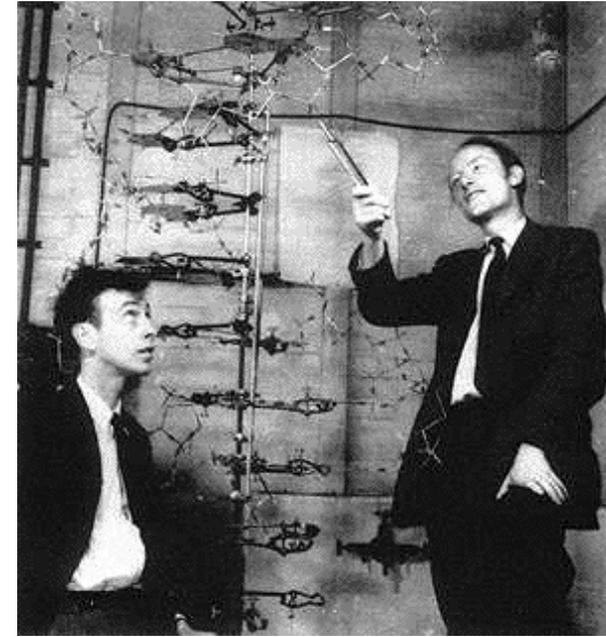


Rosalind Franklin  
Biophysicist



Measurement in Fourier space or  
momentum space

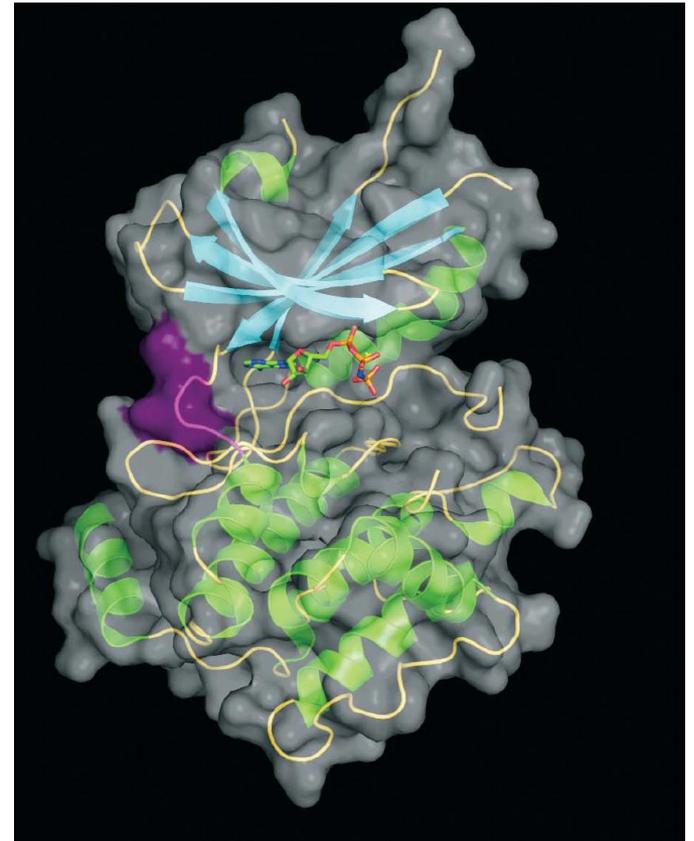
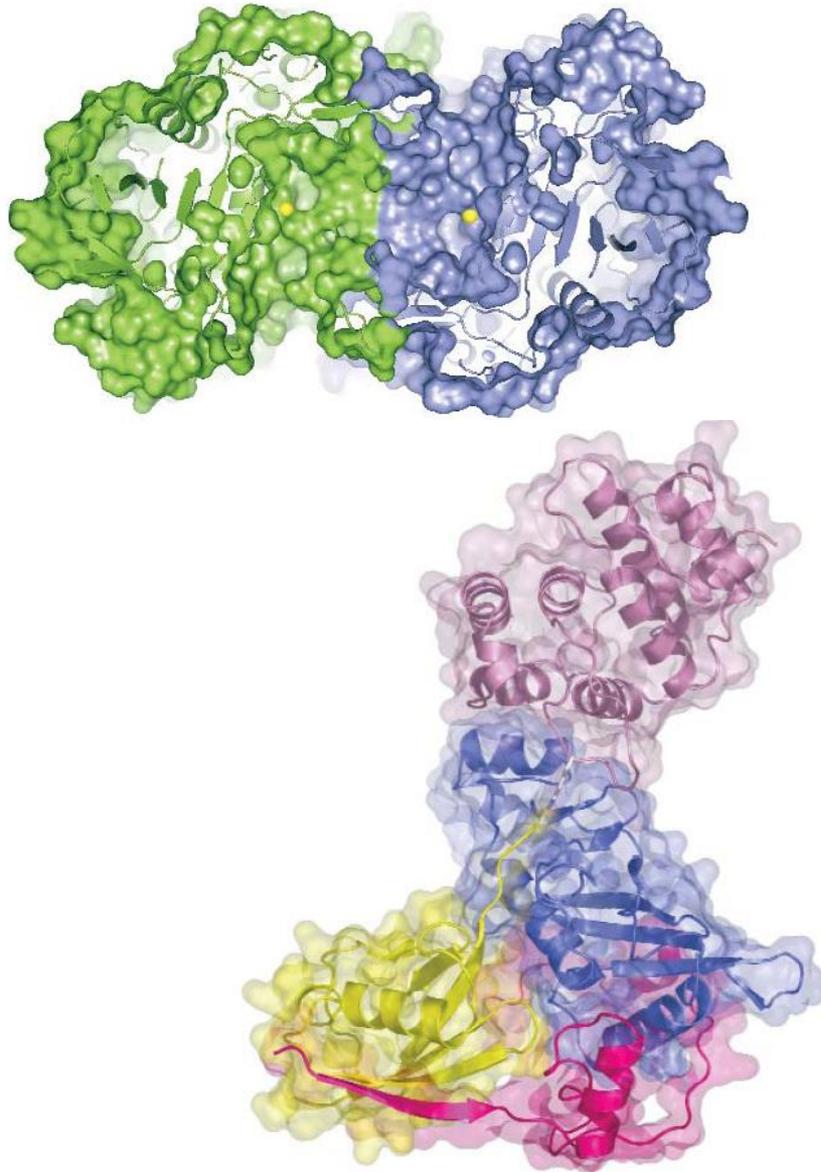
The theorists (Cambridge):



James Watson  
Biologist

Francis Crick  
Physicist

In the end, this field was taken over by biologists

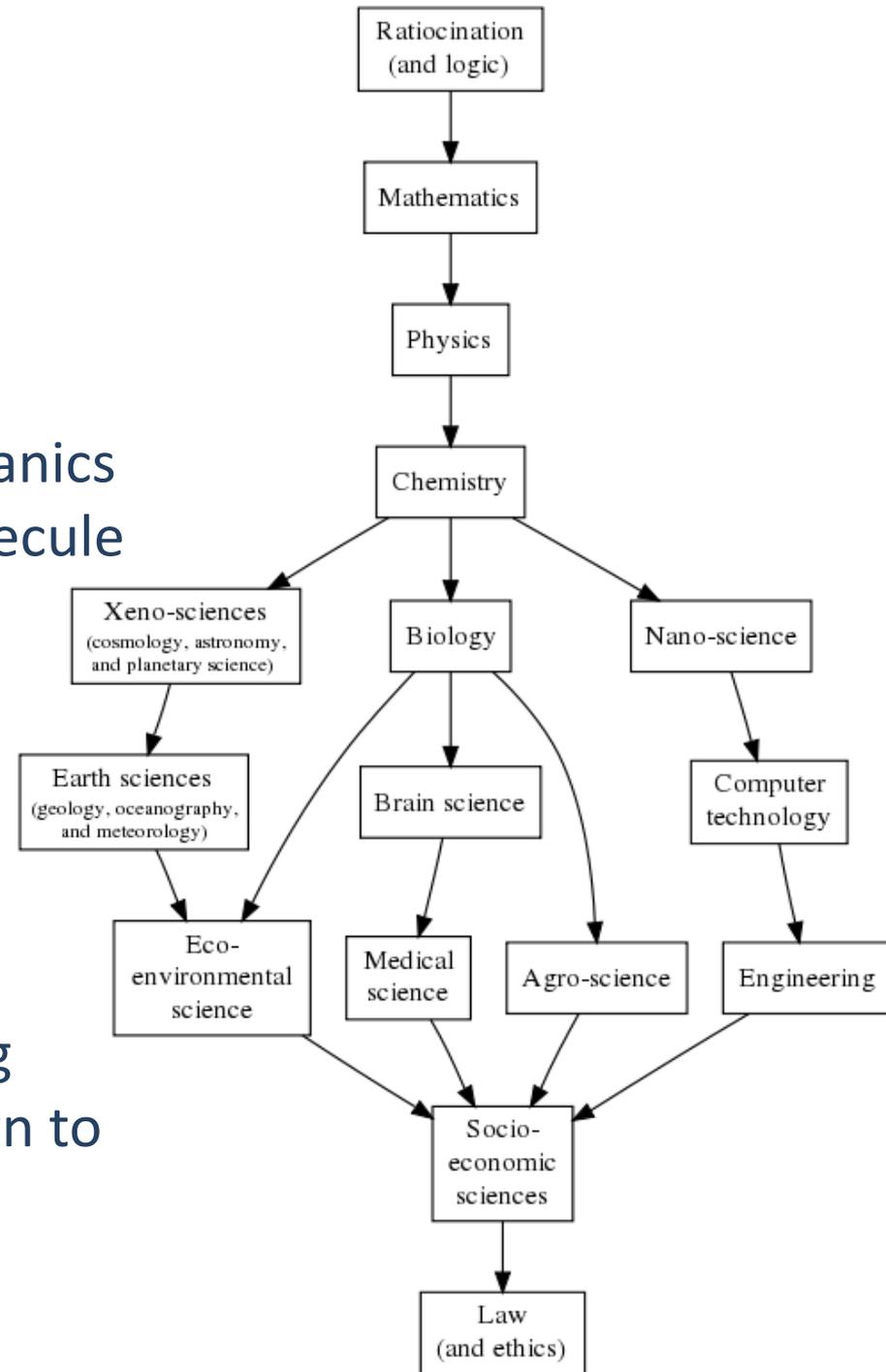


In the end, they are all doing natural philosophy, they just think about it differently.

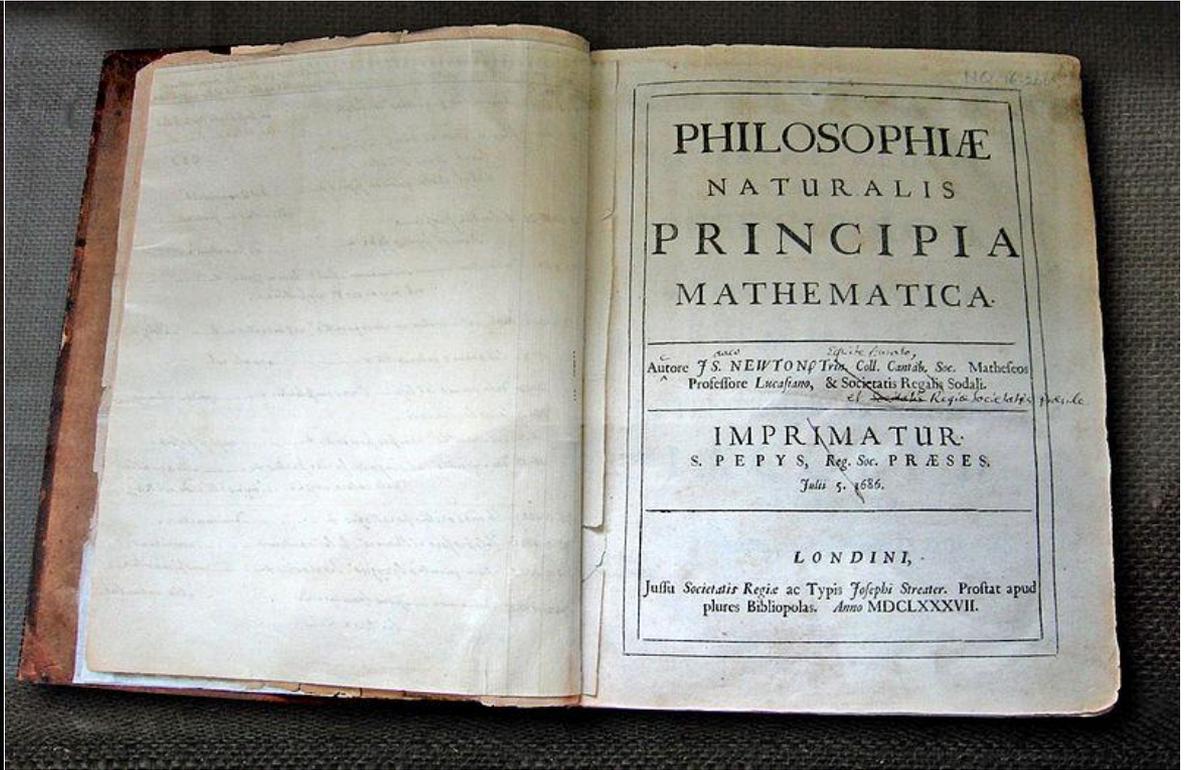
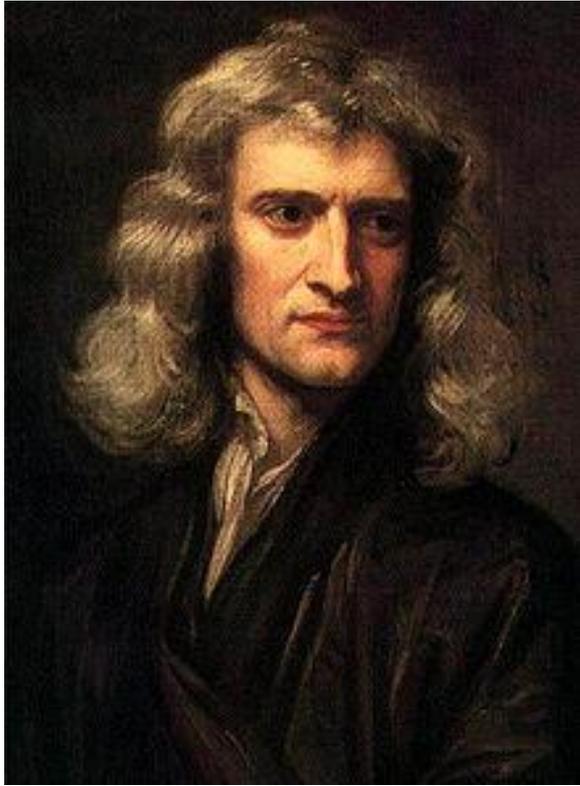
DNA=large molecule:  
Physicists see atoms, quantum mechanics and work their way up to a large molecule (biophysics)

Chemists see molecules and their reactions (biochemistry)

Biologists see cells, and basic building blocks of life and work their way down to molecules (molecular biology)



# Isaac Newton's definition of physics



*Philosophiæ Naturalis Principia Mathematica*, (Mathematical Principles of Natural Philosophy), or simply the *Principia* (1687)

So physics is natural philosophy based on mathematical principles

A physicist sees a natural phenomenon and wants to understand the basic underlying principles.

