FALL 2015 – GENERAL CHEMISTRY 210-0002 (6136)

Co-requisite: CHEM 212 - General Chemistry Laboratory I

Principal Instructor – Prof. Narayan S. Hosmane, FR 305, 753-3556 hosmane@niu.edu (short e-mail questions)
Co-Instructors: Prof. Chong Zheng, LAT 326, 753-6871 czheng@niu.edu; Prof. Tao Xu, LAT 412, 753-6357 txu@niu.edu, and Mr. Michael Vega, LAT 313, mvega5@niu.edu.

Lectures: MWF, 11:00 –11:50 AM in La Tourette Hall 200

Office Hours – Tu and Th, 1:00 - 2:30 PM, or by appointment


Exams, Assignments, Quizzes and Grading

Exams - Tentative dates for 100-points hourly exams are indicated below in the lecture schedule. Make-up exams will NOT be given under any circumstances. Missed Exam will be considered as one of the lowest exam grades. However, one of the lowest exam grades will be replaced by ALEKS grade that is mandatory!

Surprise Quizzes (Pop Quizzes) – Surprise quizzes (believe me you will be surprised!) for a TOTAL 100 points (worth a grade of ONE exam) will be given during the lecture and/or the recitation hours. However, only BEST TEN quiz grades will be counted toward your FINAL GRADE. There will be ABSOLUTELY no make-up quizzes. Exam and Quiz Grades will not be curved!

Online Homework and Self-Assessment through ALEKS is mandatory (100 Points)

Total points = 600 points (best two hourly exams = 200; surprise quizzes = 100; ALEKS = 100; final exam = 200)

Grading scale: A > 90% (540 pts.), B > 80% (480 pts.), C > 70% (420 pts.), D > 60% (360 pts.), F < 60%

TENTATIVE LECTURE SCHEDULE

<table>
<thead>
<tr>
<th>WEEK</th>
<th>CHAPTER/TOPIC</th>
<th>Exams Dates</th>
</tr>
</thead>
<tbody>
<tr>
<td>01. August 24</td>
<td>Introduction/Chapter1: Language of Chemistry: Matter and Measurements</td>
<td></td>
</tr>
<tr>
<td>02. August 31</td>
<td>Ch. 2: Dalton’s Atomic Theory of Matter: Origins of the Universe</td>
<td></td>
</tr>
<tr>
<td>03. September 07</td>
<td>Monday: Labor Day</td>
<td>No Class</td>
</tr>
<tr>
<td>04. September 08</td>
<td>Ch. 2: Continued; Ch. 3: Chemical Reactions: Origin of Life</td>
<td></td>
</tr>
<tr>
<td>05. September 14</td>
<td>Ch. 3: Continued; Ch. 4: Stoichiometry: Quantitative Relationships</td>
<td></td>
</tr>
<tr>
<td>06. September 21</td>
<td>Ch. 4: Continued; Ch. 5: Thermochemistry: Energy for Life</td>
<td></td>
</tr>
<tr>
<td>07. September 28</td>
<td>Ch. 5: Continued; Ch. 6: Gases: Their Properties and Behavior</td>
<td></td>
</tr>
<tr>
<td>08. October 05</td>
<td>Exam 1 is on Monday, October 5, 2015; Ch. 6: Continued.</td>
<td>Exam 1 (10/05)</td>
</tr>
<tr>
<td>09. October 12</td>
<td>Ch. 7: Quantum Theory and Atomic Structure</td>
<td></td>
</tr>
<tr>
<td>10. October 19</td>
<td>Ch. 7 Continued; Ch. 8: Electron Configuration and Valence Electrons</td>
<td></td>
</tr>
<tr>
<td>11. October 26</td>
<td>Ch. 8: Continued.</td>
<td></td>
</tr>
<tr>
<td>12. November 02</td>
<td>Exam 2 is on Monday, Nov. 2, 2015; Ch. 9: Chemical Bonding: Basic Concepts</td>
<td></td>
</tr>
<tr>
<td>13. November 09</td>
<td>Ch. 9: Continued.</td>
<td></td>
</tr>
<tr>
<td>14. November 16</td>
<td>Ch. 10: Molecular Geometry: Bonding Concepts</td>
<td></td>
</tr>
<tr>
<td>15. November 23</td>
<td>Exam 3 is on Monday, November 23, 2015</td>
<td>Exam 3 (11/23)</td>
</tr>
<tr>
<td>16. November 25-29</td>
<td>Thanksgiving Holidays</td>
<td>No Classes</td>
</tr>
<tr>
<td>17. November 30</td>
<td>Ch.10: Continued.</td>
<td></td>
</tr>
<tr>
<td>18. December 09</td>
<td>FINAL EXAM IS ON WEDNESDAY FROM 11:00 AM – 12:50 PM (LAT 200)</td>
<td>Final Exam (12/9)</td>
</tr>
</tbody>
</table>

