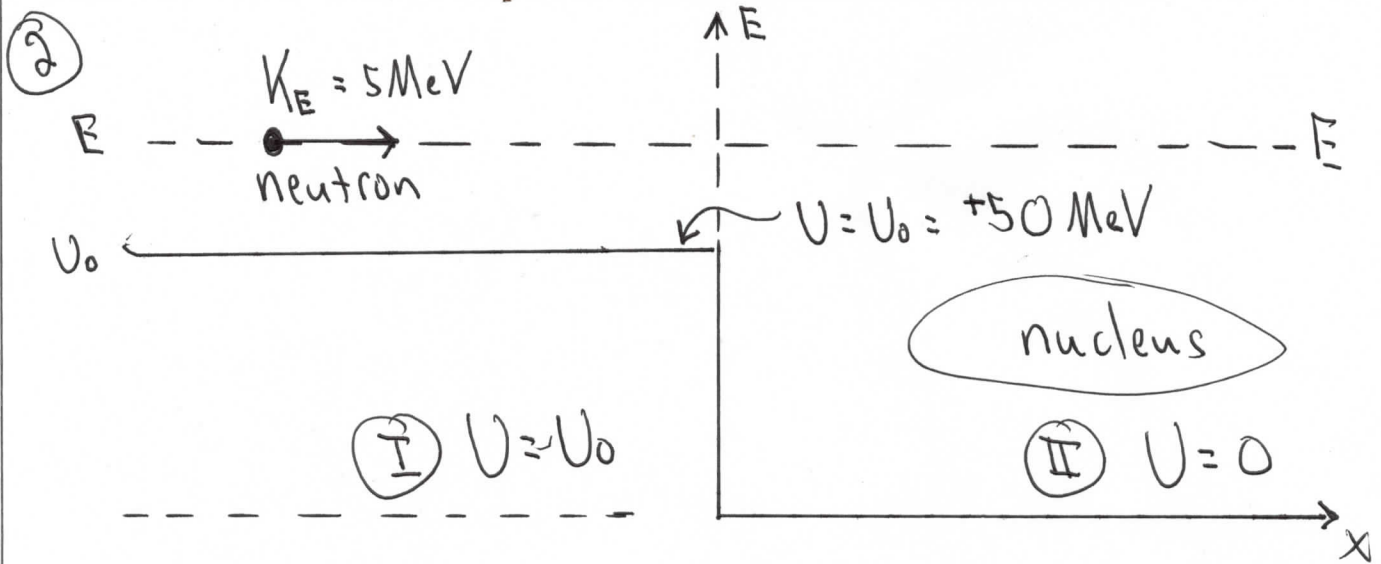


Physics 283

Quiz # 3 Solutions



(a) $\Psi_{\text{I}}(x) = \underbrace{A e^{i k_{\text{I}} x}}_{\substack{\uparrow \\ \text{wave travelling} \\ \text{to the right}}} + \underbrace{B e^{-i k_{\text{I}} x}}_{\substack{\uparrow \\ \text{reflected wave} \\ \text{traveling to the left}}}$

(b) $\Psi_{\text{II}}(x) = C e^{i k_{\text{II}} x} + D e^{-i k_{\text{II}} x}$

(c) For Region I: $-\frac{\hbar^2}{2m} \frac{d^2 \Psi_{\text{I}}}{dx^2} + U(x) \Psi_{\text{I}} = E \Psi_{\text{I}}$

$$\frac{d^2 \Psi_{\text{I}}}{dx^2} = -\frac{2m}{\hbar^2} (E - U_0) \Psi_{\text{I}}$$

$$\frac{d^2 \Psi_{\text{I}}}{dx^2} = -k_{\text{I}}^2 \Psi_{\text{I}}$$

$$k_{\text{I}} = \sqrt{\frac{2m}{\hbar^2} (E - U_0)}$$

For Region II: $-\frac{\hbar^2}{2m} \frac{d^2 \Psi_{\text{II}}}{dx^2} + \cancel{U(x)} \Psi_{\text{II}} = E \Psi_{\text{II}}$
 $U(x) = 0$

$$\frac{d^2 \psi_{II}}{dx^2} = -\frac{2mE}{\hbar^2} \psi_{II}$$

$$= -k_{II}^2 \psi_{II}$$

$$k_{II} = \sqrt{\frac{2mE}{\hbar^2}}$$

(d) Coefficient D in Region (II) is zero

$$D = 0$$

since there is no boundary to reflect from in region (II) to produce a reflected wave