Understanding for a physicist means setting up a mathematical model/theory that fits real life/experiment

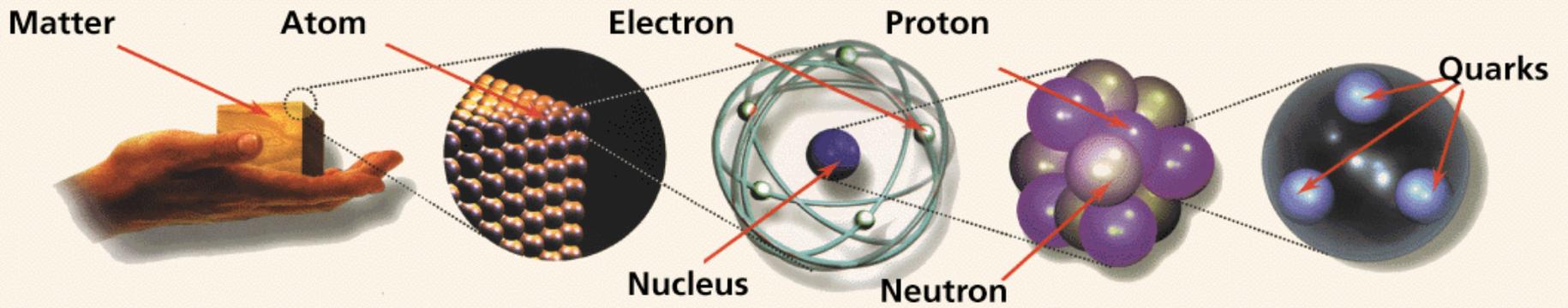
- A conceptual model is a good starting point, but it does not satisfy a physicist (for example, Darwin's theory of evolution is not a good physics model)
- A phenomenological model also does not make a physicist happy (for example, reaction dynamics in chemistry, although extremely useful, do not satisfy a physicist).

## Why is physics so hard (mathematically speaking)?

Because physicists do 'simple' things, but since they do it based on fundamental principles/laws, it get very complicated

For example, the structure of particles:

- Just quarks, electrons, certainly not as difficult as DNA or a living cell



**Matter Particles**

All ordinary particles belong to this group

These particles existed just after the Big Bang. Now they are found only in cosmic rays and accelerators.

LEPTONS		
FIRST FAMILY	<b>Electron</b> Responsible for electricity and chemical reactions; it has a charge of $-1$	<b>Electron neutrino</b> Particle with no electric charge, and possibly no mass; billions fly through your body every second
SECOND FAMILY	<b>Muon</b> A heavier relative of the electron; it lives for two-millionths of a second	<b>Muon neutrino</b> Created along with muons when some particles decay
THIRD FAMILY	<b>Tau</b> heavier still, it is extremely unstable. It was discovered in 1975	<b>Tau neutrino</b> Not yet discovered but believed to exist

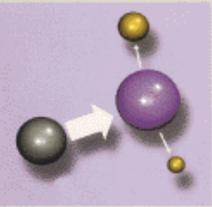
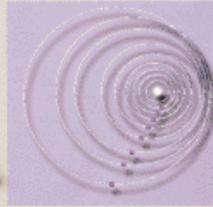
**QUARKS**

<b>Up</b> Has an electric charge of plus two-thirds; protons contain two, neutrons contain one		<b>Down</b> Has an electric charge of minus one-third; protons contain one, neutrons contain two	
<b>Charm</b> A heavier relative of the up; found in 1974		<b>Strange</b> A heavier relative of the down; found in 1974	
<b>Top</b> Heavier still, found in 1995		<b>Bottom</b> Heavier still; measuring bottom quarks is an important test of electroweak theory	

**Force Particles**

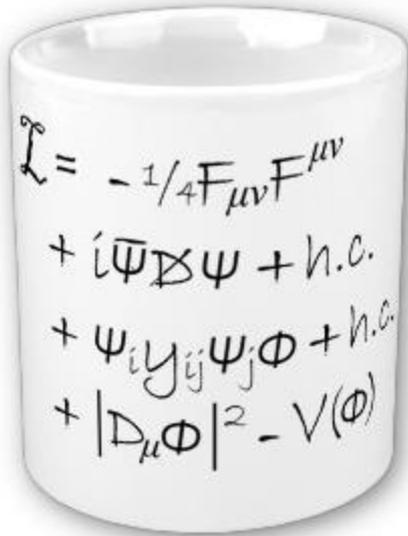
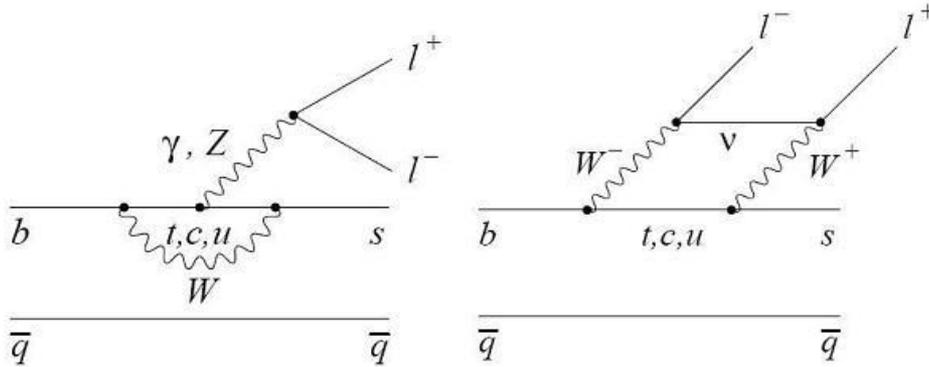
These particles transmit the four fundamental forces of nature although gravitons have so far not been discovered

<b>Gluons</b> Carriers of the <b>strong force</b> between quarks	 <p>Felt by: quarks</p>	<b>Photons</b> Particles that make up light; they carry the <b>electromagnetic force</b>	 <p>Felt by: quarks and charged leptons</p>
The explosive release of nuclear energy is the result of the <b>strong force</b>		Electricity, magnetism and chemistry are all the results of the <b>electromagnetic force</b>	

<b>Intermediate vector bosons</b> Carriers of the <b>weak force</b>	 <p>Felt by: quarks and leptons</p>	<b>Gravitons</b> Carriers of <b>gravity</b>	 <p>Felt by: all particles with mass</p>
Some forms of radioactivity are the result of the <b>weak force</b>		All the weight we experience is the result of the <b>gravitational force</b>	

Of course, that's a pretty poster

However, for a physicist to say (s)he knows the standard model, because (s)he memorized the poster, is like



$$\mathcal{L}_{GWS} = \sum_f (\bar{\Psi}_f (i\gamma^\mu \partial_\mu - m_f) \Psi_f - e Q_f \bar{\Psi}_f \gamma^\mu \Psi_f A_\mu) +$$

$$+ \frac{g}{\sqrt{2}} \sum_i (\bar{a}_L^i \gamma^\mu b_L^i W_\mu^+ + \bar{b}_L^i \gamma^\mu a_L^i W_\mu^-) + \frac{g}{2c_w} \sum_f \bar{\Psi}_f \gamma^\mu (I_f^3 - 2s_w^2 Q_f - I_f^3 \gamma_5) \Psi_f Z_\mu +$$

$$-\frac{1}{4} |\partial_\mu A_\nu - \partial_\nu A_\mu - ie(W_\mu^- W_\nu^+ - W_\mu^+ W_\nu^-)|^2 - \frac{1}{2} |\partial_\mu W_\nu^+ - \partial_\nu W_\mu^+ +$$

$$-ie(W_\mu^+ A_\nu - W_\nu^+ A_\mu) + ig' c_w (W_\mu^+ Z_\nu - W_\nu^+ Z_\mu)|^2 +$$

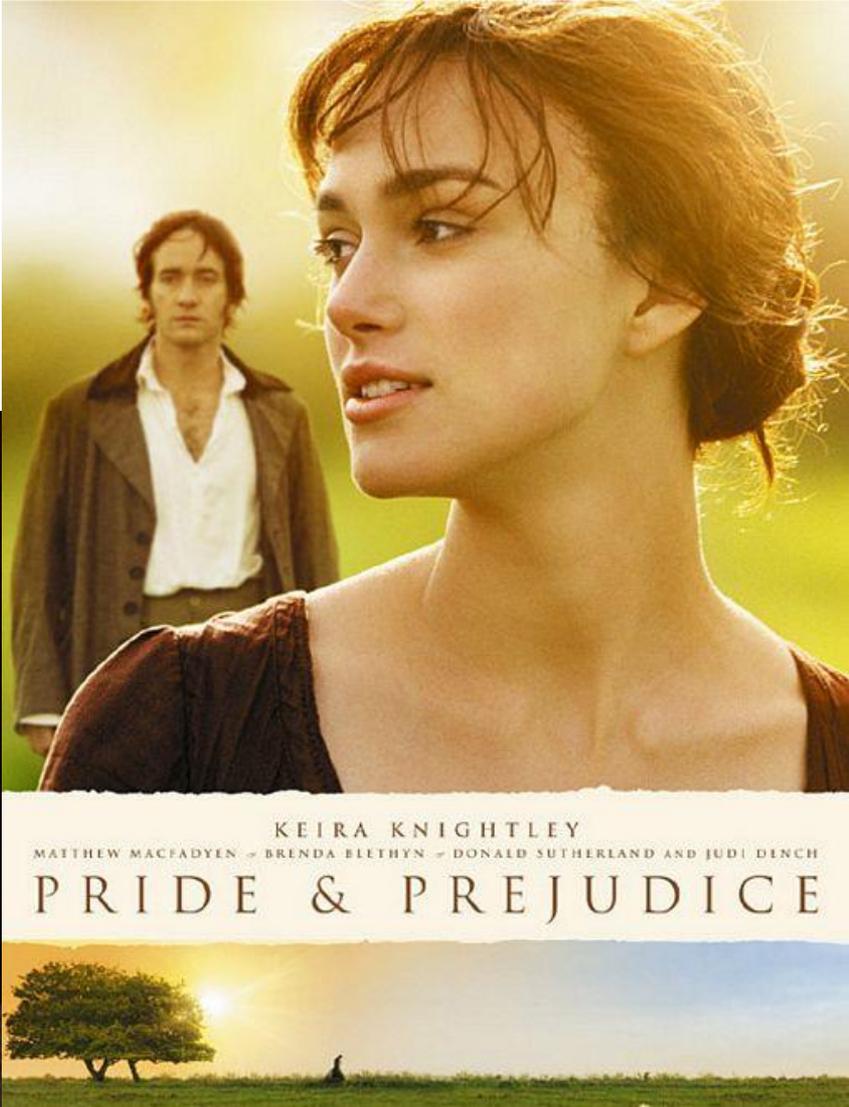
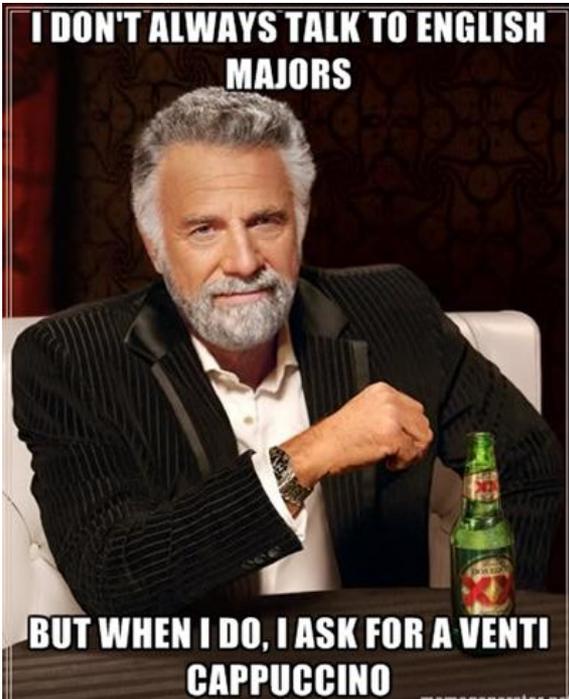
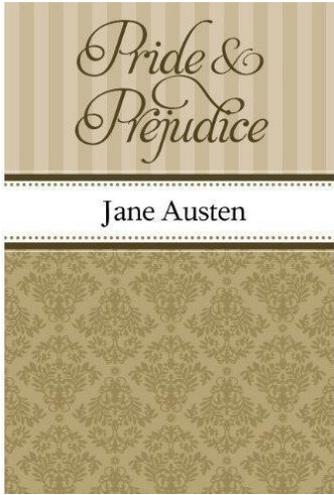
$$-\frac{1}{4} |\partial_\mu Z_\nu - \partial_\nu Z_\mu + ig' c_w (W_\mu^- W_\nu^+ - W_\mu^+ W_\nu^-)|^2 +$$

$$-\frac{1}{2} M_\eta^2 \eta^2 - \frac{g M_\eta^2}{8 M_W} \eta^3 - \frac{g'^2 M_\eta^2}{32 M_W} \eta^4 + |M_W W_\mu^+ + \frac{g}{2} \eta W_\mu^+|^2 +$$

