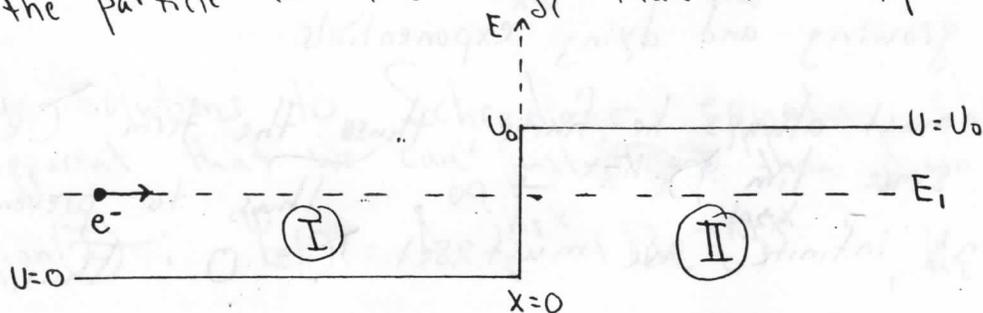


# Physics 360 - Lecture Notes # 2

## III Scattering at a Step Potential

Let's look at the problem of a particle scattering from a step when the particle has less energy than the step height:



Since the \$e^-\$ does not have enough energy to make it over the step, then Region II is a classically forbidden region. Let's write down and solve the Schrodinger equation in regions (I) + (II).

In Region (I):

$$-\frac{\hbar^2}{2m} \frac{d^2 \Psi_I}{dx^2} + U(x) \Psi_I = E \Psi_I \Rightarrow \frac{d^2 \Psi_I}{dx^2} = \frac{-2mE}{\hbar^2} \Psi_I = -k_I^2 \Psi_I$$

The solution should be quite familiar by now:

$$\Psi_I(x) = A e^{i k_I x} + B e^{-i k_I x}, \quad k_I = \sqrt{\frac{2mE}{\hbar^2}}$$

In Region (II):

$$-\frac{\hbar^2}{2m} \frac{d^2 \Psi_{II}}{dx^2} + U(x) \Psi_{II} = E \Psi_{II} \Rightarrow \frac{d^2 \Psi_{II}}{dx^2} = -\frac{2m}{\hbar^2} (E - U_0) \Psi_{II} = -k_{II}^2 \Psi_{II}$$

Note that \$E - U\_0 < 0\$, then ~~we~~  $k_{II} = \sqrt{\frac{2m}{\hbar^2} (E - U_0)} = \sqrt{\frac{2m}{\hbar^2} (-1)(U_0 - E)}$

$$= \sqrt{-1} \sqrt{\frac{2m}{\hbar^2} (U_0 - E)}$$

$$= i \sqrt{\frac{2m}{\hbar^2} (U_0 - E)}$$

$$= i \alpha_{II}$$

The solution is

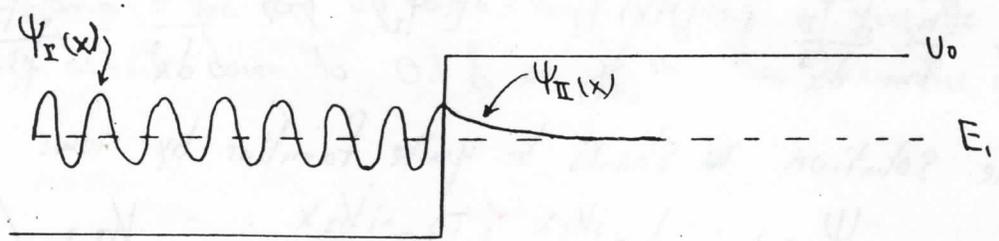
$$\begin{aligned} \Psi_{II}(x) &= C e^{i k_{II} x} + D e^{-i k_{II} x} = (C e^{i(\alpha_{II})x} + D) e^{-i(\alpha_{II})x} \\ &= C e^{-\alpha_{II} x} + D e^{\alpha_{II} x} \end{aligned}$$

which are growing and dying exponentials.

But,  $\Psi_{II}(x)$  must always be finite, thus the term  $D e^{\alpha_{II} x}$  is not allowed since  $\lim_{x \rightarrow \infty} D e^{\alpha_{II} x} \rightarrow \infty$ . Thus, to prevent  $\Psi_{II}(x)$  from becoming infinite, we must set  $D = 0$ . Thus, in Region II:

$$\Psi_{II}(x) = C e^{-\alpha_{II} x} \quad (\text{a dying exponential})$$

Surprisingly, there is a finite probability for the particle to exist in the classically forbidden region. A plot of  $\Psi_I, \Psi_{II}$  follows:



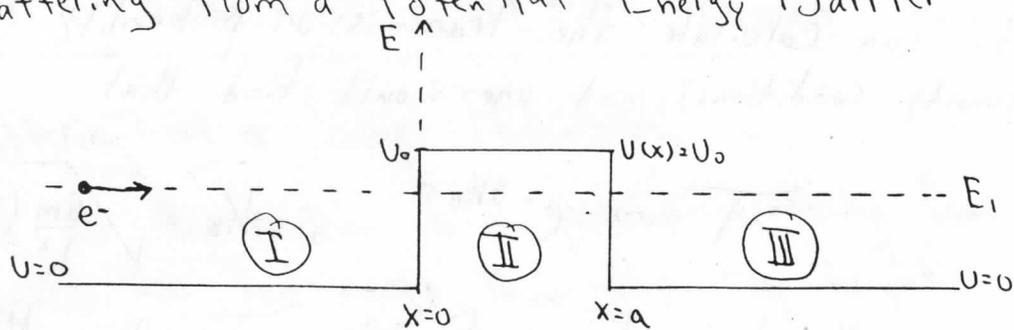
The depth of penetration of the particle into the classically forbidden region is characterized by a  $(1/e)$  length, that is, the value of  $x$  at which the probability drops by  $1/e$ :

$$\text{Probability} = |\Psi_{II}|^2 \sim e^{-2\alpha_{II} x_e} = \frac{1}{e} = e^{-1}$$

$$\begin{aligned} 2\alpha_{II} x_e &= 1 \\ x_e &= \frac{1}{2\alpha_{II}} = \frac{\hbar}{2\sqrt{2m(U_0 - E)}} = \text{penetration depth} \end{aligned}$$

Such behavior allows particles to penetrate barriers, as we will see in the next scattering problem.

# IV Scattering from a Potential Energy Barrier



The solutions to Schrodinger's equation should now be so apparent that we can just write them down:

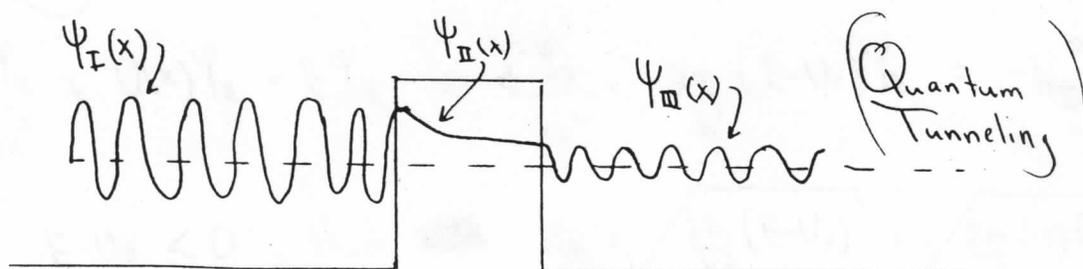
Region I:  $\Psi_I(x) = Ae^{ik_I x} + Be^{-ik_I x}$ ,  $k_I = \sqrt{\frac{2mE}{\hbar^2}}$

Region II:  $\Psi_{II}(x) = Ce^{-\alpha_{II} x} + De^{\alpha_{II} x}$ ,  $\alpha_{II} = \sqrt{\frac{2m(U_0 - E)}{\hbar^2}}$

Region III:  $\Psi_{III}(x) = Fe^{ik_{III} x} + Ge^{-ik_{III} x}$ ,  $k_{III} = k_I = \sqrt{\frac{2mE}{\hbar^2}}$

In this case  $D \neq 0$  since this function  $\Psi_{II}(x)$  only exists over the interval  $0 \leq x \leq a$ .

But  $G = 0$  since there are no reflected waves (waves traveling to the left) in Region III. A sketch of these wavefunctions follows

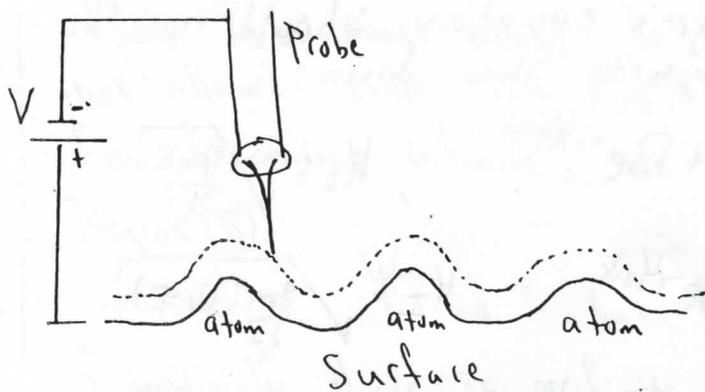


Classically, a particle can never appear in region  $x > a$  since it does not have enough energy to overcome the barrier. Quantum mechanically, the particle can penetrate the barrier — this is called the tunneling effect.

One can calculate the transmission probability (using the continuity conditions), and one would find that

$$T \sim e^{-2K_{II}a}, \quad K_{II} = \sqrt{\frac{2m}{\hbar^2}(U_0 - E)}$$

This is the basis for a Scanning Tunneling Microscope



The metal probe has a tip so fine that it is a single atom. Electrons need several electron volts to escape the surface, but when only  $V = 10\text{mV}$  is applied between probe and surface, electrons can tunnel through the ~~small~~ air gap if the spacing is small enough (a nanometer or two).

Keeping current ~~constant~~ (which is sensitive to gap distance) constant by mechanically moving the tip up and down gives a map of the surface. One can see detail down to  $0.1 \text{ \AA}$  or  $\frac{1}{10}$ th the diameter of an atom.