(e) $\psi_I(x=0) = \psi_{II}(x=0)$

$$Ae^0 + Be^0 = Ce^0$$

\Rightarrow

$$A + B = C$$

$$\left. \frac{d\psi_I}{dx} \right|_{x=0} = \left. \frac{d\psi_{II}}{dx} \right|_{x=0}$$

$$ik_I A e^0 - ik_I B e^0 = ik_{II} C e^0 \Rightarrow k_I A - k_I B = k_{II} C$$

(f) Inserting $C = A + B$ into the 2nd relation gives

$$k_I A - k_I B = k_{II} (A + B)$$

$$(k_I - k_{II}) A = (k_{II} + k_I) B$$

$$\Rightarrow \frac{B}{A} = \frac{k_I - k_{II}}{k_I + k_{II}}$$

$$R = \frac{k_{II} |B|^2}{k_{II} |A|^2} = \left(\frac{k_{II} - k_{II}}{k_{II} + k_{II}} \right)^2$$

$$(g) E = K_E + U = 5 \text{ MeV} + 50 \text{ MeV} = 55 \text{ MeV}$$

$$R = \left(\frac{\sqrt{\frac{2m}{\hbar^2}(E-U_0)} - \sqrt{\frac{2mE}{\hbar^2}}}{\sqrt{\frac{2m}{\hbar^2}(E-U_0)} + \sqrt{\frac{2mE}{\hbar^2}}} \right)^2 = \left[\frac{\sqrt{\frac{2m}{\hbar^2}} \left(\frac{\sqrt{E-U_0} - \sqrt{E}}{\sqrt{E-U_0} + \sqrt{E}} \right) \right]^2$$

$$= \frac{\sqrt{E-U_0} - \sqrt{E}}{\sqrt{E-U_0} + \sqrt{E}} = \left(\frac{\sqrt{(55-50) \text{ MeV}} - \sqrt{55 \text{ MeV}}}{\sqrt{(55-50) \text{ MeV}} + \sqrt{55 \text{ MeV}}} \right)^2$$

$$= 0.288 \approx 29\%$$

Thus, according to this simple theory, a sizeable fraction of the neutrons are reflected.

⇒ 29% of incident neutrons do not contribute to nuclear fission

③ Carbon: $[\text{He}]2s^2 2p^2$

(a) For the two $2p$ electrons ($l=1$)

$$e^{-\#1} : l_1=1, s_1=\frac{1}{2}$$

$$e^{-\#2} : l_2=1, s_2=\frac{1}{2}$$

$$(b) \vec{L} = \vec{L}_1 + \vec{L}_2 \Rightarrow L = |l_1+l_2| \text{ to } |l_1-l_2| = |1+1| \text{ to } |1-1| = 2, 1, 0$$

$$\Rightarrow \boxed{L = 2, 1, 0}$$

$$\vec{S} = \vec{S}_1 + \vec{S}_2 \Rightarrow S = |s_1+s_2| \text{ to } |s_1-s_2| = |\frac{1}{2}+\frac{1}{2}| \text{ to } |\frac{1}{2}-\frac{1}{2}| = 1, 0$$