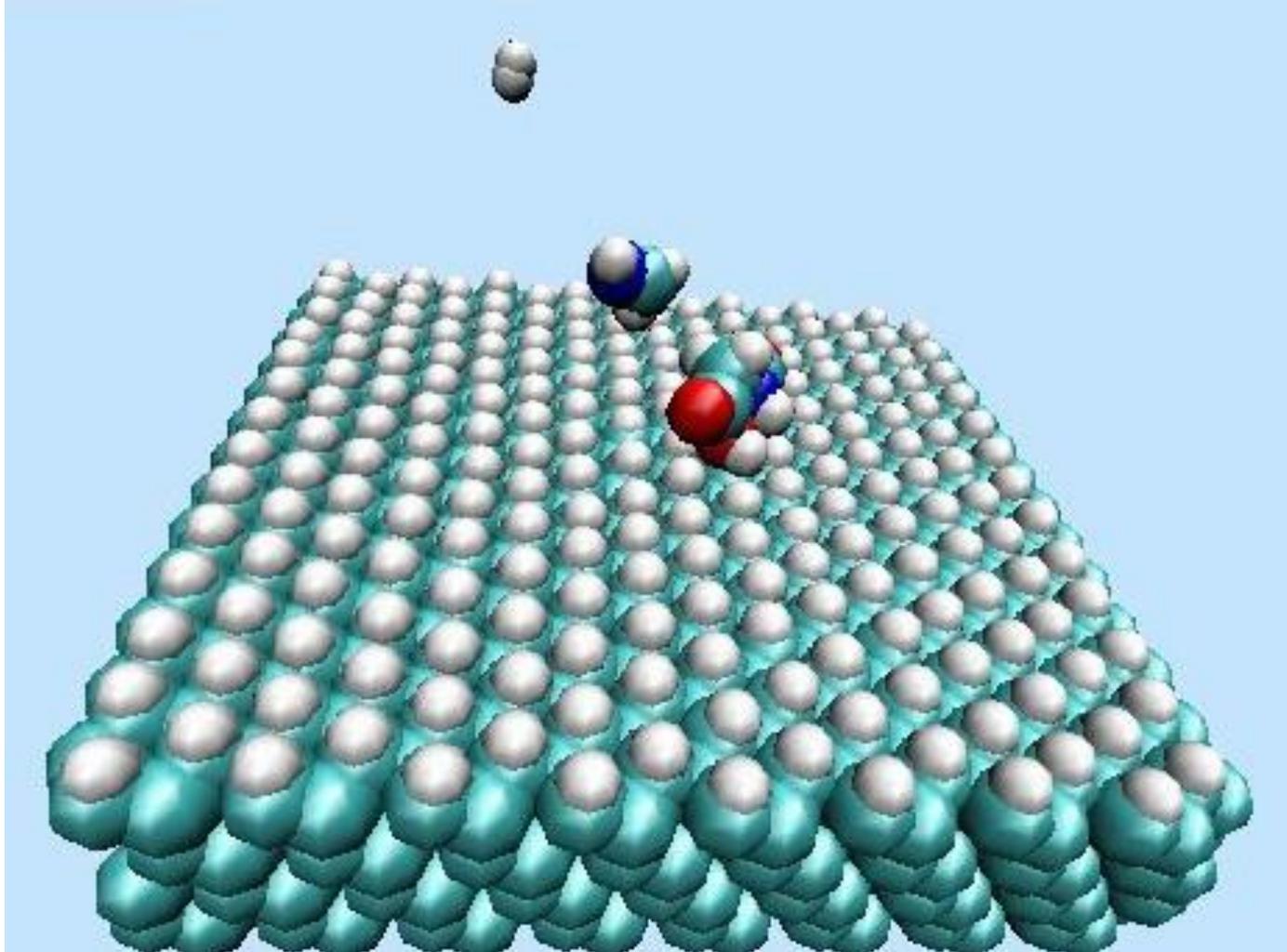
$$+\frac{1}{2} |\partial_\mu \eta + i M_Z Z_\mu + \frac{ig}{2c_w} \eta Z_\mu|^2 - \sum_f \frac{g}{2} \frac{m_f}{M_W} \bar{\Psi}_f \Psi_f \eta$$

Don't worry we won't do anything of this sort...

Is like an English major saying the know Jane Austen,  
when (s)he has just seen the movie

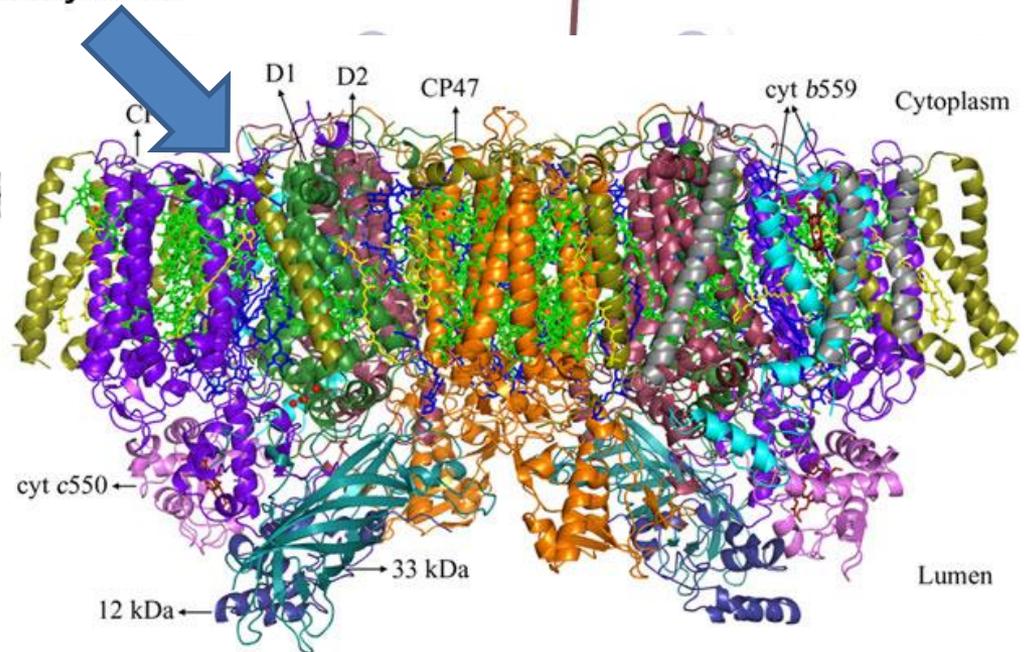
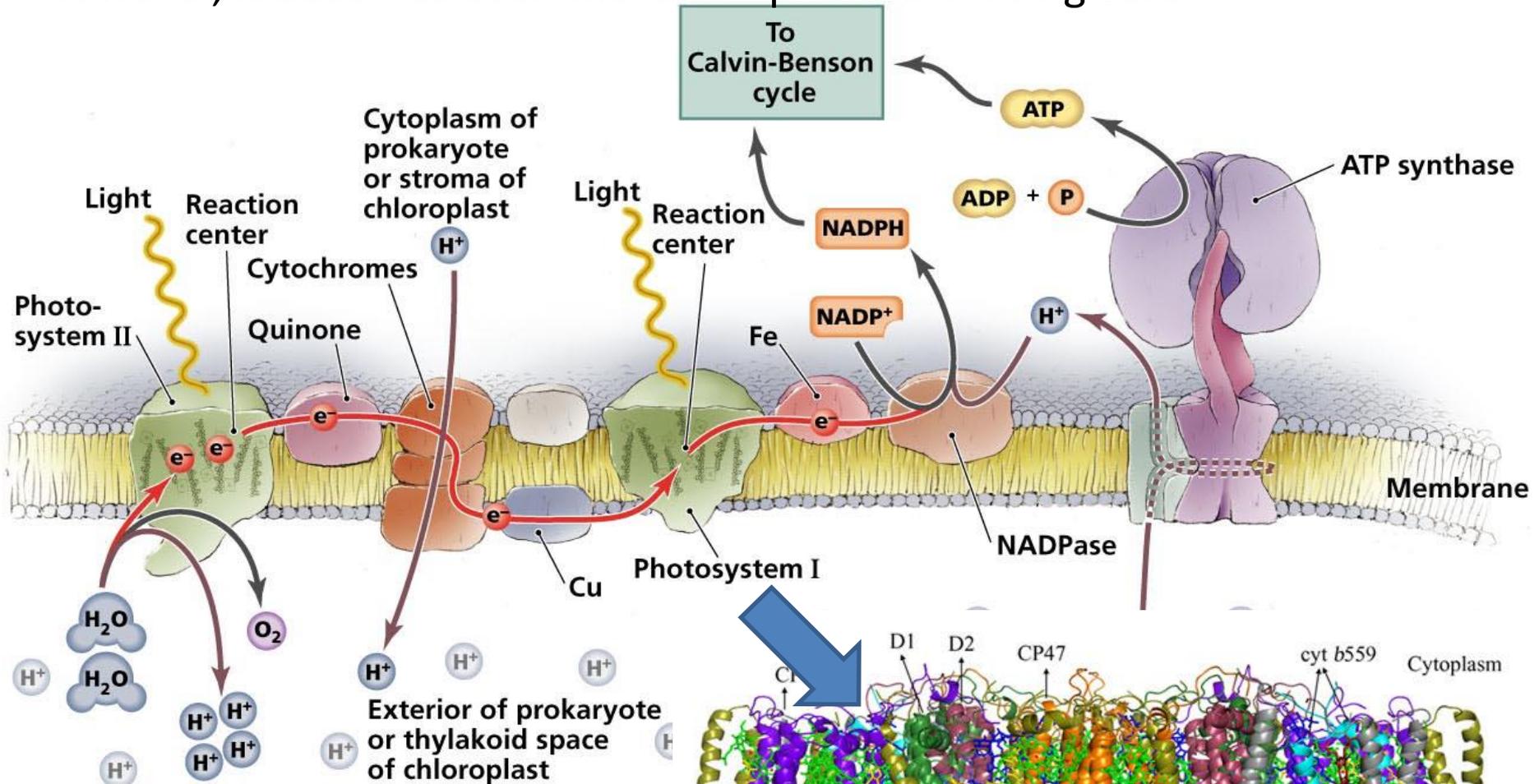


However, chemical reactions are more than just quantum chemistry: there is kinetics (surfaces, temperature, pressure, activation energies), thermodynamics, bond breaking, catalysts, energy landscapes, etc



This often requires phenomenology: chemistry

However, it becomes even more complicated in living cells



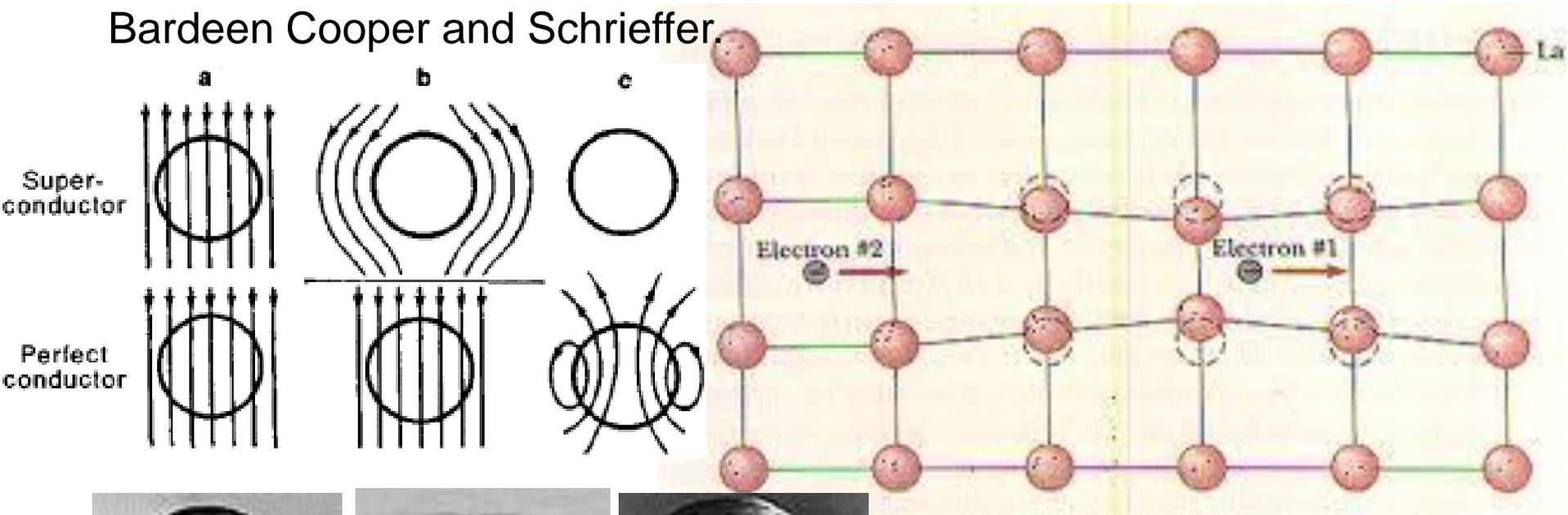
How do plants use sun light to break water and then turn it into sugars?