Tutors and Lab TA Office Hours: The Department of Chemistry and Biochemistry maintains a free Tutor Room for General Chemistry students. The Tutor Room is in Faraday 247 and is staffed Monday through Thursday from 8:30 AM to 3:30 PM with a lunch break. On Fridays, the Tutor Room closes at 2:30 PM. General Chemistry laboratory TA office hours are held in Faraday 412 (Penthouse). The laboratory TA office hour schedule is posted outside Faraday 412, and at the departmental stockroom window. Students are also encouraged to ask laboratory TAs for assistance in understanding the lecture material.

Paid Tutors - Names of tutors for hire are available: see Linda Davis in FR 319 (departmental office).
Getting Started with ALEKS

You must register and complete your Initial Assessment by 11:00 pm on the first day of Classes (but it’s great to start now to get ahead!)

**ALEKS IS VERY UNFRIENDLY TO PROCRASTINATORS.** ALEKS was designed by psychologists who specialized in learning, who know that procrastinating until the night before an assignment is due and then cramming until the wee hours is not a good way to learn. You can’t do ALEKS that way. You must put in some time every day, or else you will fall so far behind on basic topics that you won’t have time to complete the Objectives. It is impossible to earn a good score in ALEKS by cramming all night right before the due date!

To register as an ALEKS user:

2. Click on the link marked "SIGN UP NOW" (upper left corner of the screen).
3. On the next screen you will be asked to provide the following course code: **GGUVE-LJXPT**
4. The next screen will ask for your full name, your email address, and your NIU student Z-ID number. Please provide all the information requested, even the information that is listed as optional. Remember, your NIU student Z-ID number is NOT your social security number. **You must enter your NIU Z-ID number to receive credit for your work.**
5. On the next screen you will receive your ALEKS login name and a temporary password, and you will have the chance to change your temporary password. **We recommend that you change your temporary password.**
6. ALEKS will now walk you through a tutorial of how to input answers.
7. Once you have completed the input tutorial, ALEKS will prompt you to complete the Initial Assessment. **This initial assessment is VERY important.** It will identify any pre-requisite topics you need to learn before you can be successful in this course, it can also identify what you already know, so you don’t waste time on topics you already know. This is not a test for points – it is ALEKS trying to build an efficient and customized learning path for you to achieve success. Don’t use any outside resources during this assessment – if ALEKS thinks you know more than you do – your homework will be really hard and take you much longer to do! Also, when you know how to do a problem – do it! If you prove mastery of that topic, you’ll have less to do in learning mode.

**NOTE:** ALEKS has two modes, assessment mode and learning mode. Assessment mode measures what you know today (and will figure out if you have forgotten any previously learned topics). Learning mode is where you learn, practice and master the topics – and grow your ALEKS pie (your knowledge). You will frequently be assessed throughout the ALEKS course to ensure ALEKS is providing you the best instruction possible.

**ALEKS Technical Support**

**No one in the department of chemistry at NIU can provide you with competent technical support for ALEKS.** Do not write to your teacher with operational questions about ALEKS…he will not be able to help you. Instead, try the following resources on the ALEKS website:

1. **Frequently Asked Questions:** [http://www.aleks.com/faqs](http://www.aleks.com/faqs)
3. **Troubleshooting:** [http://www.aleks.com/support/troubleshooting](http://www.aleks.com/support/troubleshooting)
4. **Email the Support Team:** [http://support.aleks.com](http://support.aleks.com)

