## CHEMISTRY 212 - LABORATORY SCHEDULE Fall 2016

**TEXT/MATERIALS:** Laboratory worksheets and POGIL handouts are available on the Blackboard website. Lab manual: *CHEM 212*, Fountainhead Press (2016); ISBN 978-1-68036-295-4.

**REQUIRED EYE PROTECTION:** Students must wear the approved goggles issued by the department *at all times in the laboratory—NO EXCEPTIONS*.

| Week of:                  | EXPERIMENT   |
|---------------------------|--|
| 1. Aug. 22 <sup>nd</sup>  | CHECK-IN / SAFETY ORIENTATION: Safety in the Laboratory; Glassware & Equipment/Data Representation and Recording |
| 2. Aug. 29 <sup>th</sup>  | Separation and Recovery of Components in a Ternary Mixture   |
| 3. Sept. 5 <sup>th</sup>  | LABOR DAY (Holiday – no lab meetings for the week.)  |
| 4. Sept. 12 <sup>th</sup> | The Bohr Model and Spectroscopy of the Hydrogen Atom   |
| 5. Sept. 19 <sup>th</sup> | Periodic Properties of Some Elements   |
| 6. Sept. 28 <sup>th</sup> | Lewis Structures and Molecular Shape   |
| 7. Oct. 3 <sup>rd</sup>   | Determining the Empirical Formula of Magnesium Oxide   |
| 8. Oct. 10 <sup>th</sup>  | LAB MIDTERM EXAM   |
| 9. Oct. 17 <sup>th</sup>  | Stoichiometry - A Mole Ratio Study   |
| 10. Oct. 24 <sup>th</sup> | Classifying Chemical Reactions   |
| 11. Oct. 31st             | Redox Chemistry - Activity of Metals   |
| 12. Nov. 7 <sup>th</sup>  | Calorimetry and Hess's Law   |
| 13. Nov. 14 <sup>th</sup> | Determination of the Universal Gas Constant, R / CHECK OUT   |
| 14. Nov. 21st             | THANKSGIVING BREAK   |
| 15. Nov. 28 <sup>th</sup> | LAB FINAL EXAM   |

## \*FAILURE TO CHECK OUT MAY RESULT IN A FAILING GRADE FOR THE ENTIRE SEMESTER.

Grading: The overall lab grade is a weighted average, and is calculated using the formula below:

(Lab average x 0.70) + (Midterm Exam x 0.15) + (Final Exam x 0.15) = lab grade

Letter grades are assigned based on the overall lab grade compared to the following cutoffs:

90% = A; 80% = B; 70% = C; 60% = D; <60% = F