This is not ready yet for a physics bottom-up approach

In physics, understanding how interaction can lead to emerging phenomena is a very difficult problem

Superconductivity: BCS theory, 1972 Nobel prize

Bardeen Cooper and Schrieffer.



# Discovery of superconductivity

Kamerlingh Onnes in Leiden

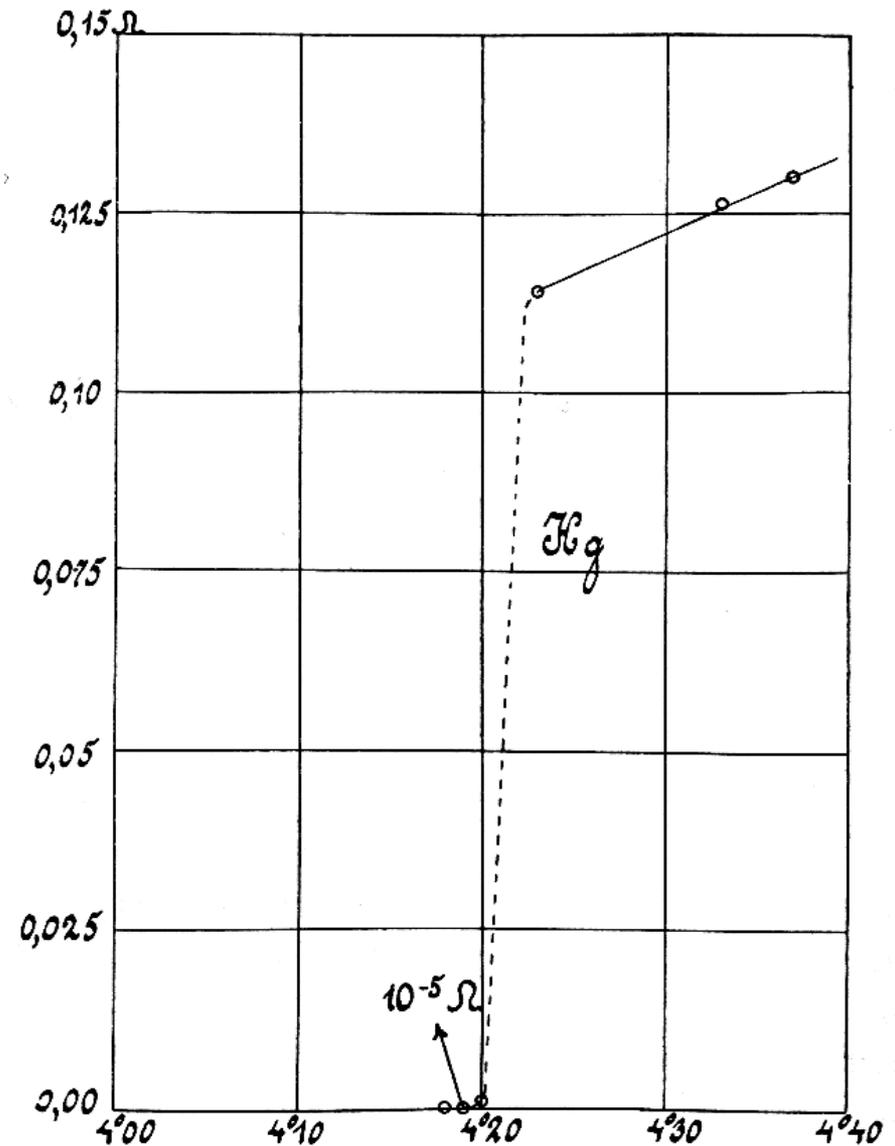
in 1911

Mercury (Hg),  $T_c \sim 4$  Kelvin

Liquid helium: 4.2 Kelvin



1913, Nobel Prize



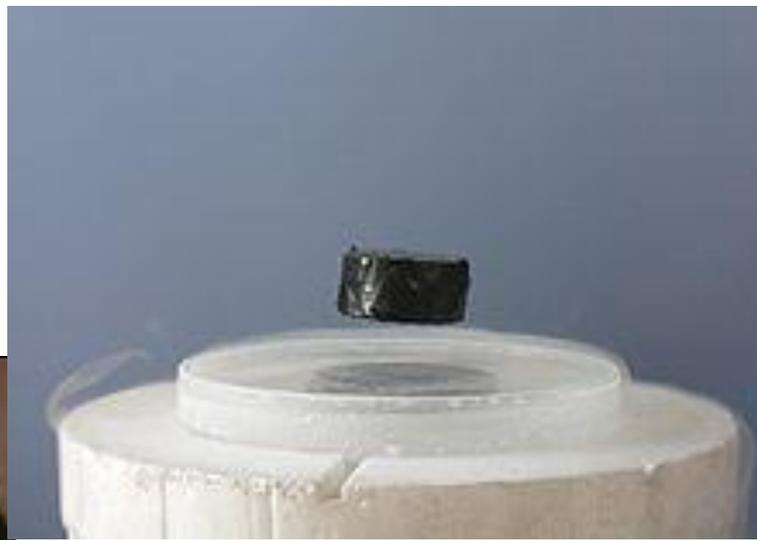
## Discovery of high-T<sub>c</sub> superconductivity

J. G. Bednorz and K. A. Müller

in IBM Zurich

Nobel prize: in 1988

La-Sr-Cu-O, T<sub>c</sub> ~ 30 Kelvin



1989, Nobel Prize,

No explanations yet!



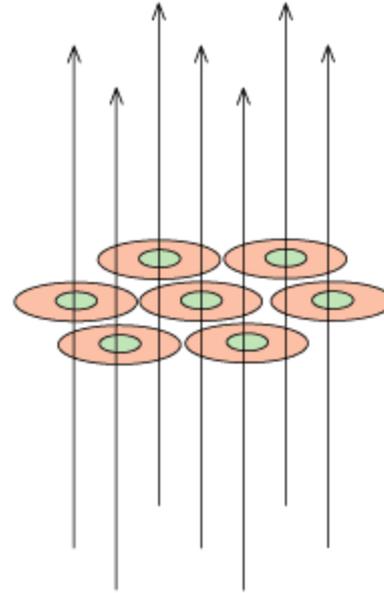
Could revolutionize magnetic levitation trains

# Ginzburg, Leggett, and Abrikosov, Nobel prize 2003

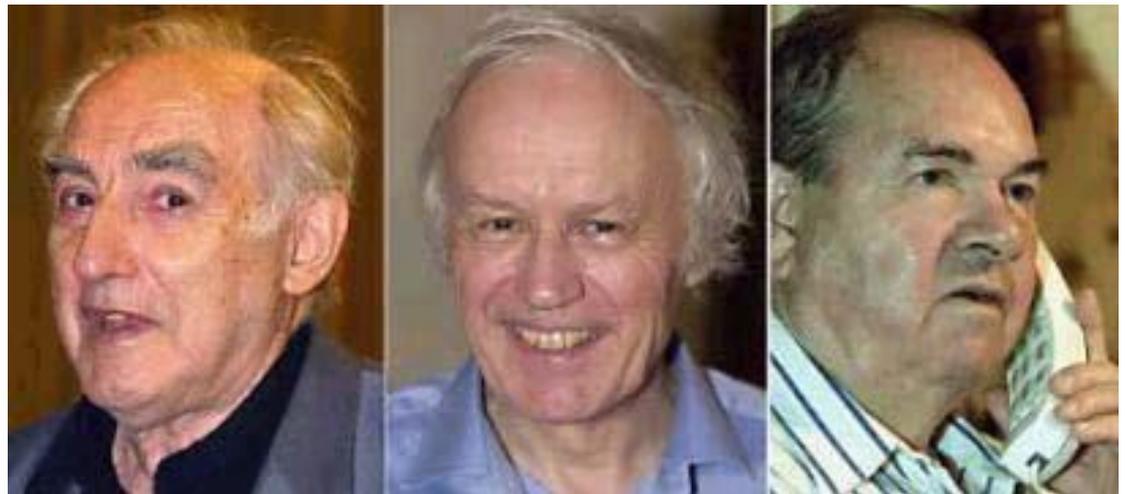
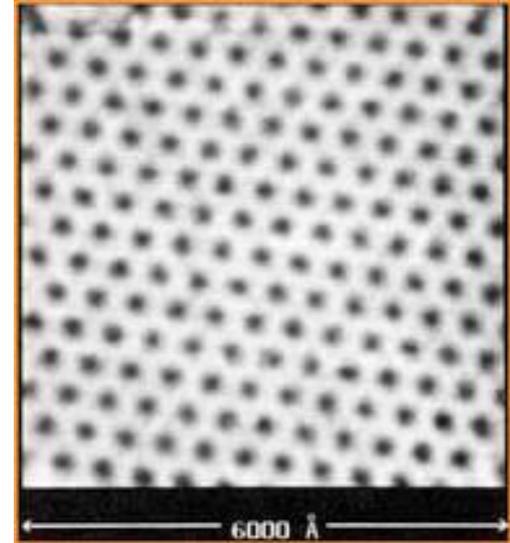
Superfluidity, superconductivity



Lev Landau, 1962

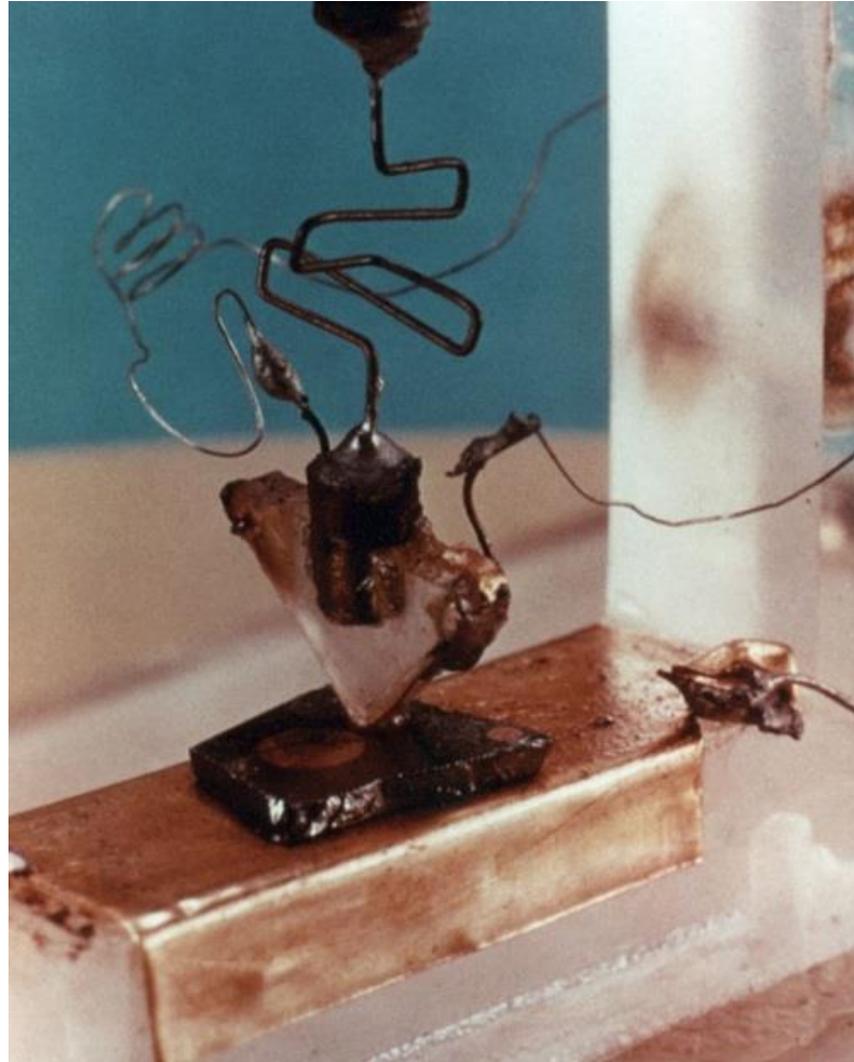


(a)vortex lattice



The physicist bottom-up approach also works in other areas

## WHAT IS THIS?



# Discovery of transistors: Engineering

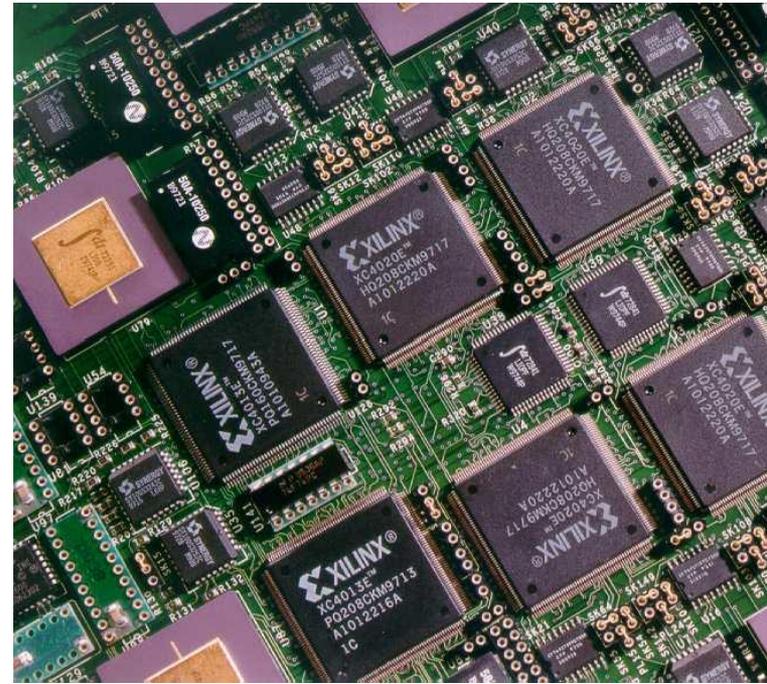


**First Germanium transistor  
in 1947**

Bardeen, Brattain, Shockley (All physicists)