**SCHEDULE FOR MANDATORY RECITATION**

**Recitation Teaching Assistants** – Sydney Watkins - swatkins@niu.edu (Sections R005, R007, R008 and R012)
Ashley De Lio - adelio1@niu.edu (Sections R006, R009, R011 and R013)

**TA Office Hours** – To be announced or by appointment, Faraday 351 (Watkins) and De Lio (LaT 310)

<table>
<thead>
<tr>
<th>Section</th>
<th>Lecture Time</th>
<th>Location</th>
<th>Recitation Time</th>
<th>FR</th>
</tr>
</thead>
<tbody>
<tr>
<td>R005</td>
<td>MWF, 11:00 AM</td>
<td>La Tourette 200</td>
<td>Recitation Wednesday, 08:00 AM</td>
<td>FR 205</td>
</tr>
<tr>
<td>R007</td>
<td>MWF, 11:00 AM</td>
<td>La Tourette 200</td>
<td>Recitation Wednesday, 10:00 AM</td>
<td>FR 205</td>
</tr>
<tr>
<td>R008</td>
<td>MWF, 11:00 AM</td>
<td>La Tourette 200</td>
<td>Recitation Wednesday, 12:00 PM</td>
<td>FR 205</td>
</tr>
<tr>
<td>R012</td>
<td>MWF, 02:00 PM</td>
<td>Faraday 143</td>
<td>Recitation Wednesday, 05:00 PM</td>
<td>FR 205</td>
</tr>
</tbody>
</table>

**Recitation** - Surprise quizzes for a TOTAL of 100 points (worth a grade of ONE exam) will be given during the lecture and/or the recitation hours. However, only **BEST TEN** quiz grades will be counted toward your FINAL GRADE. There will be **ABSOLUTELY no** make-up quizzes. Exam and Quiz Grades will not be curved! **Your attendance at each assigned recitation session is mandatory**! However, help is given for the online assignments, if requested.

2 EXAM-I: October 5; EXAM-II: November 2; EXAM-III: November 23; FINAL EXAM: December 09
THE DUE DATES FOR ONLINE (ALEKS) ASSIGNMENTS ARE ALL FIXED AND REQUESTS FOR EXTENSION OF DUE DATES WILL NOT BE HONORED UNDER ANY CIRCUMSTANCES!

CHEMISTRY 210 - GENERAL EDUCATION AND COURSE CONTENT OBJECTIVES

General Education Course Objectives

- Improve ability to think critically and logically
- Improve ability to reason quantitatively and to perform basic chemical computations
- Improve ability to interpret mathematical models
- Learn how to use the scientific method and theories to understand chemical phenomena
- Develop an appreciation for the importance of the role of chemistry in everyday life
- Develop an understanding of the historical development of the field of chemistry

Content Objectives of this Course

- Understand the components of atoms and ions
- Learn how to write chemical formulas, and how to name compounds
- Learn how to balance chemical equations and how to perform simple stoichiometry calculations
- Understand the behavior of gases, liquids, and solids
- Become familiar with the electronic structure of atoms and understand how chemical reactivity depends on electronic structure
- Correctly predict the shapes of complex molecules and ions, and become familiar with the theories of chemical bonding.

RESTRICTIONS ON USE OF CELL (MOBILE) PHONES, WiFi, INTERNET, LAPTOPS AND iPADS IN CLASSROOMS:

- Use of Cell (Mobile) Phone during the class hours is strictly prohibited. If you are caught of using cell phone or receiving any live text and/or e-mail messages, your cell phone or laptop and/or iPad will be confiscated. Therefore, your cell phone or laptop/iPad must be either switched off or on AIRPLANE (SILENT) MODE. Therefore, you must carry a separate ordinary calculator that MUST NOT be connected to Internet through WiFi or any telephone network! Your Internet entertainment will be a distraction for others!

HOMEWORK ASSIGNMENTS:

- There is no specific homework assigned for this class. However, it is your responsibility to do as much end-of-chapter problems as possible in order to perform well in the pop quizzes, exams and ALEKS. You can check your answers in the solution manual, but the professor of this class will not evaluate and grade your homework problems.

SPECIAL ACCOMMODATION:

Any student who may need an accommodation due to a disability, please make an appointment to see me during my office hours, or when convenient. A letter from Disability Support Services authorizing your accommodations is usually needed before accommodations can be granted.