$$\Rightarrow \boxed{S = 1, 0}$$

$$\vec{J} = \vec{L} + \vec{S} \Rightarrow \left\{ \begin{array}{l} L=2 \left\{ \begin{array}{l} S=1 \Rightarrow |2+1| \text{ to } |2-1| = 3, 2, 1 \\ S=0 \Rightarrow |2+0| \text{ to } |2-0| = 2 \end{array} \right. \\ L=1 \left\{ \begin{array}{l} S=1 \Rightarrow |1+1| \text{ to } |1-1| = 2, 1, 0 \\ S=0 \Rightarrow |1+0| \text{ to } |1-0| = 1 \end{array} \right. \\ L=0 \left\{ \begin{array}{l} S=1 \Rightarrow |0+1| \text{ to } |0-1| = 1 \\ S=0 \Rightarrow |0+0| \text{ to } |0-0| = 0 \end{array} \right. \end{array} \right.$$

$$\boxed{J = |L+S| \text{ to } |L-S|}$$

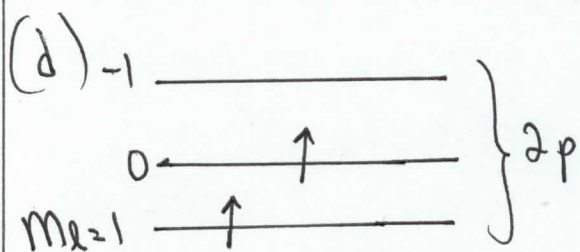
$$\Rightarrow \boxed{J = 3, 2, 1, 0}$$

(c) (S, L, J) states: $(1, 2, 3) (1, 2, 2) (1, 2, 1)$

$(0, 2, 2), (1, 1, 2), (1, 1, 1), (1, 1, 0), (0, 1, 1), (1, 0, 1)$

$\Rightarrow 10$ possible states

$(0, 0, 0)$

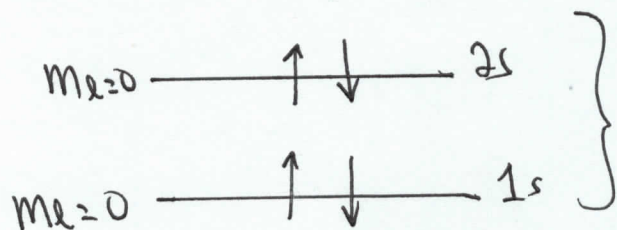


$$M_{L, \max} = m_{\ell 1} + m_{\ell 2} = 1 + 0 = 1$$

$$\Rightarrow L = 1$$

$$M_{S, \max} = m_{s1} + m_{s2} = \frac{1}{2} + \frac{1}{2} = 1$$

$$\Rightarrow S = 1$$



$L = 0, S = 0$ for filled subshells

Since the $2p$ subshell is less than half filled,

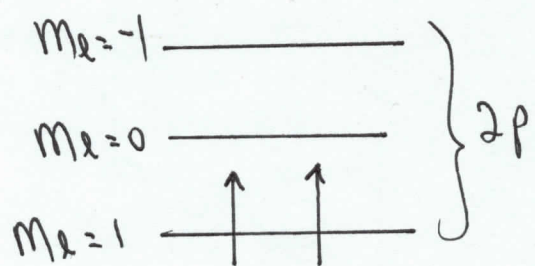
$$J = |L - S| = 1 - 1 = 0$$

The ground state (S, L, J) is $(1, 1, 0)$

(e) Spectroscopic notation: $L = 1 = P$ state
 multiplicity = $2S + 1 = 2 \cdot 1 + 1 = 3$
 $J = 0$

Spectroscopic state: 3P_0

(f) State with largest J : $(1, 2, 3) \rightarrow S = 1, L = 2$
 $J = 3$



to have $L = 2, S = 1$, we must have $M_L = m_{\ell 1} + m_{\ell 2} = 2, M_S = m_{s1} + m_{s2} = 1$
 The only way to have $M_S = 1, M_L = 2$

is to put 2 spin up electrons in the bottom level

\Rightarrow this violates the Pauli Exclusion principle

Thus, the $(1, 2, 3), (1, 2, 2), (1, 2, 1)$ states are not allowed