Bell Labs

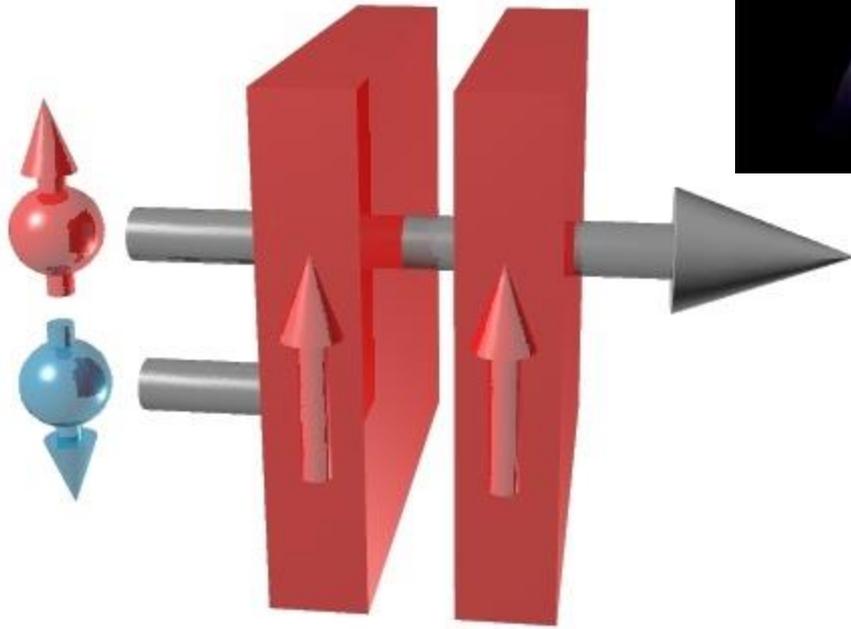
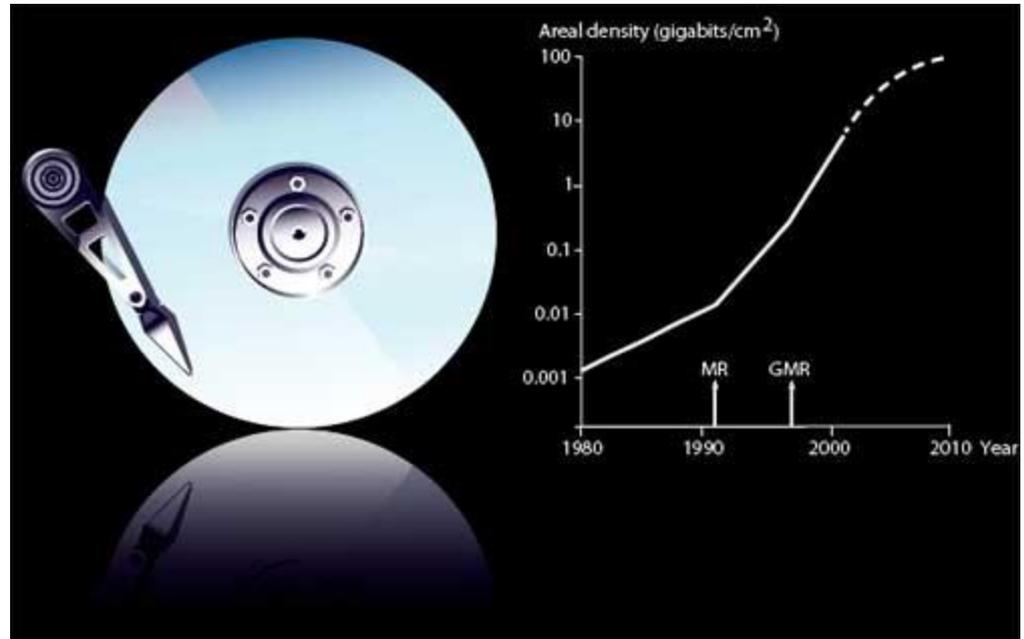
1956, Nobel prize



Then it turns into engineering

Microelectronics: 20<sup>th</sup> century's greatest achievement in Solid State Physics

# Albert Fert and Peter Grunberg, Nobel prize: 2007

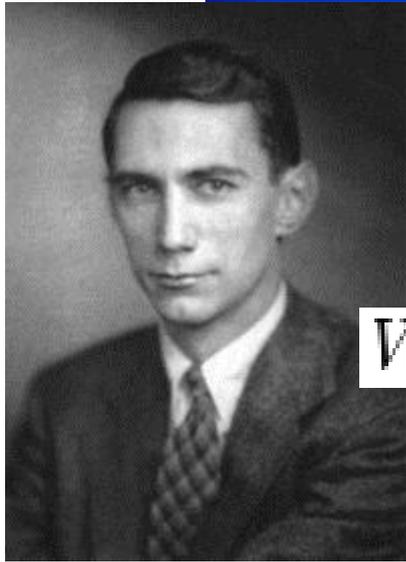


# Information theory: What are the fundamental limits on signal processing

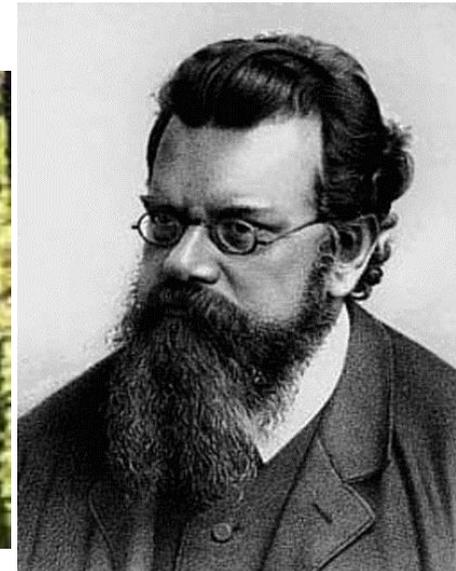


$$S = k \cdot \log W$$

2<sup>nd</sup> law of thermodynamics  
Entropy



$$W = K \log m$$



Ludwig Boltzmann (1844-1906)

Claude Shannon (1916-2001)

Does a physics way of thinking work in other areas, or only in the natural sciences?

The first Nobel prize winner in economics in 1969 was a physicist: Jan Tinbergen (he was also Dutch)

He established the field of econometrics and set up the first national macroeconomic model



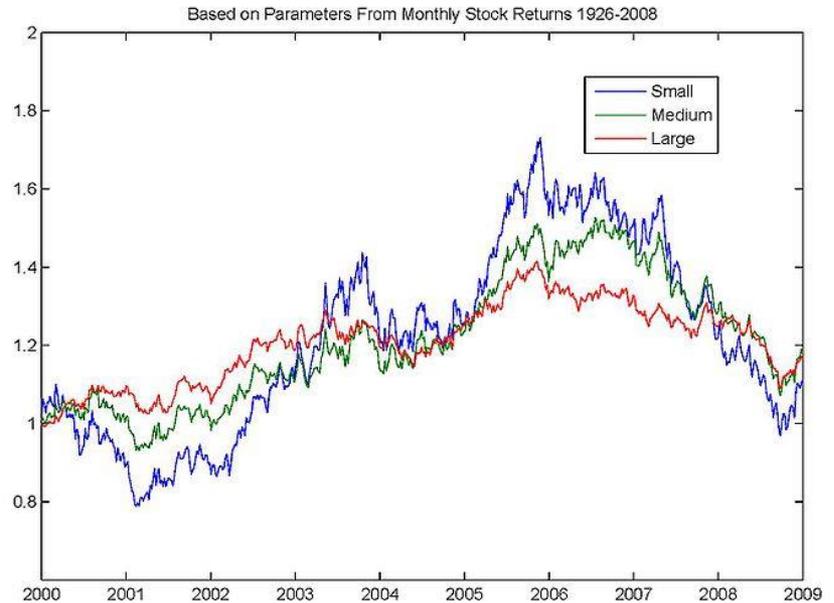
# Black-Scholes model of option pricing (Nobel prize 1997)



Myron Scholes  
Economy

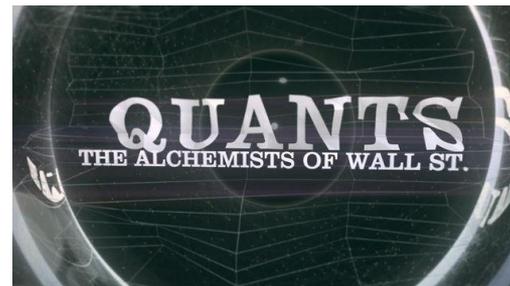
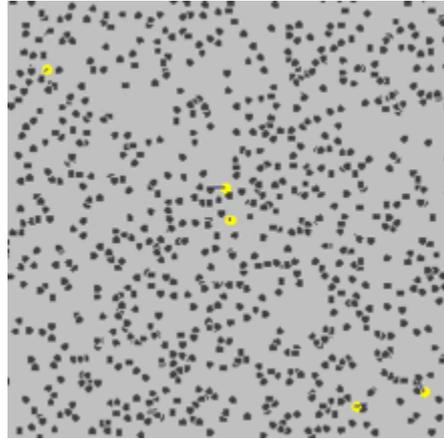
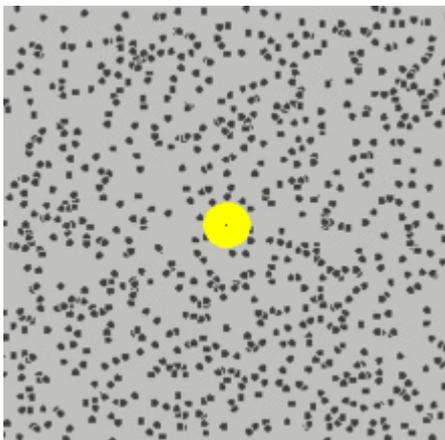
Fischer Black,  
Applied mathematics  
(switched from physics)

Simulations of Small, Medium, and Large-Cap Stock Prices



$$\frac{\partial V}{\partial t} + \frac{1}{2}\sigma^2 S^2 \frac{\partial^2 V}{\partial S^2} + rS \frac{\partial V}{\partial S} - rV = 0.$$

Which is related to  
Brownian motion  
which was explained  
by Albert Einstein



Speaking of stocks....

Just for those lovers of useless facts:

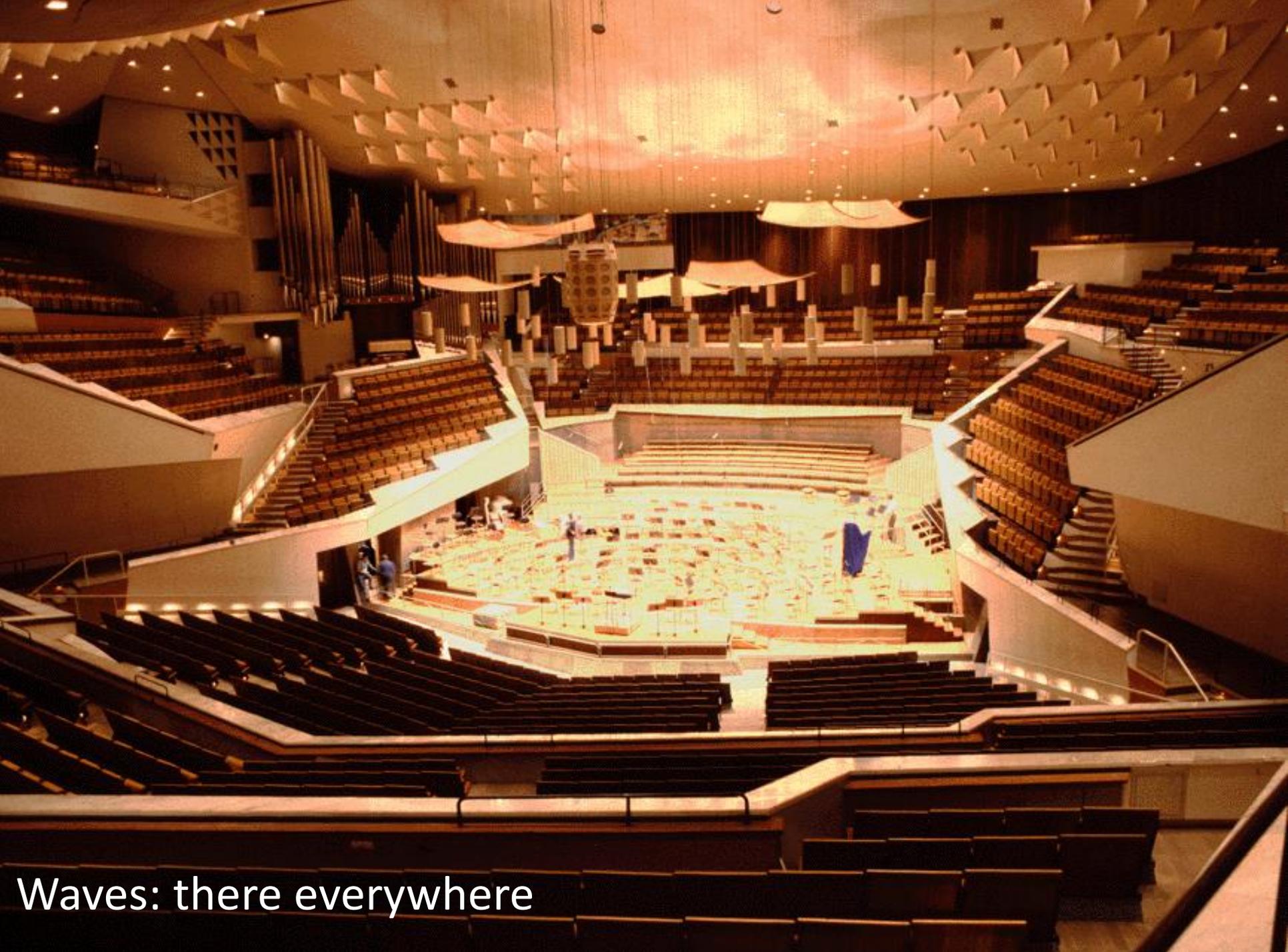
Which was the first publicly traded joint-stock company?

1602: Vereenigde Oost-Indische Compagnie (VOC)  
Dutch East India Company  
Trading in derivatives, options, etc.



Even pioneered short-selling. Forbidden in 1610

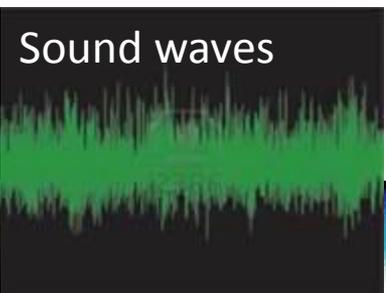
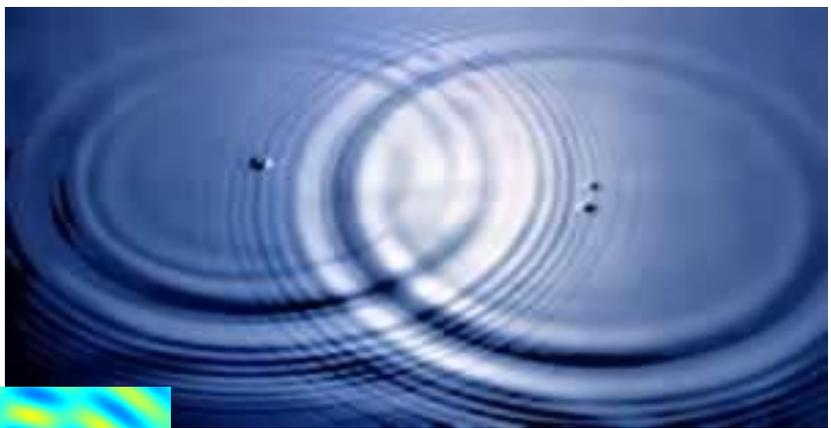
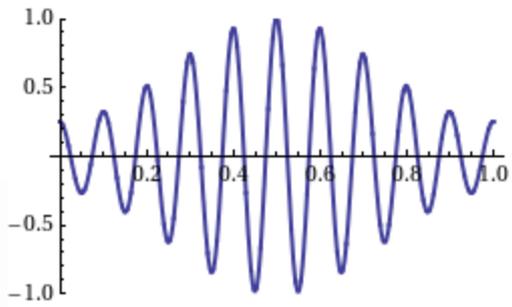
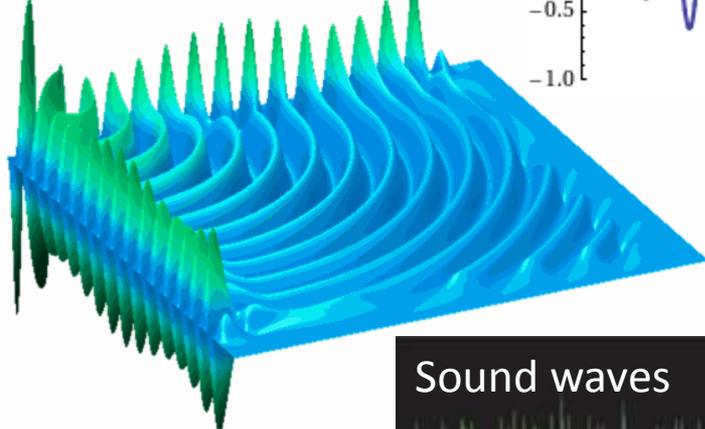




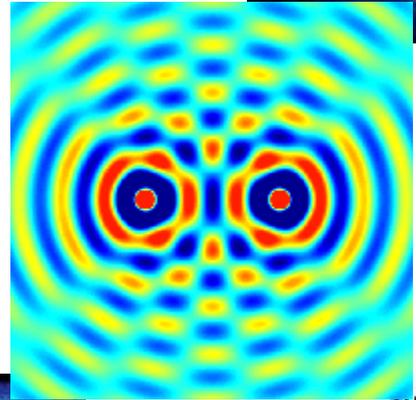
Waves: there everywhere

# WAVES

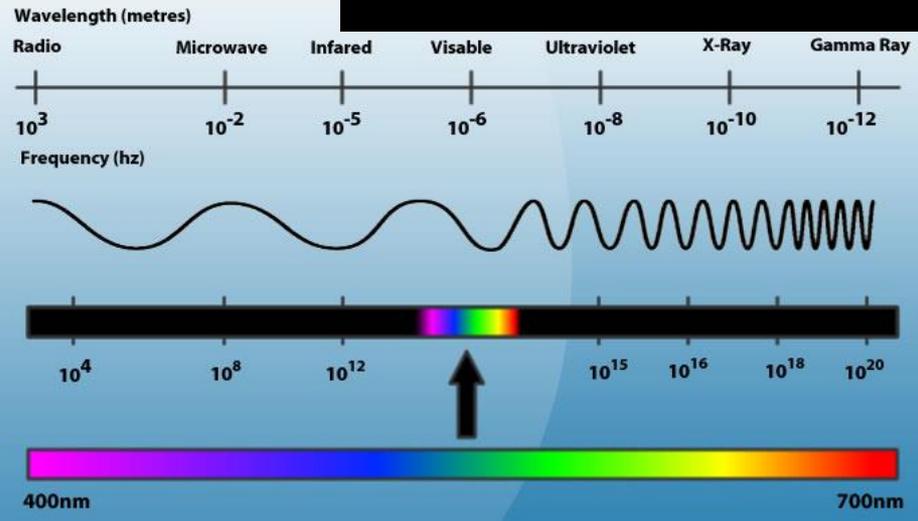
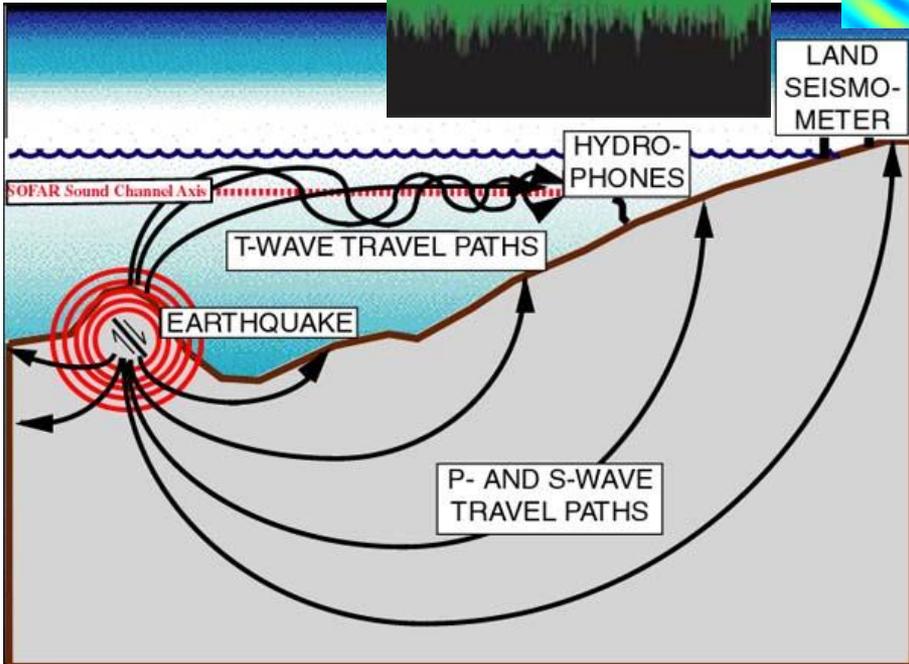
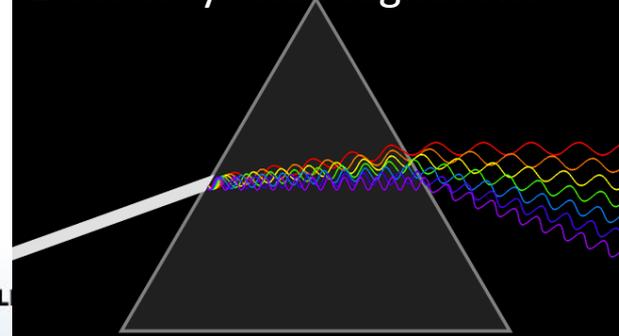
## Quantum mechanics



Sound waves



## Electricity and magnetism



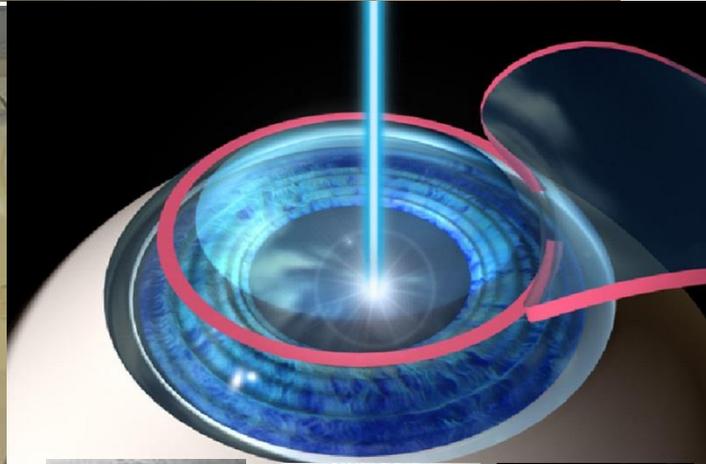
Many medical devices have basic physics breakthroughs at their origin



Rontgen



Bloch



Pierre and Marie Curie

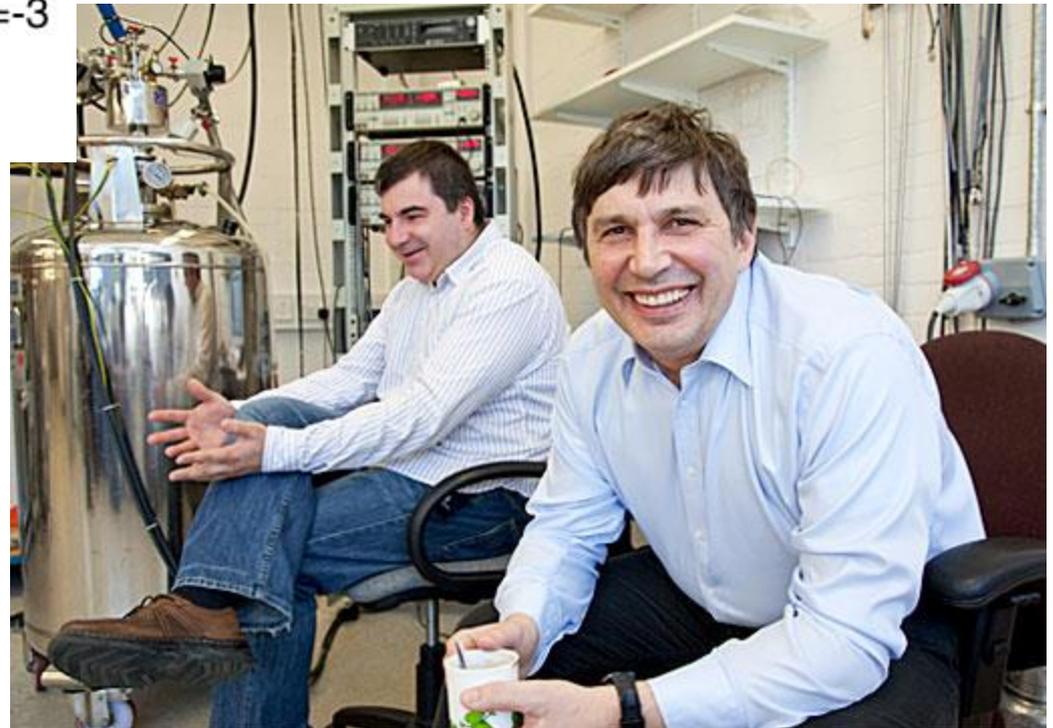
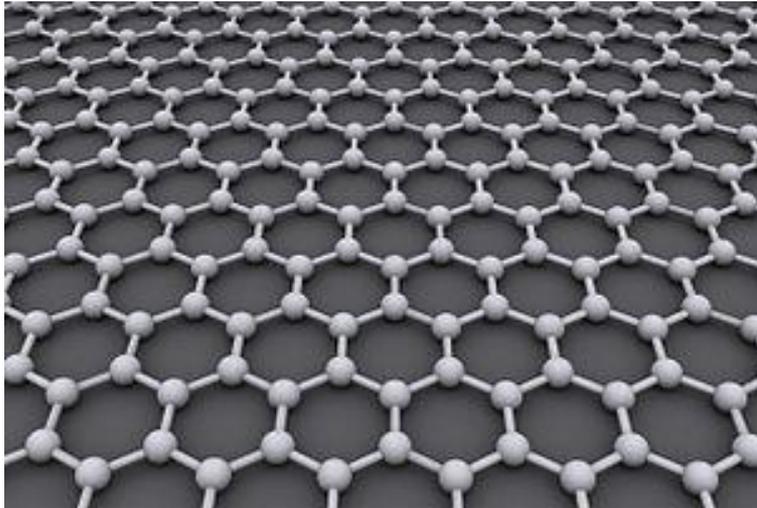
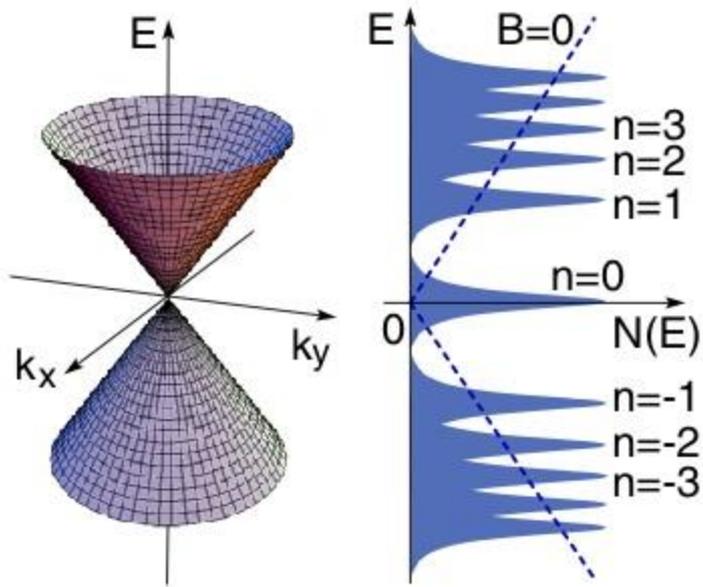


Basov

Prokhorov

Townes

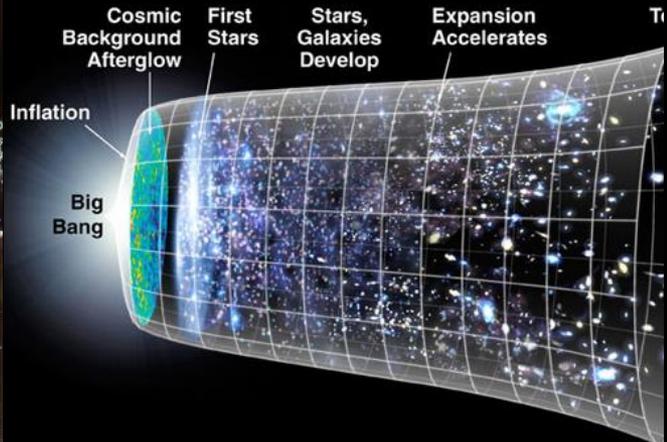
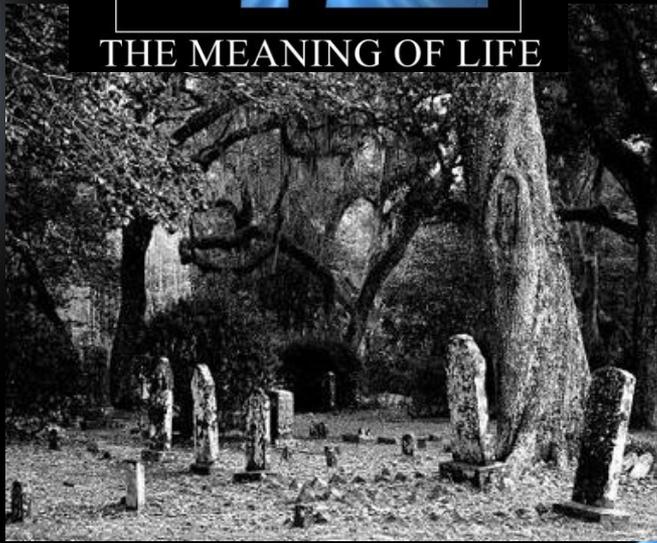
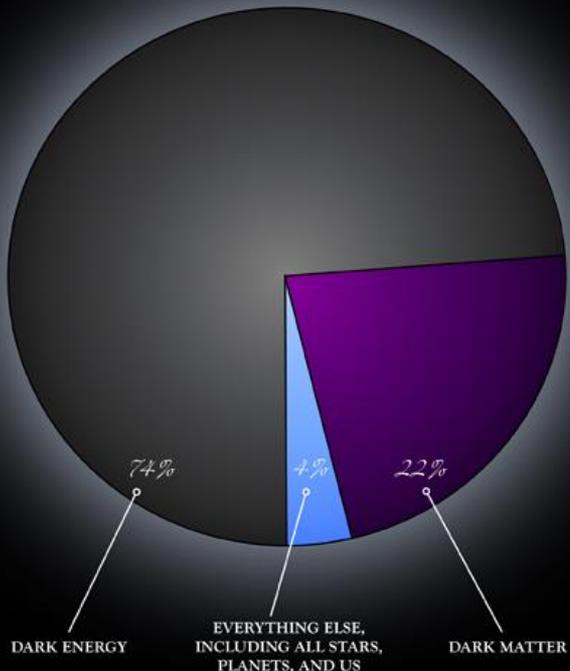
# Andre Geim and Konstantin Novoselov Nobel prize: 2010



# There are still many questions that cannot be answered

# 42

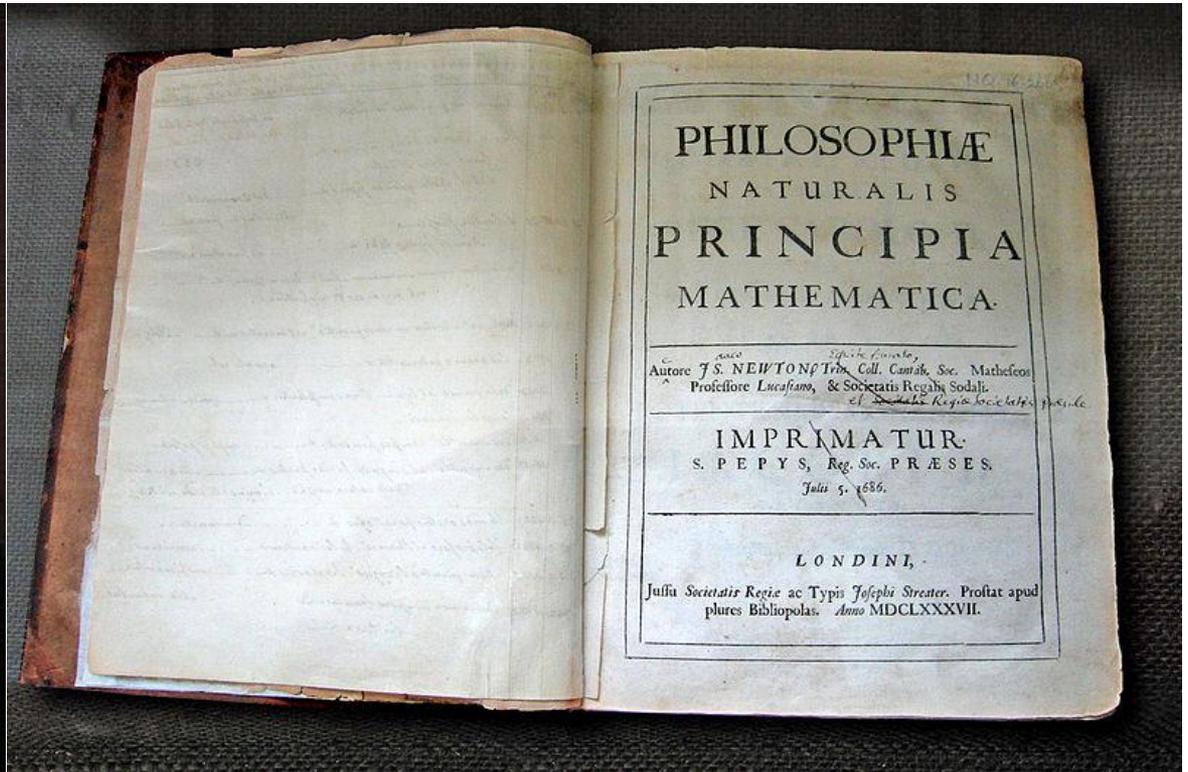
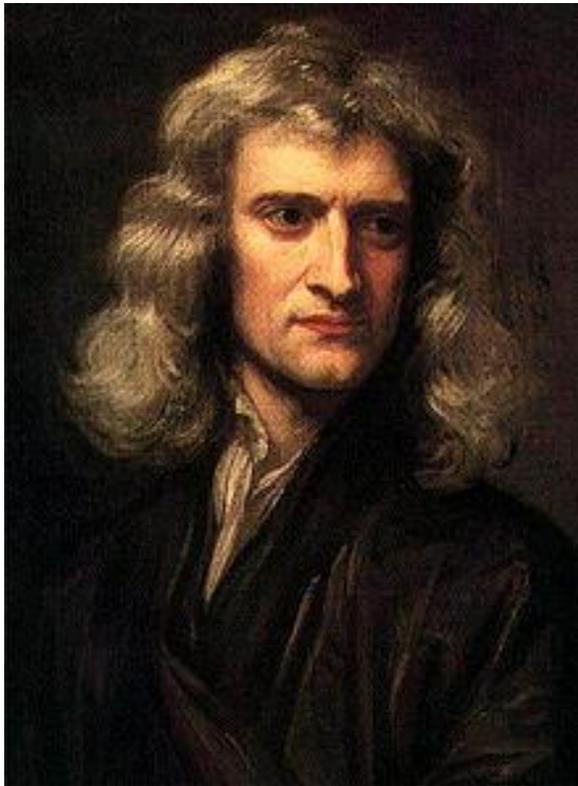
## THE MEANING OF LIFE



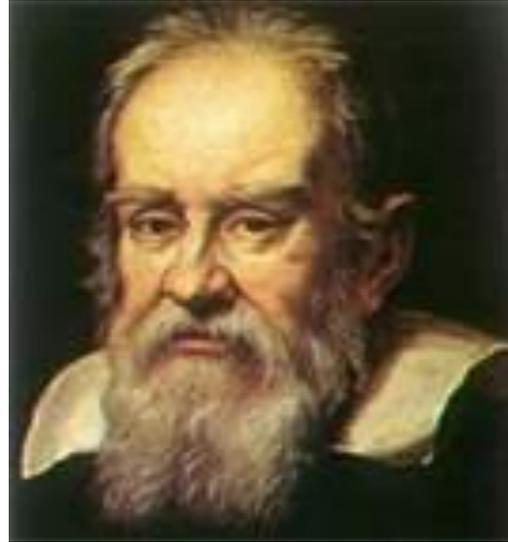
Unfortunately, we are not ready for all of that....

So we start of with classical mechanics

We begin with Newton's three laws



# Make excursions back in time to Aristotle and Galileo



See how our concepts of motion have change in time.  
And get an idea why incorrect theories can persist for so long!

We look at constant velocity



→ x

We look at constant acceleration



We look at Newton's second law

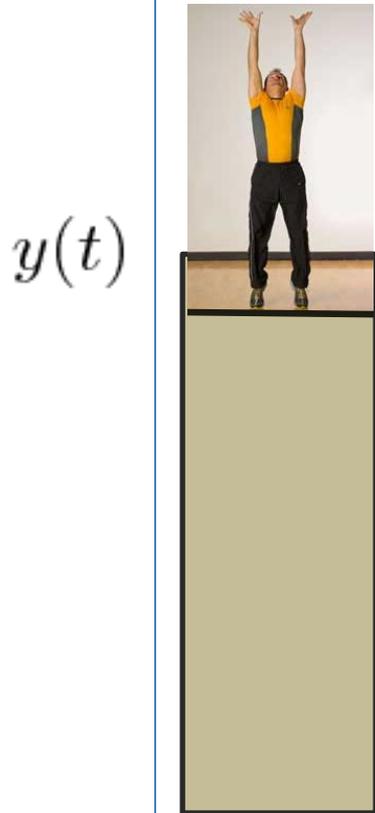
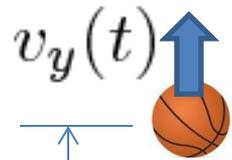
$$F=ma$$



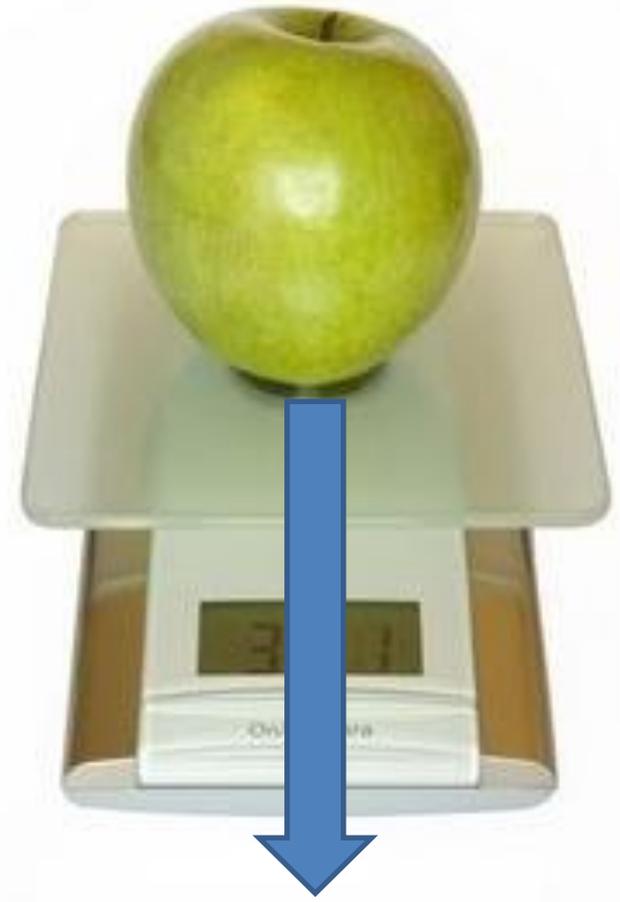
**INERTIA**

DEF: TO STAY IN MOTION UNTIL ACTED UPON BY AN EXTERNAL FORCE.

We look at gravity and how objects fall in a gravitational acceleration



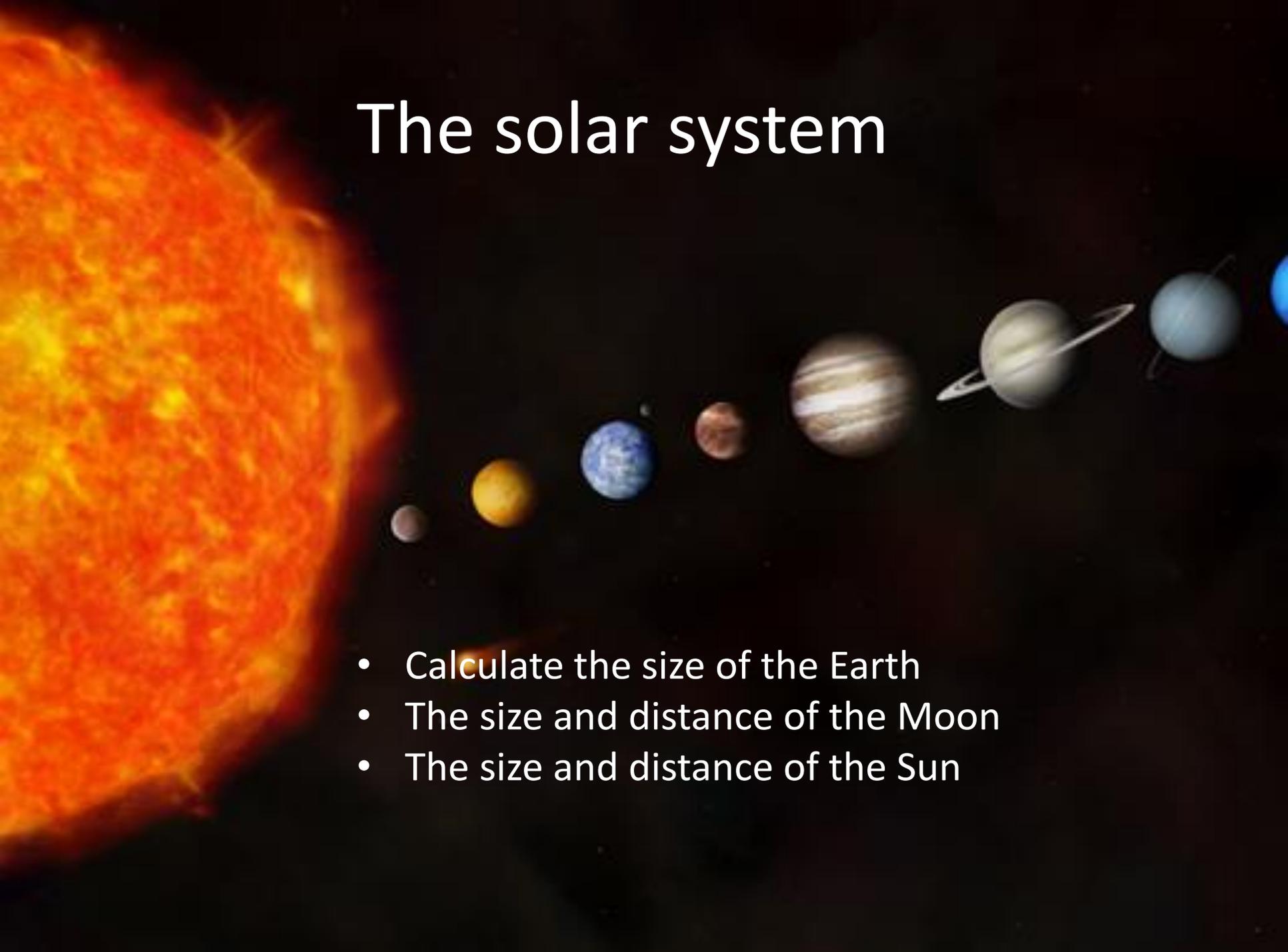
We look at weight



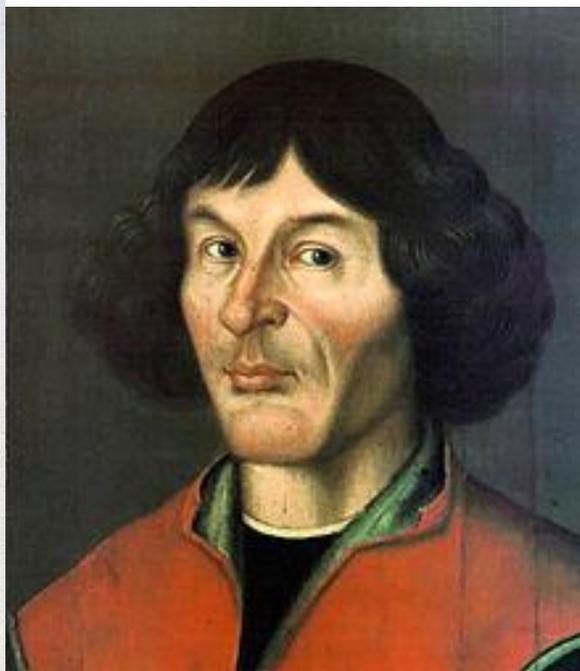
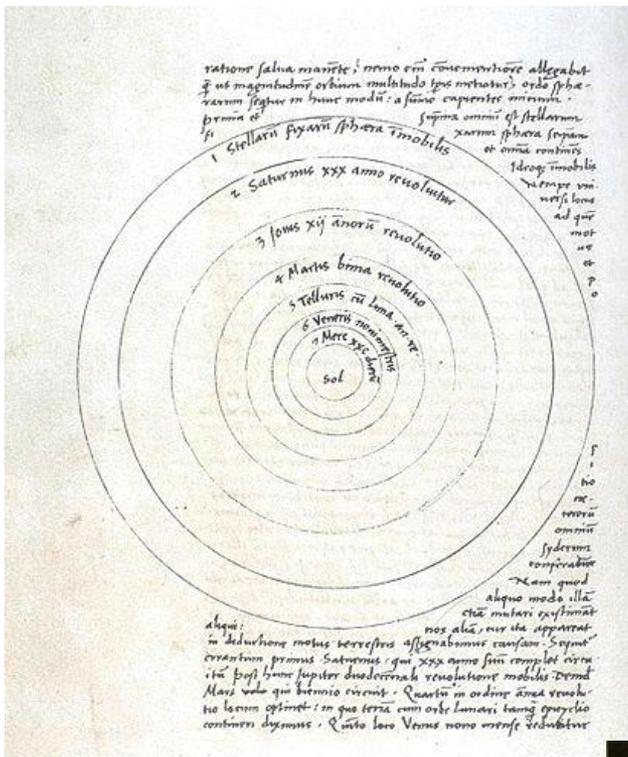
And weightlessness....



# The solar system

- 
- Calculate the size of the Earth
  - The size and distance of the Moon
  - The size and distance of the Sun

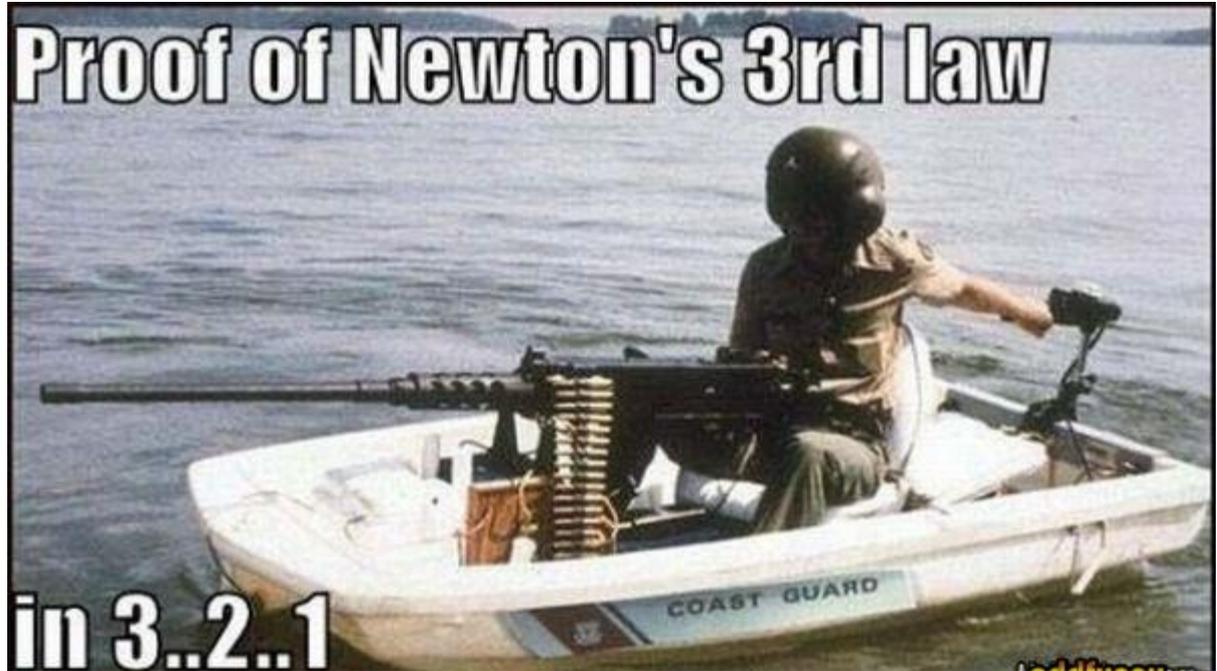
# We look at models of our Solar system



# And study Kepler's laws



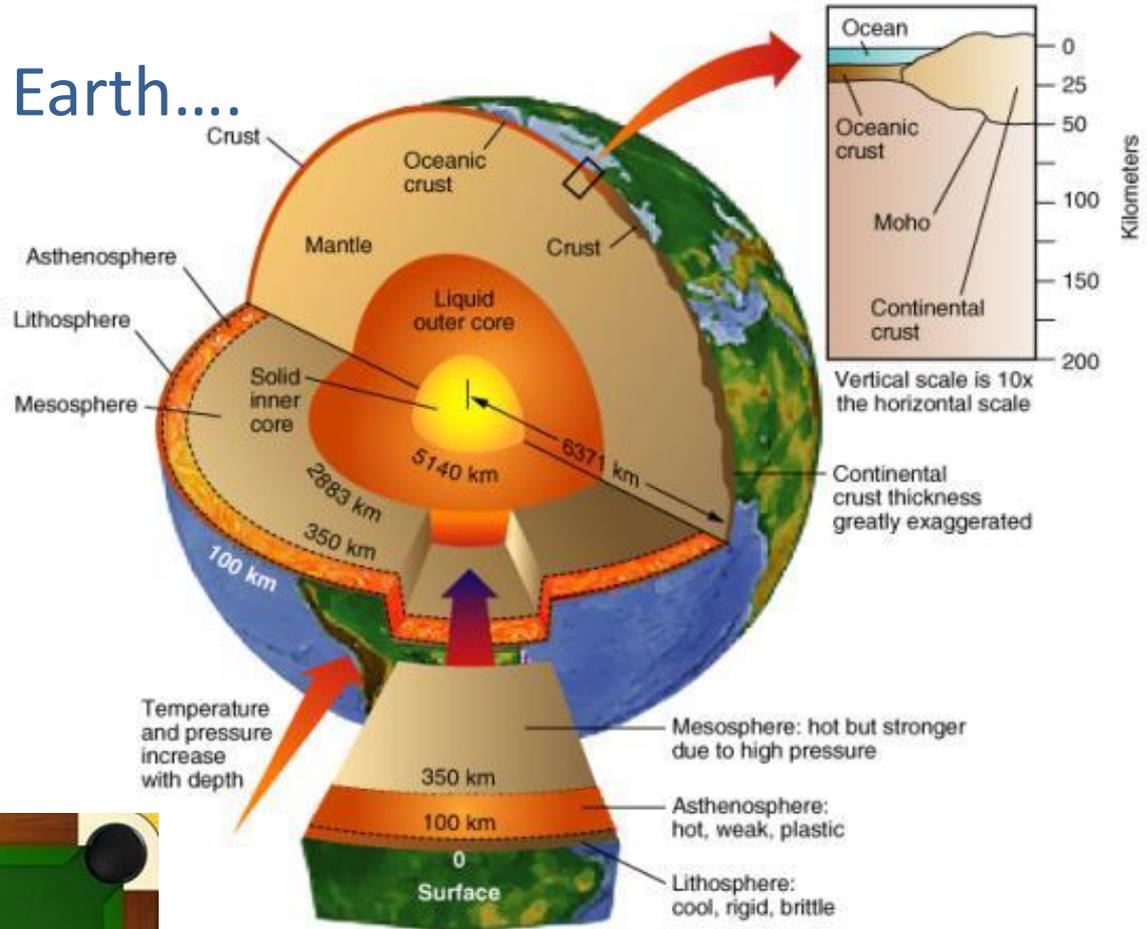
We look at Newton's third law.



Study the tides



Have a look inside the Earth....



Play some pool....

(okay, that sounds more fun than it is)

We look at the reason why certain physical quantities are conserved



An we end the classical part  
with pressure.



And after that.....  
I don't know yet.